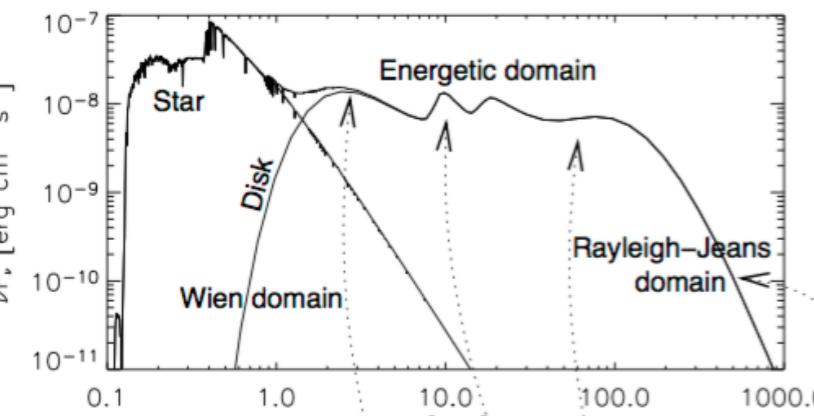


# Modelling of protoplanetary disks

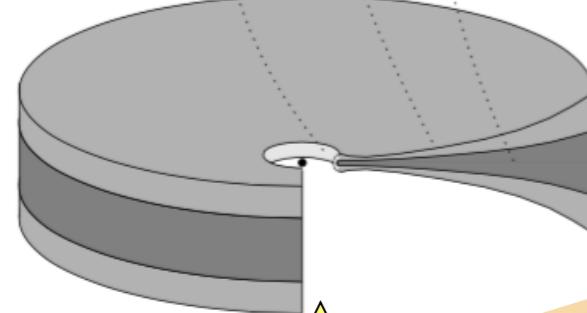
A “cookbook” for observationally-motivated disk physico-chemical models

Catherine Walsh  
NWO Veni Fellow  
Leiden Observatory

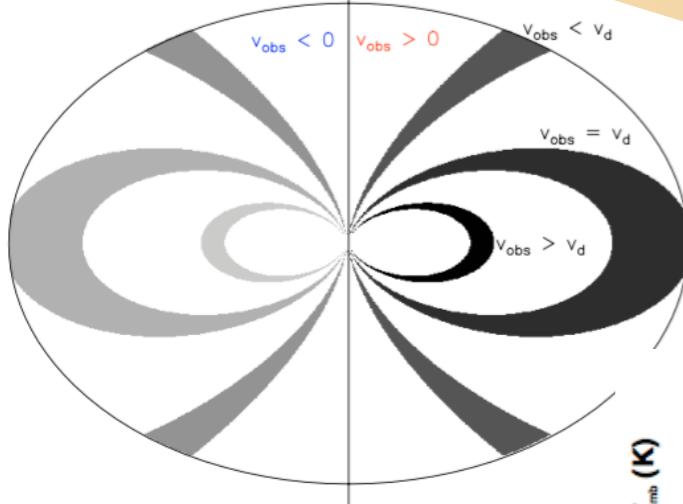
# Observations of protoplanetary disks



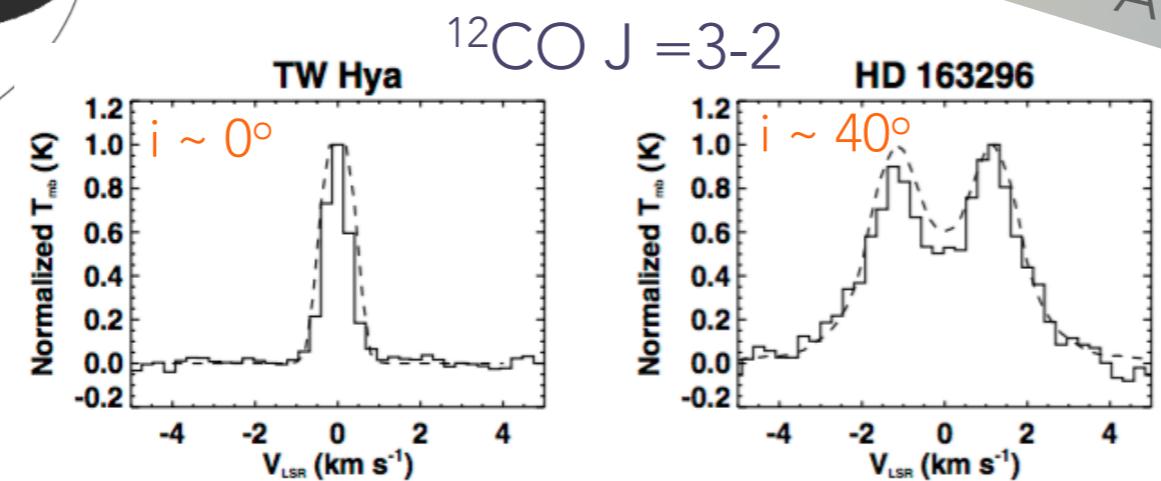
Dust



Gas

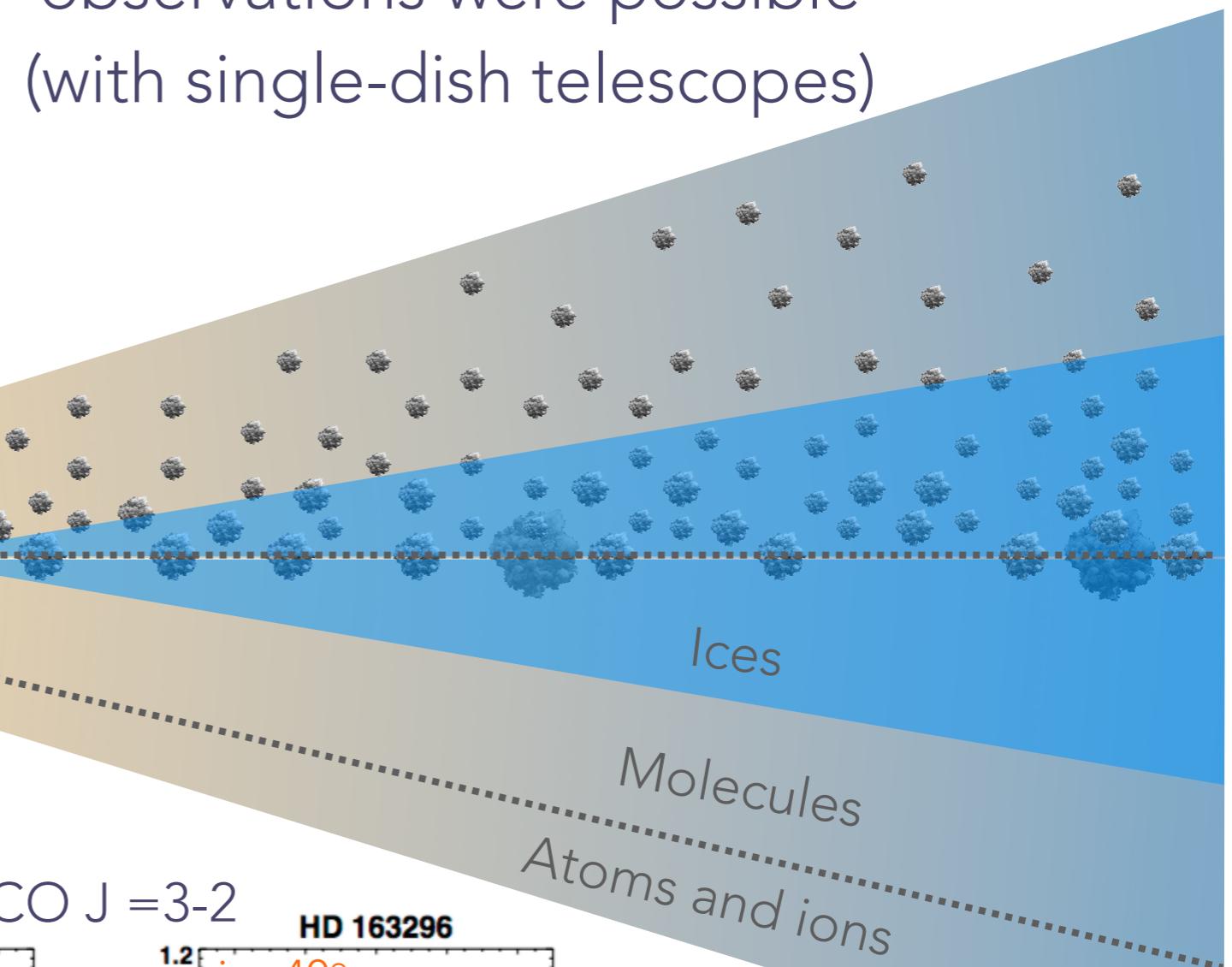


Dust  
spectral  
energy  
distribution



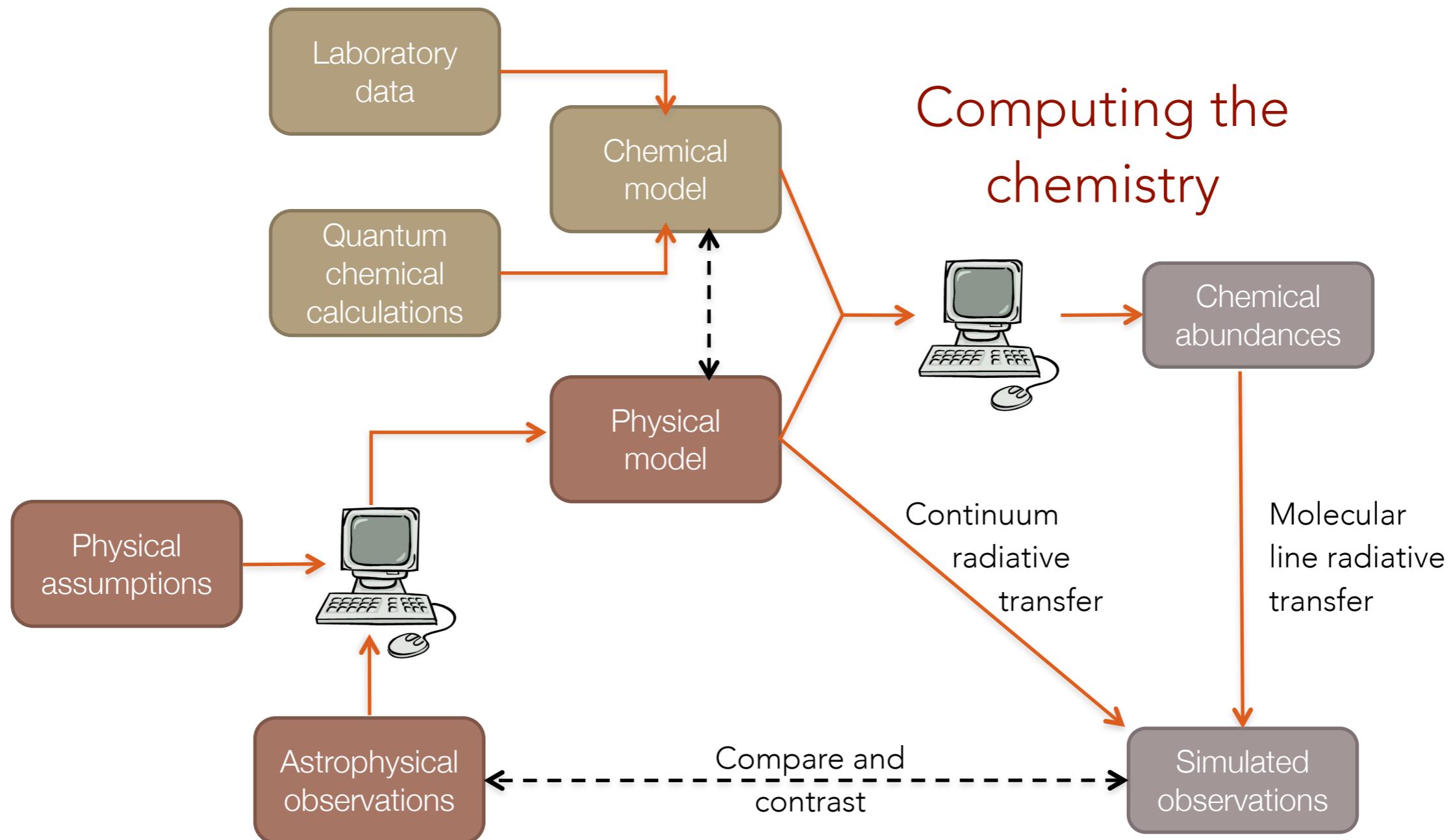
Molecular/atomic  
emission lines

Until recently: only spatially-unresolved  
observations were possible  
(with single-dish telescopes)



Ices  
Molecules  
Atoms and ions

# General outline of a physico-chemical model



# Outline

Disk physical structure:

- \* dust
- \* gas

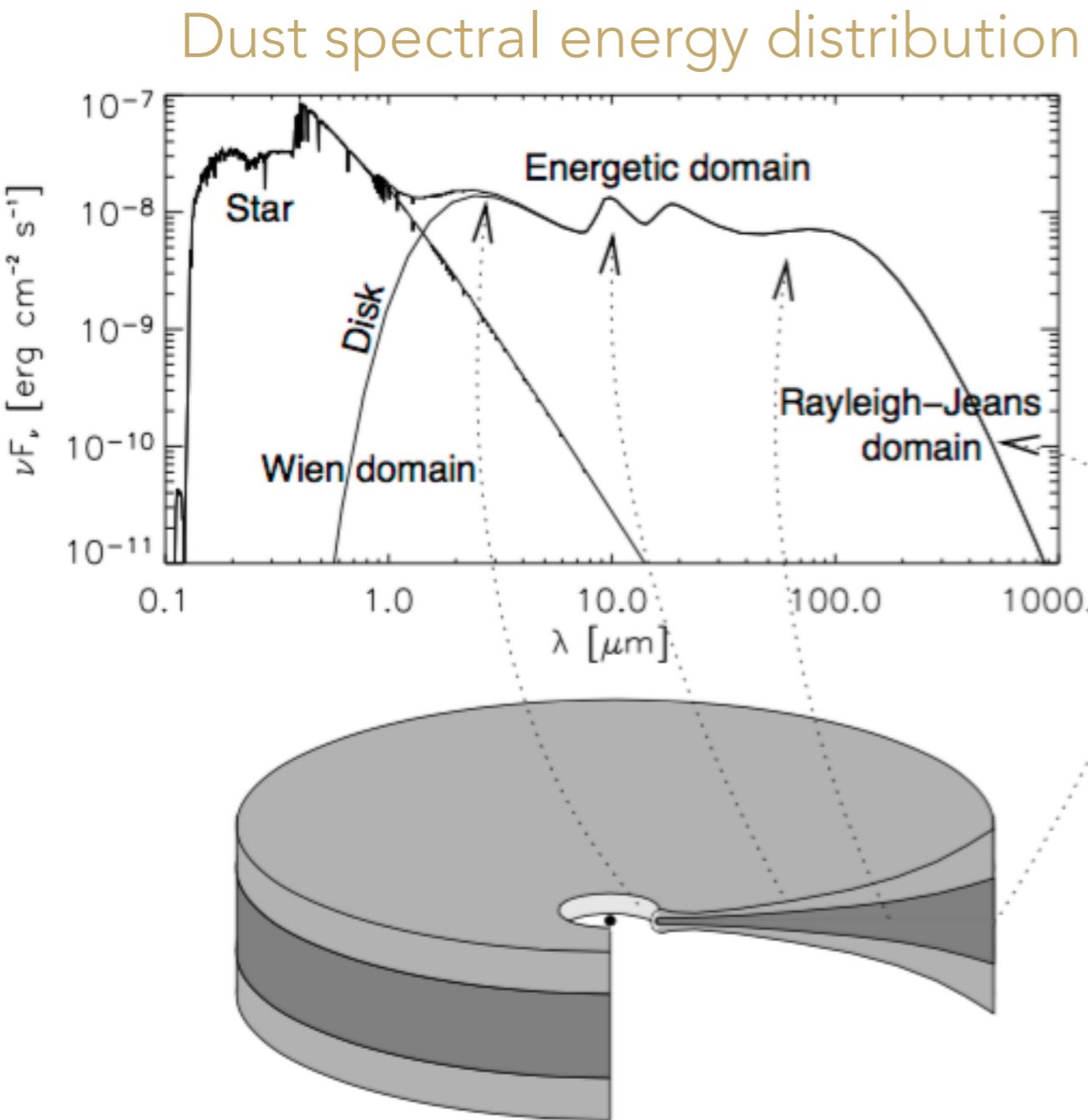
Disk chemical structure:

- \* gas-phase (vapour)
- \* solid-phase (ice)

Building physico-chemical models

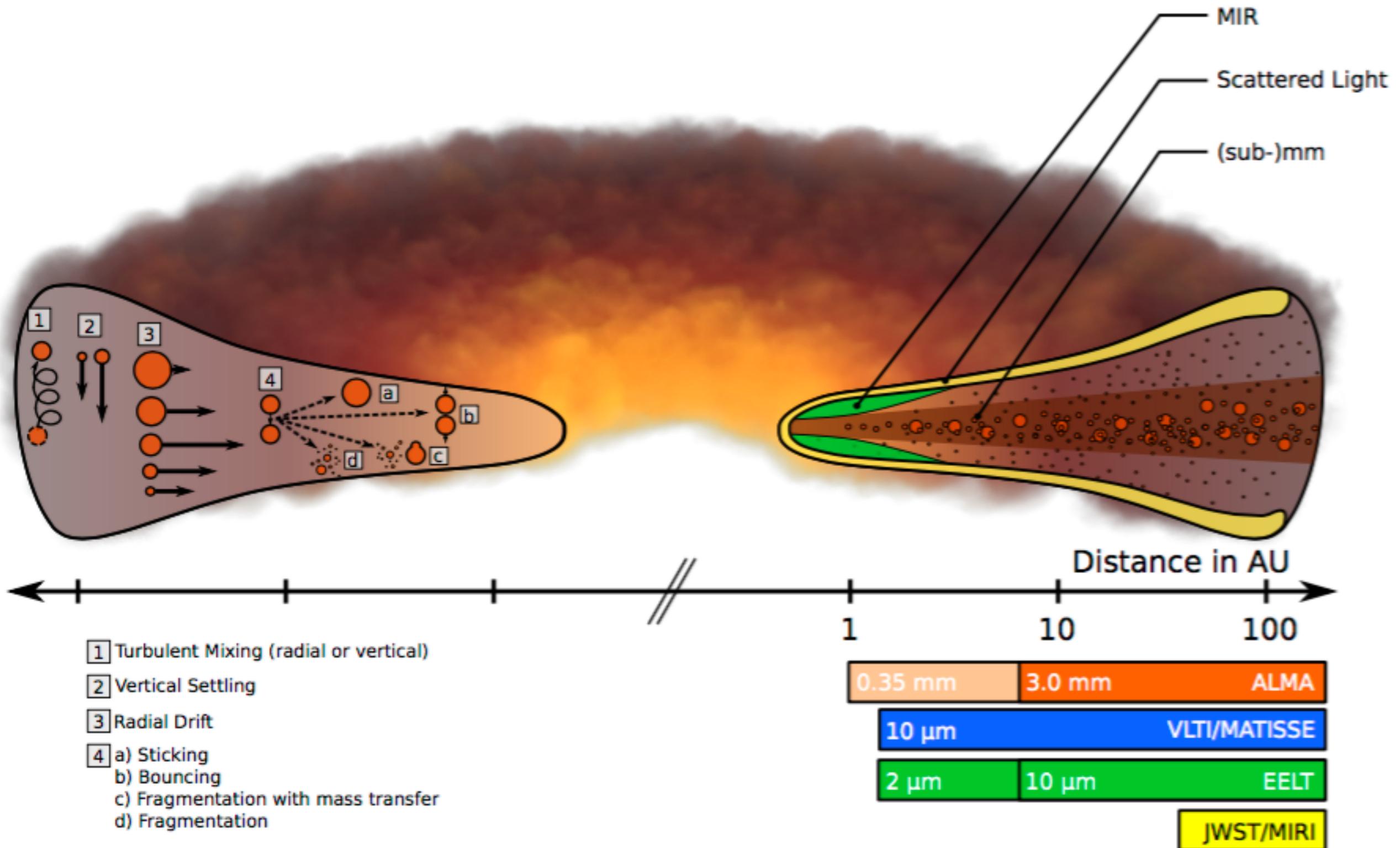
# Observations of dust in protoplanetary disks

Early modelling efforts concerning the dust focussed primarily on reproducing the spectral energy distribution (SED)



- ▶ Optical wavelengths: scattered light; hence no temperature/density information
- ▶ Near-infrared wavelengths: originates mainly from “hot” inner rim
- ▶ Mid-infrared wavelengths: originates from “warm” dust close to the star (< 10 AU) and is typically optically thick; hence, no density information but probes temperature of dust photosphere
- ▶ Sub-mm/mm wavelengths: originates from “cold” dust in outer disk (> 10 AU) and is typically optically thin; hence, has both temperature and density information

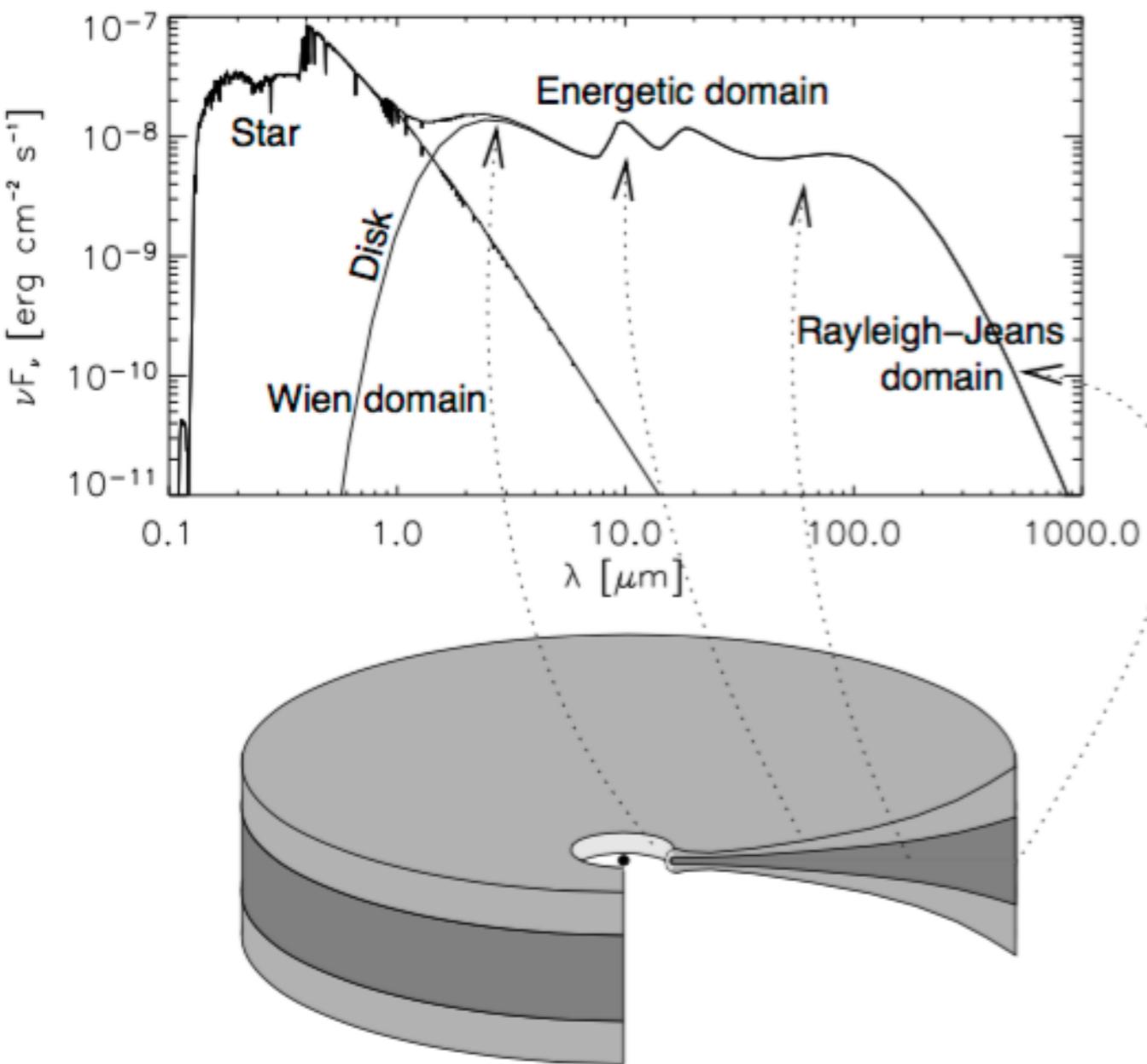
# Observations of dust in protoplanetary disks



# Observations of dust in protoplanetary disks

Some simple assumptions about the (sub)mm opacity and dust temperature can yield an estimation of the disk mass

Dust spectral energy distribution



Optical depth

$$\tau_\nu = \int \rho \kappa_\nu ds = \kappa_\nu \Sigma,$$

Dust mass opacity

$$\kappa_\nu = 0.1 \left( \frac{\nu}{10^{12} \text{ Hz}} \right)^\beta \text{ cm}^2 \text{ g}^{-1}.$$

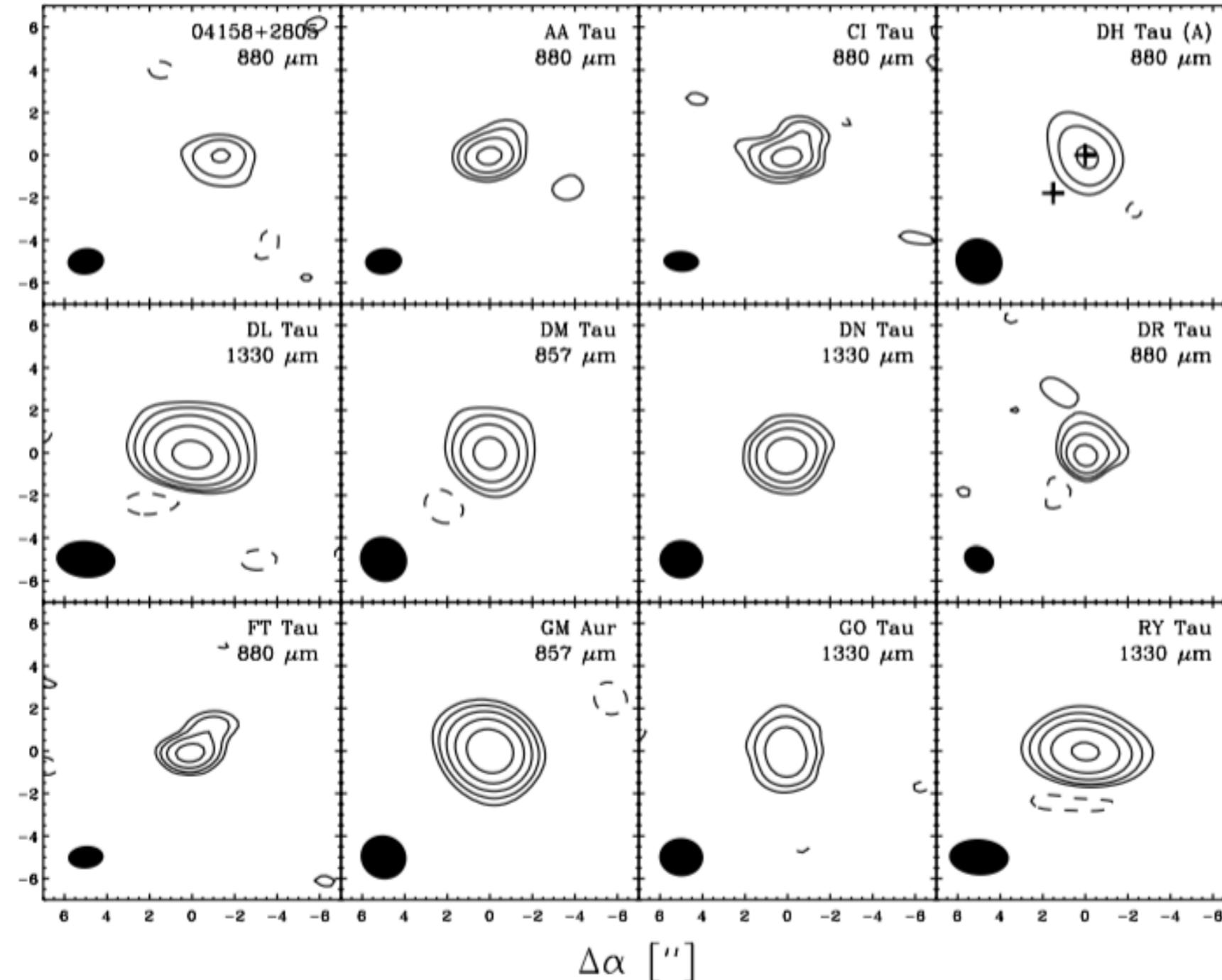
Disk mass

$$M(\text{gas + dust}) = \frac{F_\nu d^2}{\kappa_\nu B_\nu(T)},$$

Gas-to-dust mass ratio  $\sim 100$

# Observations of dust in protoplanetary disks

The advent of (sub)mm interferometry required more sophisticated models to describe the radial disk structure



Disk surface density, temperature, and dust mass opacity were “well-fit” using power laws

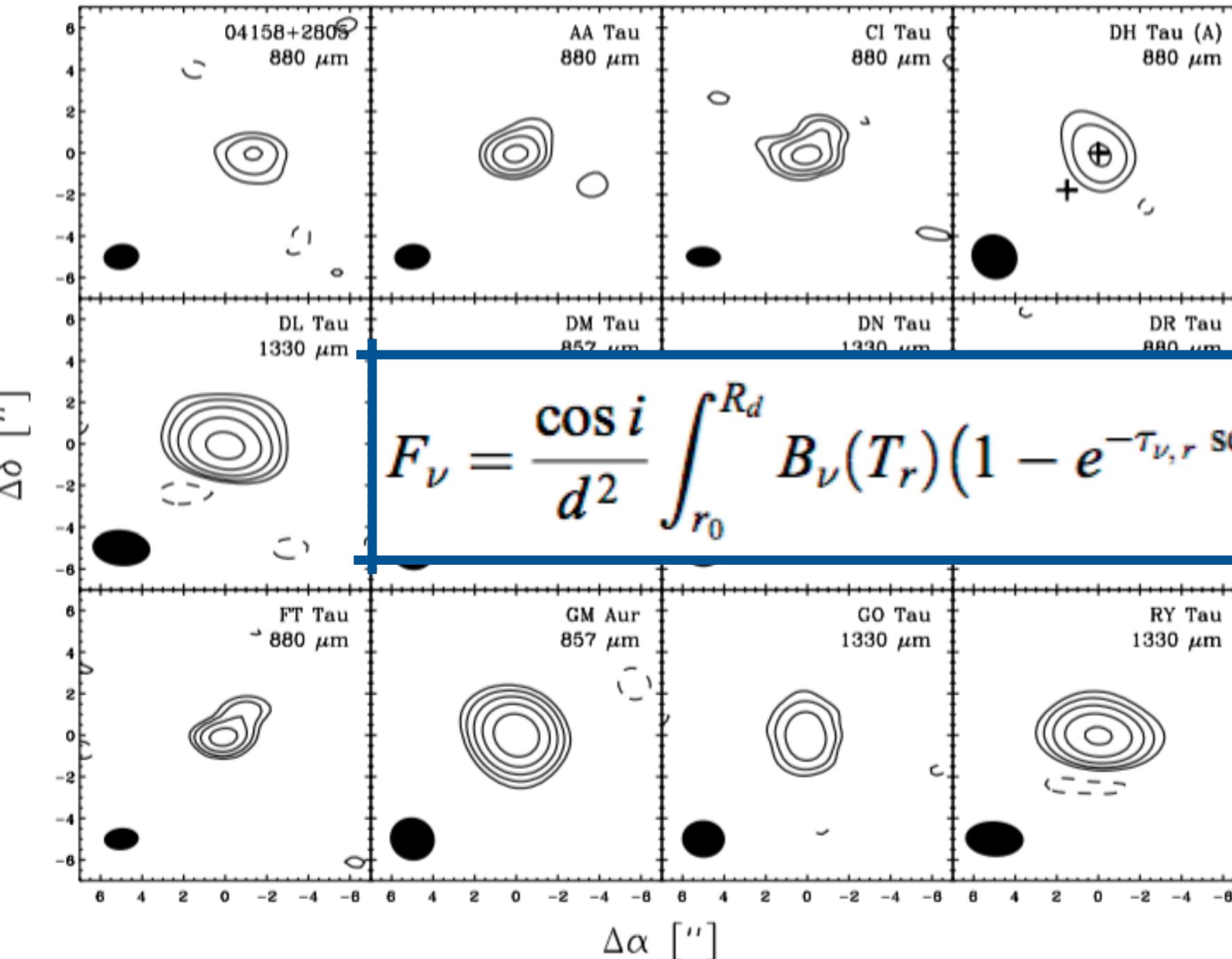
$$T_r = T_1 \left( \frac{r}{1 \text{ AU}} \right)^{-q},$$

$$\Sigma_r = \Sigma_5 \left( \frac{r}{5 \text{ AU}} \right)^{-p},$$

$$\kappa_\nu = \kappa_0 \left( \frac{\nu}{\nu_0} \right)^\beta,$$

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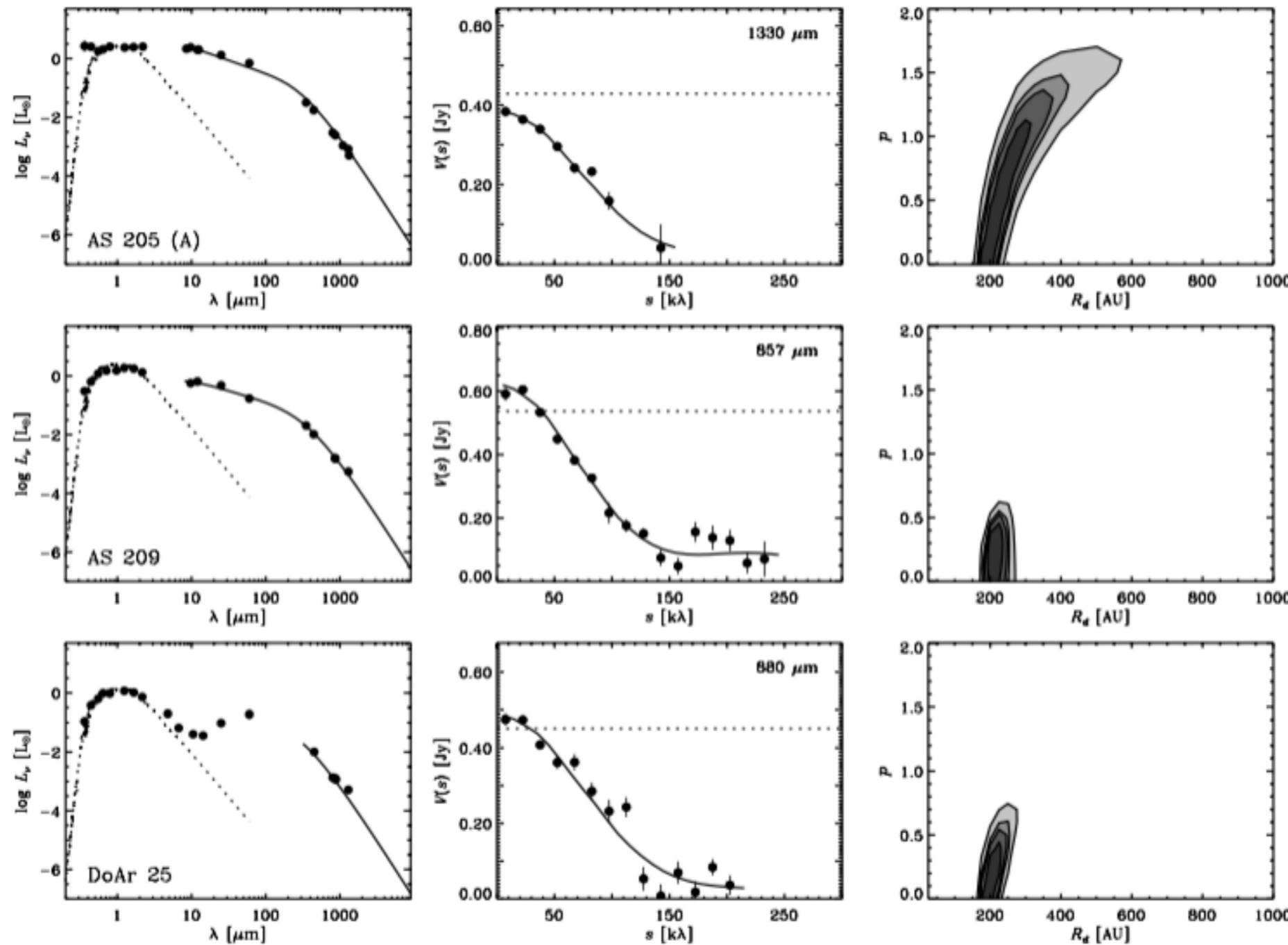
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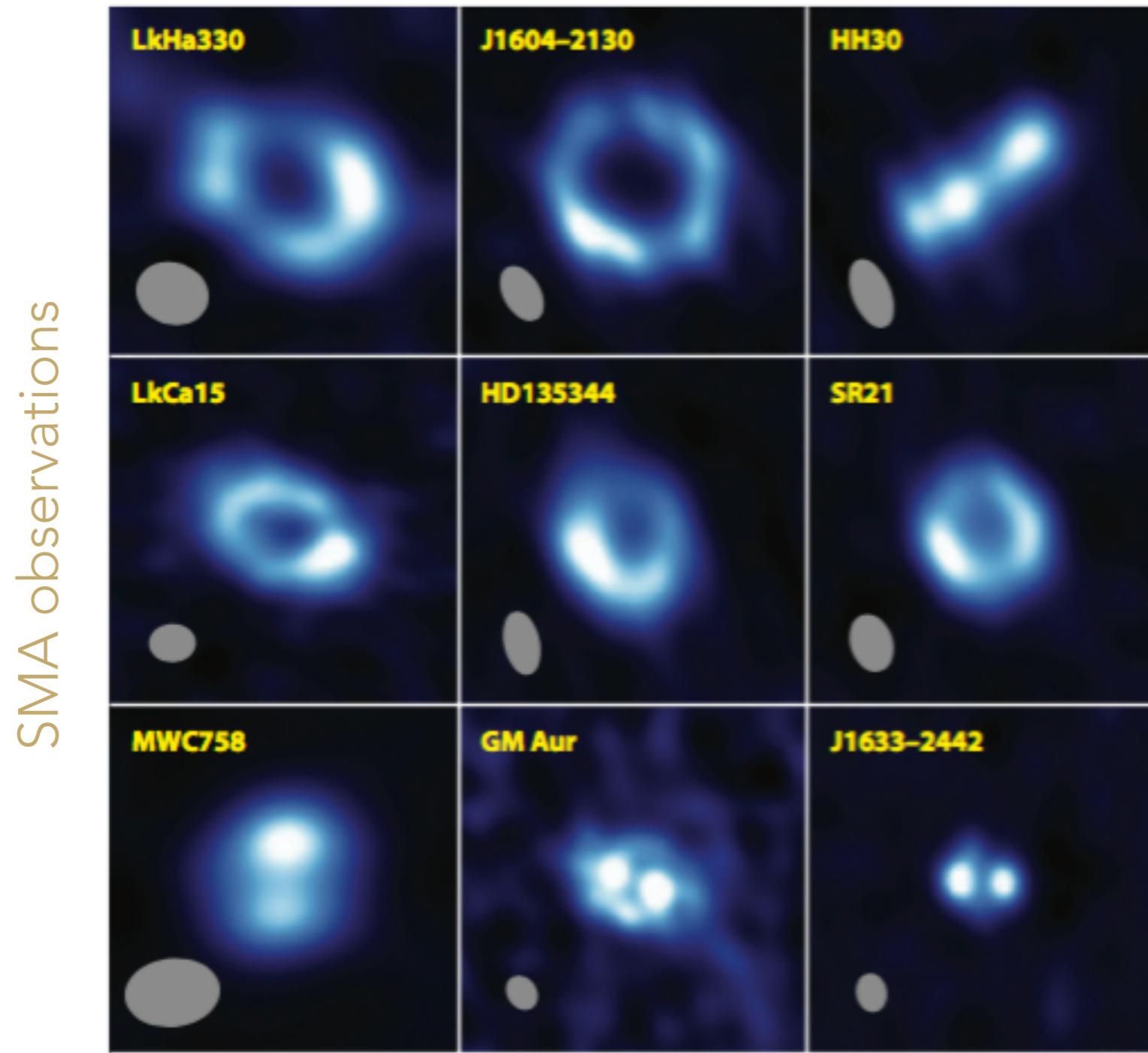
# Observations of dust in protoplanetary disks

Fitting is done directly to the interferometric data (so-called visibilities) which are the Fourier transform of the intensity distribution



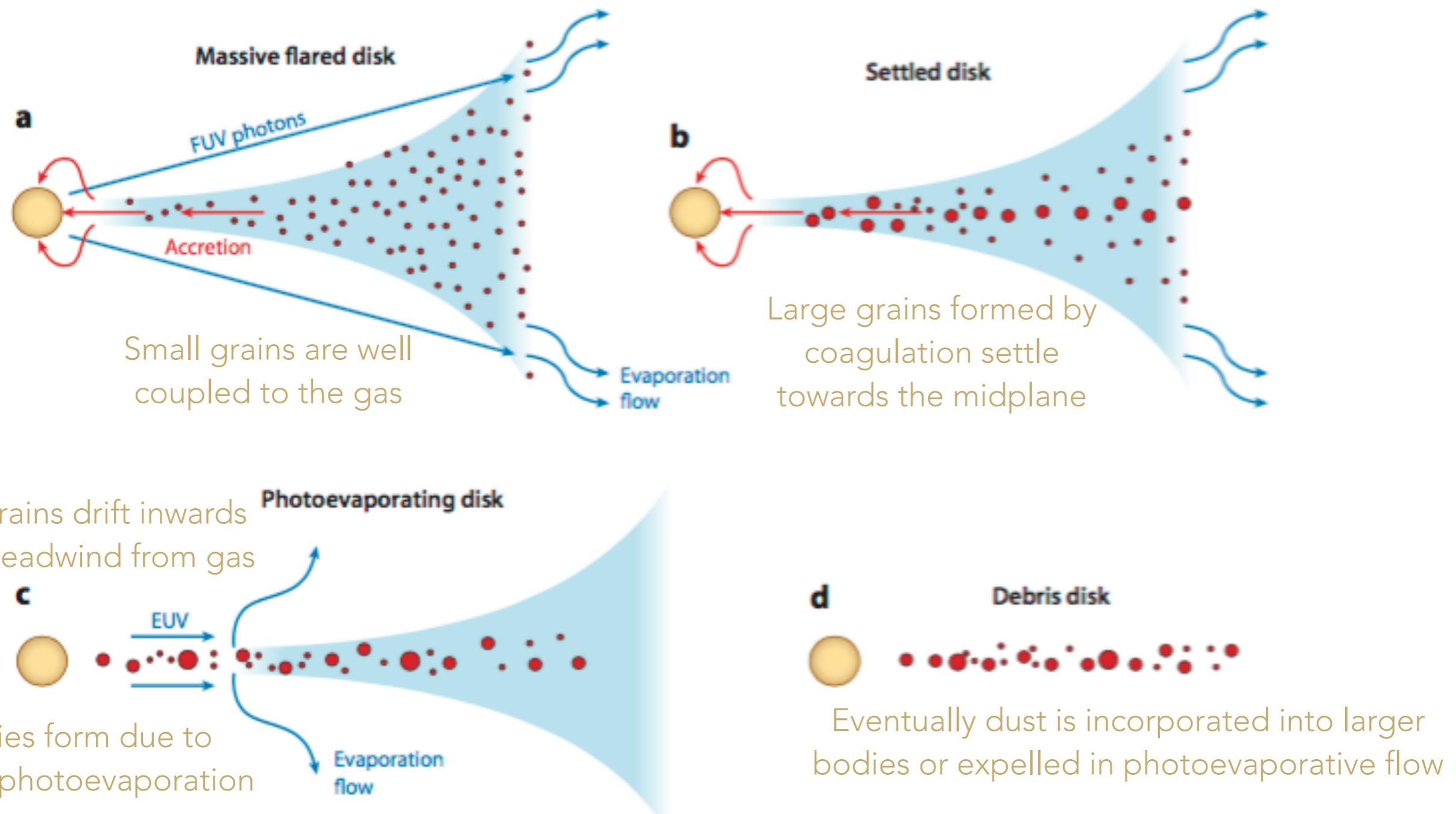
# Observations of dust in protoplanetary disks

Higher-resolution (sub-arcsecond) interferometric data required additional considerations for models of the dust emission: cavities and rings



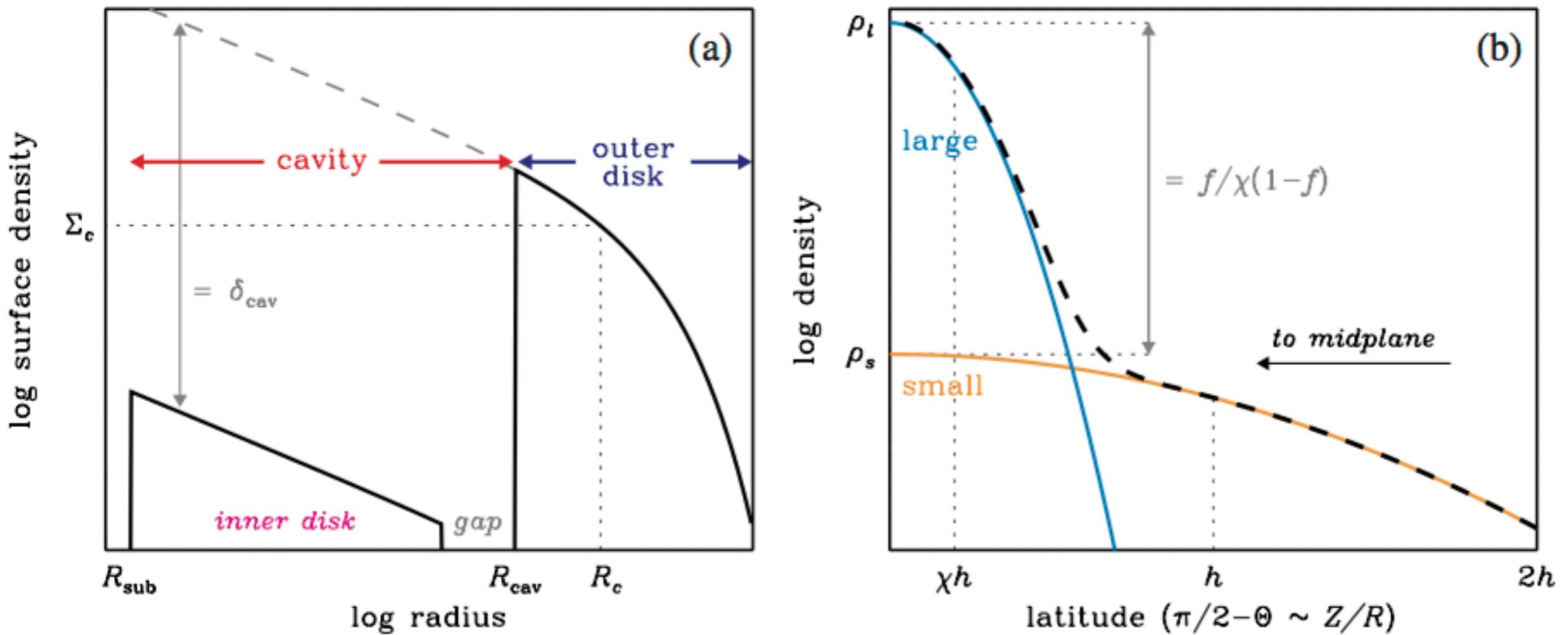
# Evolution of dust in protoplanetary disks

## Generalised picture of dust evolution in protoplanetary disks



# Modelling of dust in protoplanetary disks

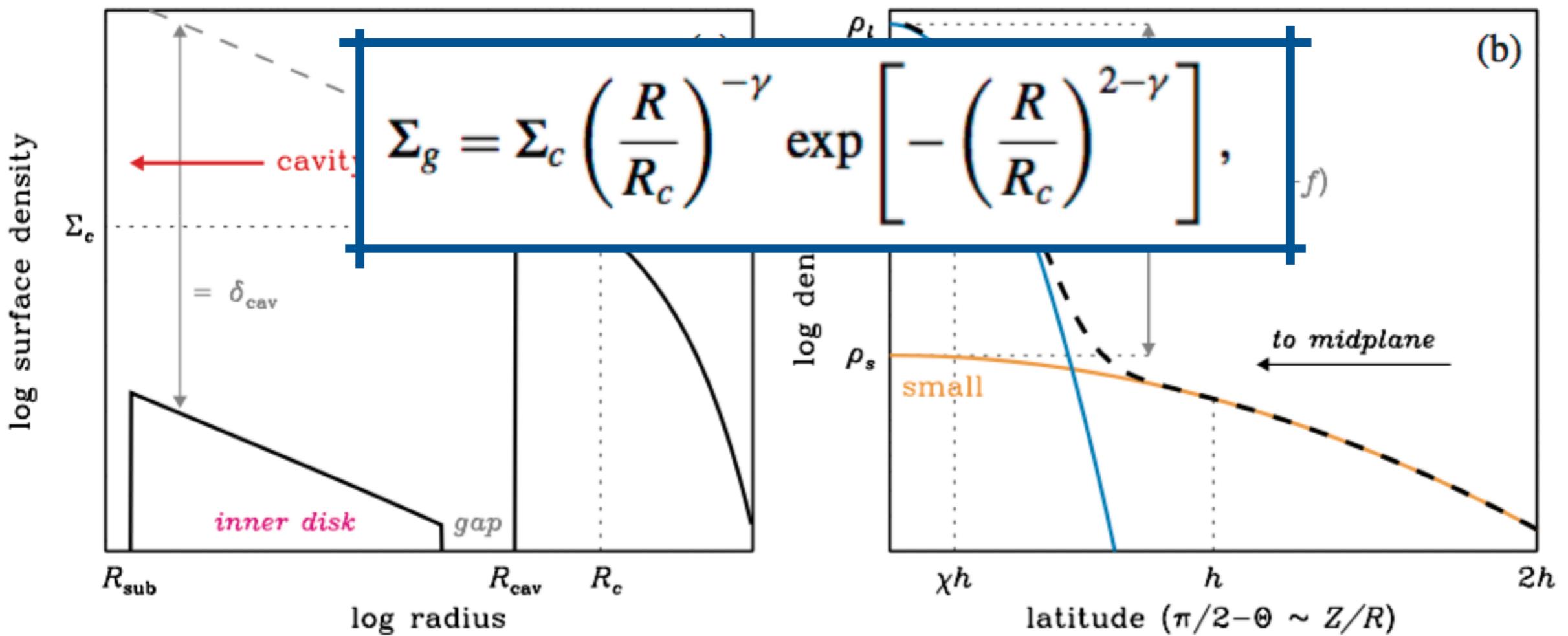
Simple power-law models are still used, but with gaps and cavities, and “small” and “large” grains are decoupled to simulate settling



Despite more complex models being used to model more complex data, significant degeneracies still remain in the models

# Modelling of dust in protoplanetary disks

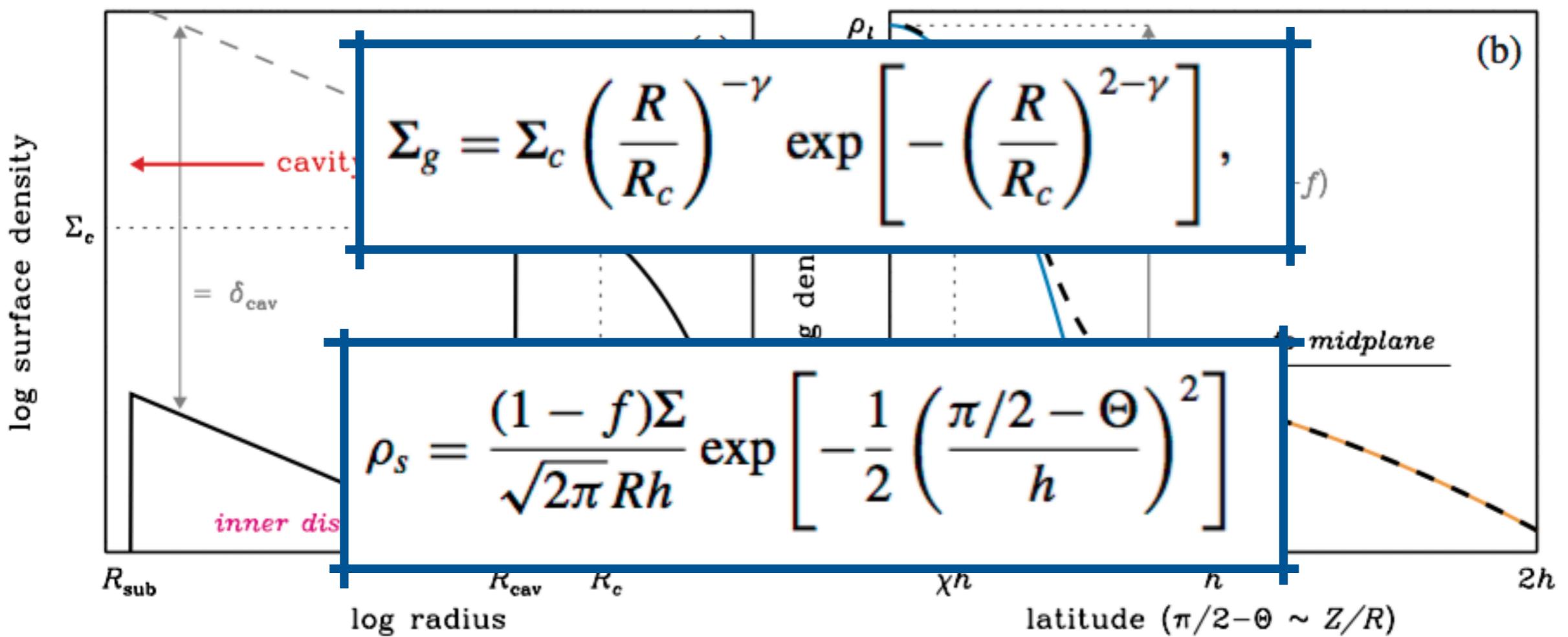
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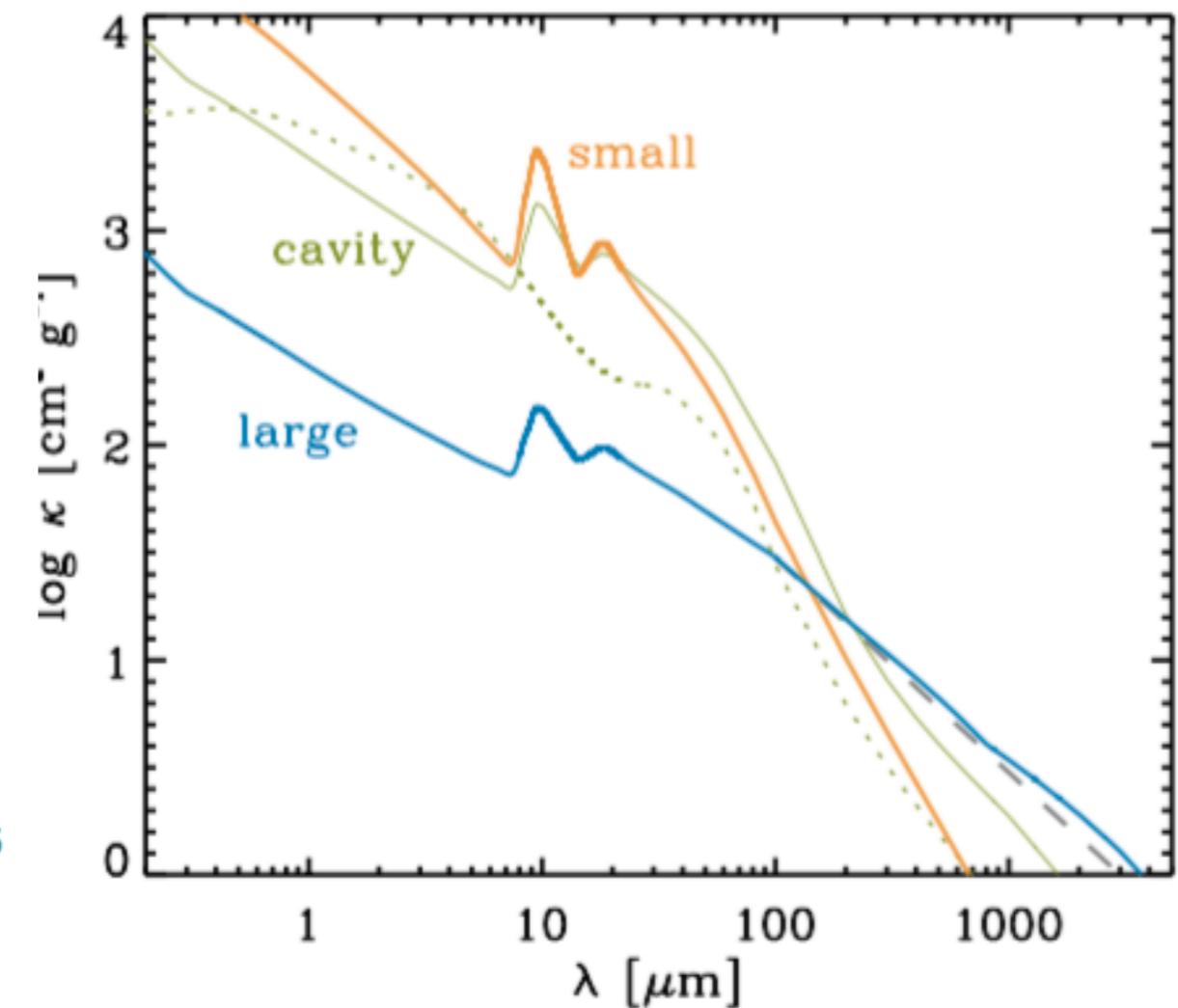
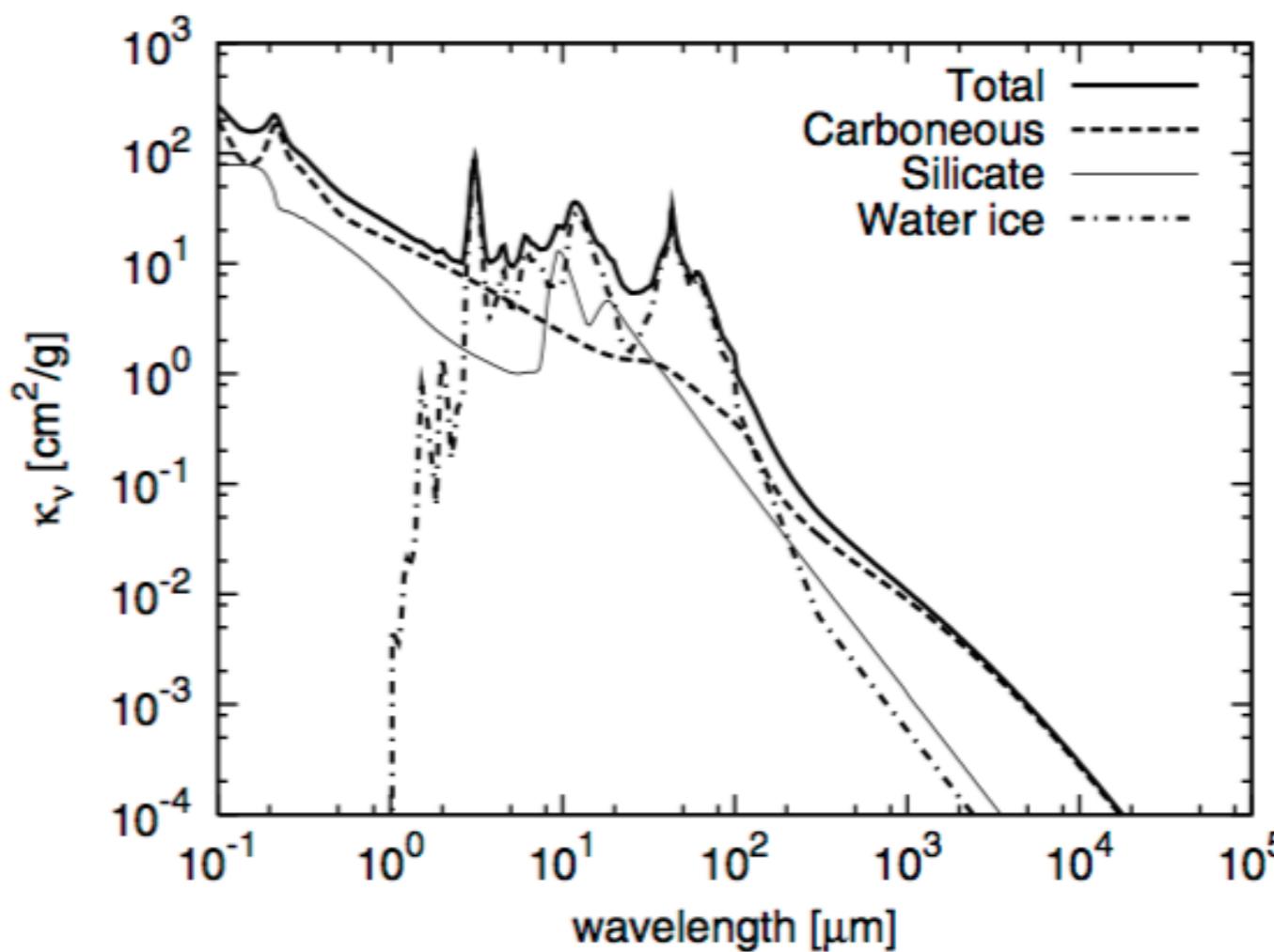
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# Dust opacity

The dust density and size distribution sets the temperature structure of the disk: dust composition and opacity are required

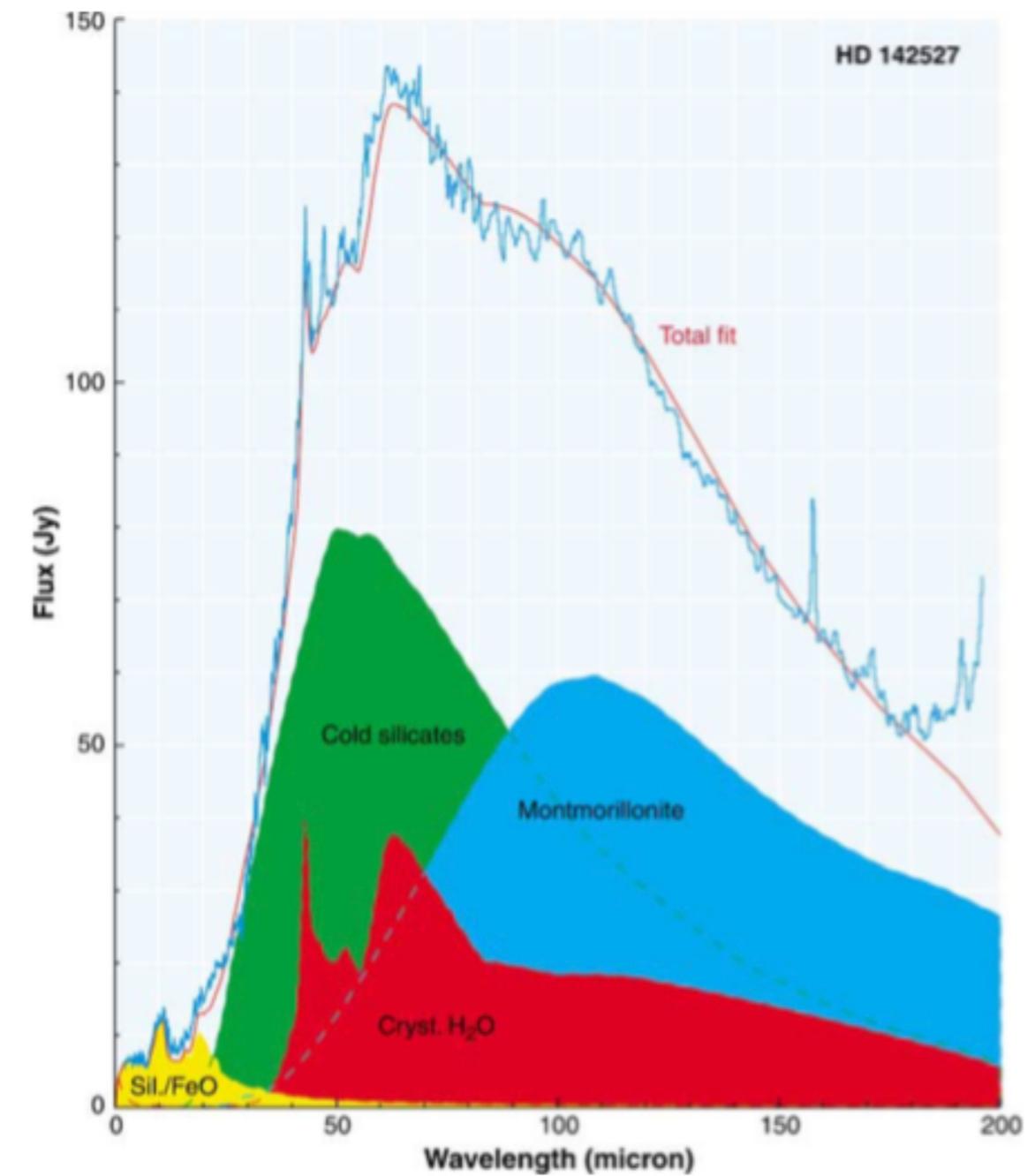
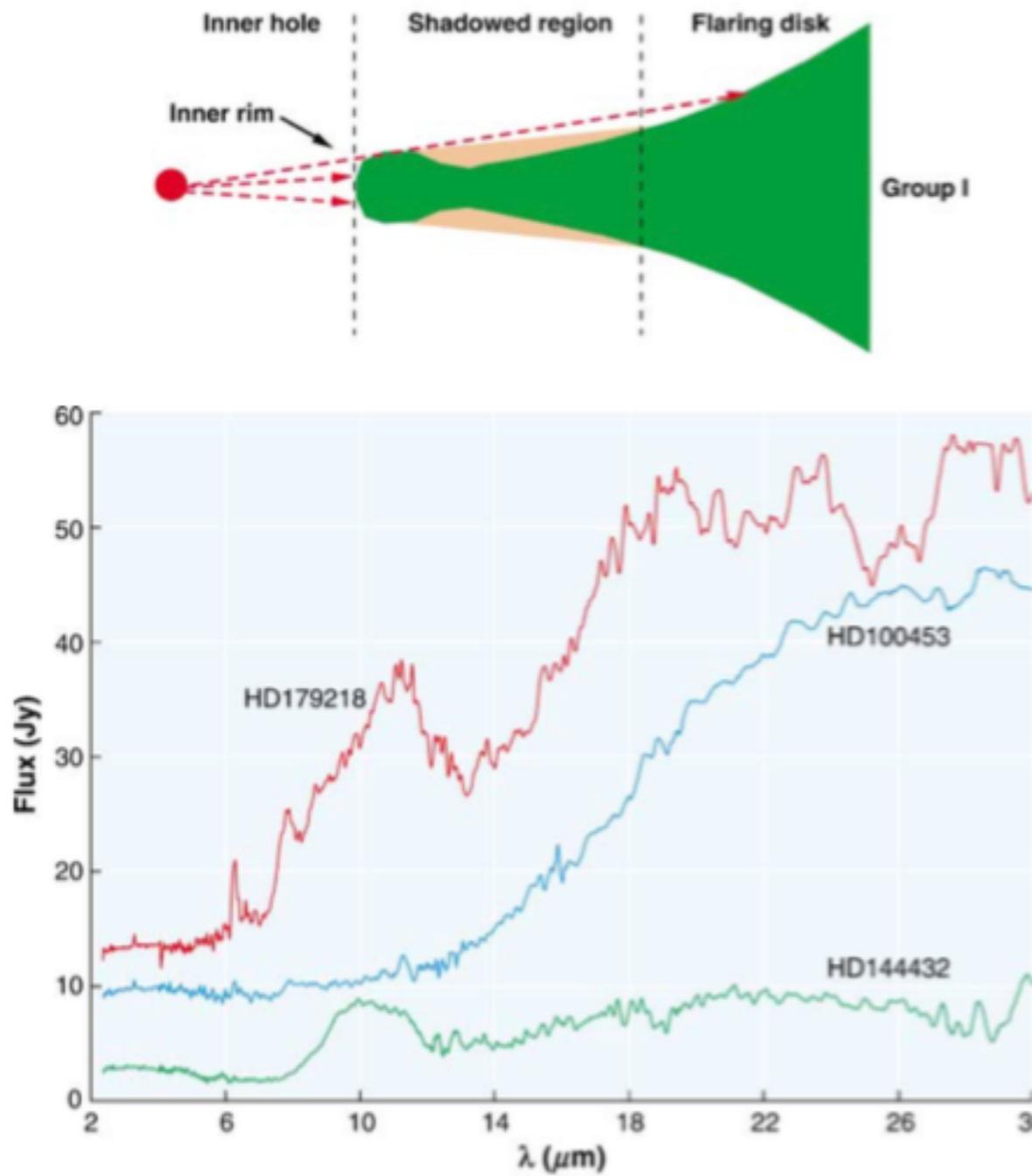


Single MRN size distribution:  
 $dn/da = C \times a^{-3.5}$

Two-populations of MRN-like grains:  
“small” and “large”

# How do we know the composition of the dust?

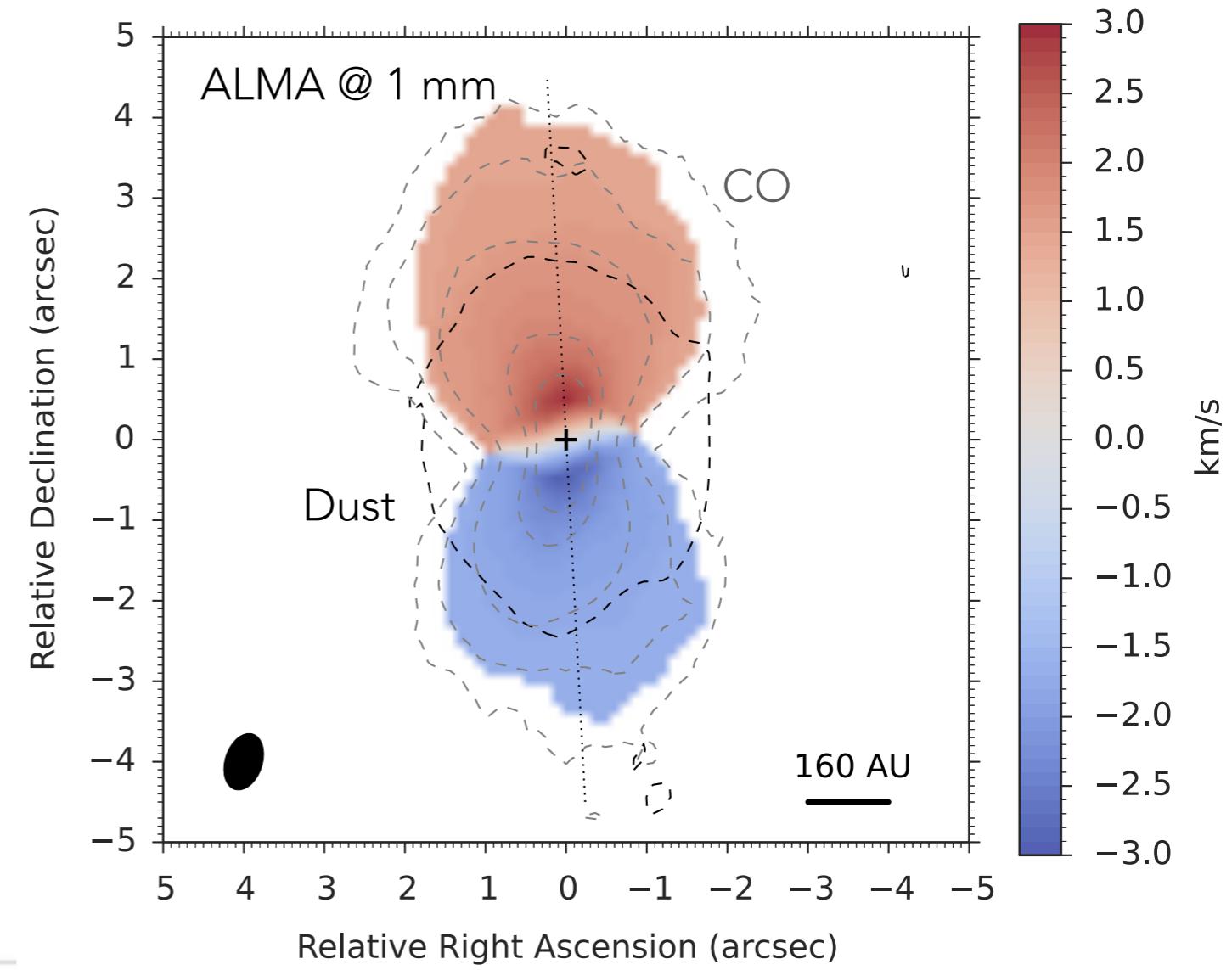
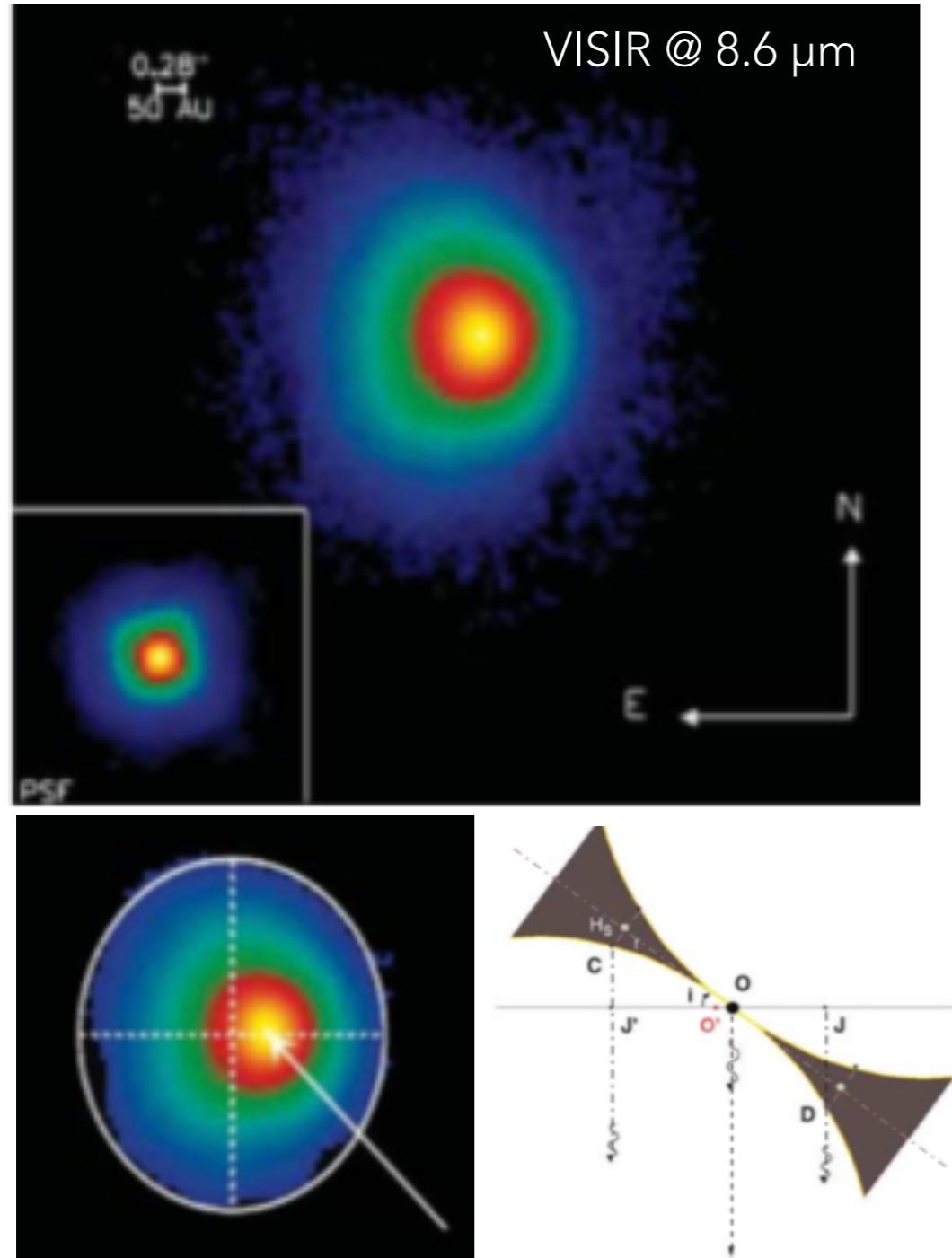
The dust emission at mid- to far-IR wavelengths shows spectral features which can be attributed to different grain components



# How do we know disks are flared?

Spatially-resolved mid-IR imaging has revealed the flared morphology of emission from small grains (PAHs) in nearby protoplanetary disks

HD 97048: a group I Herbig Ae disk

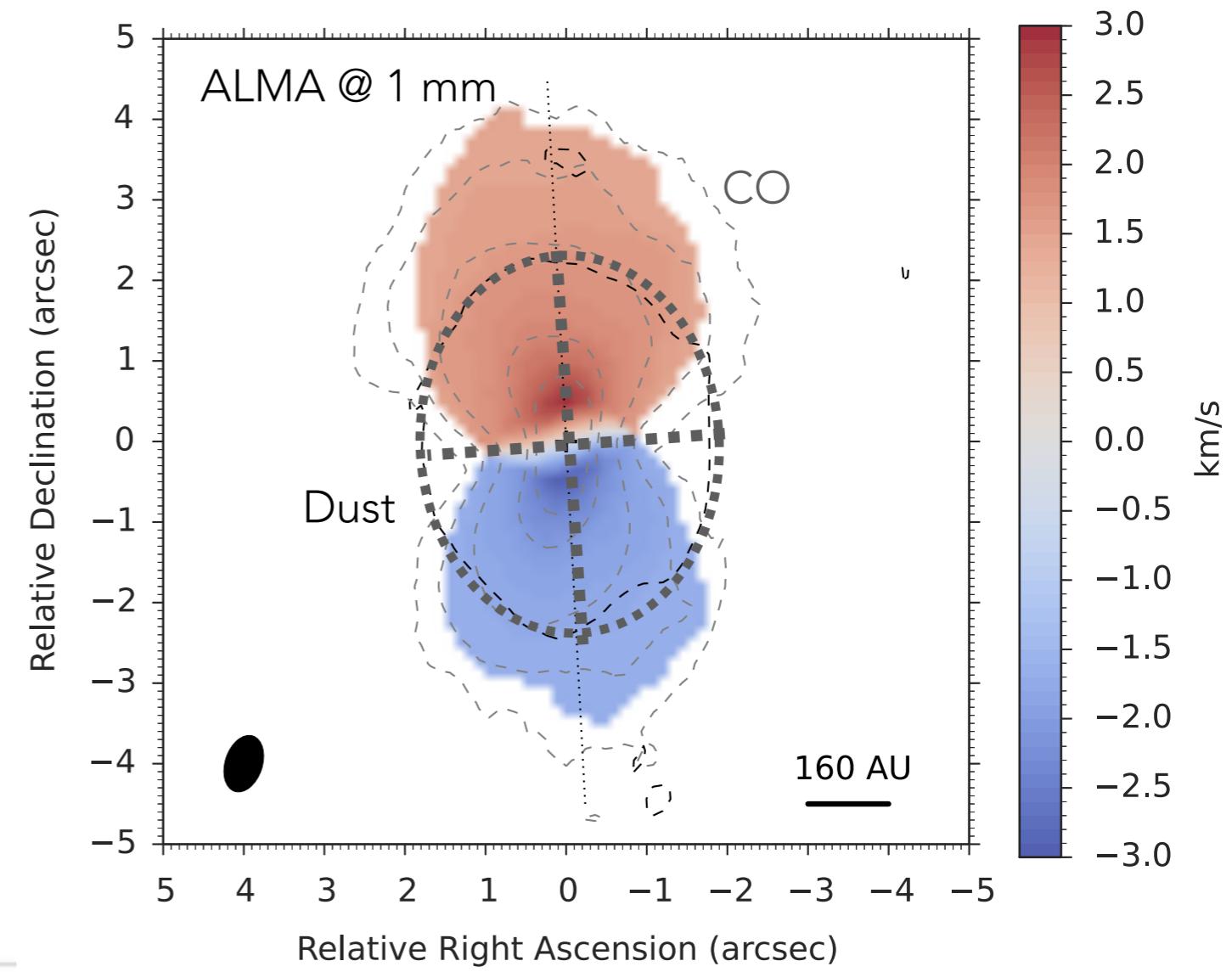
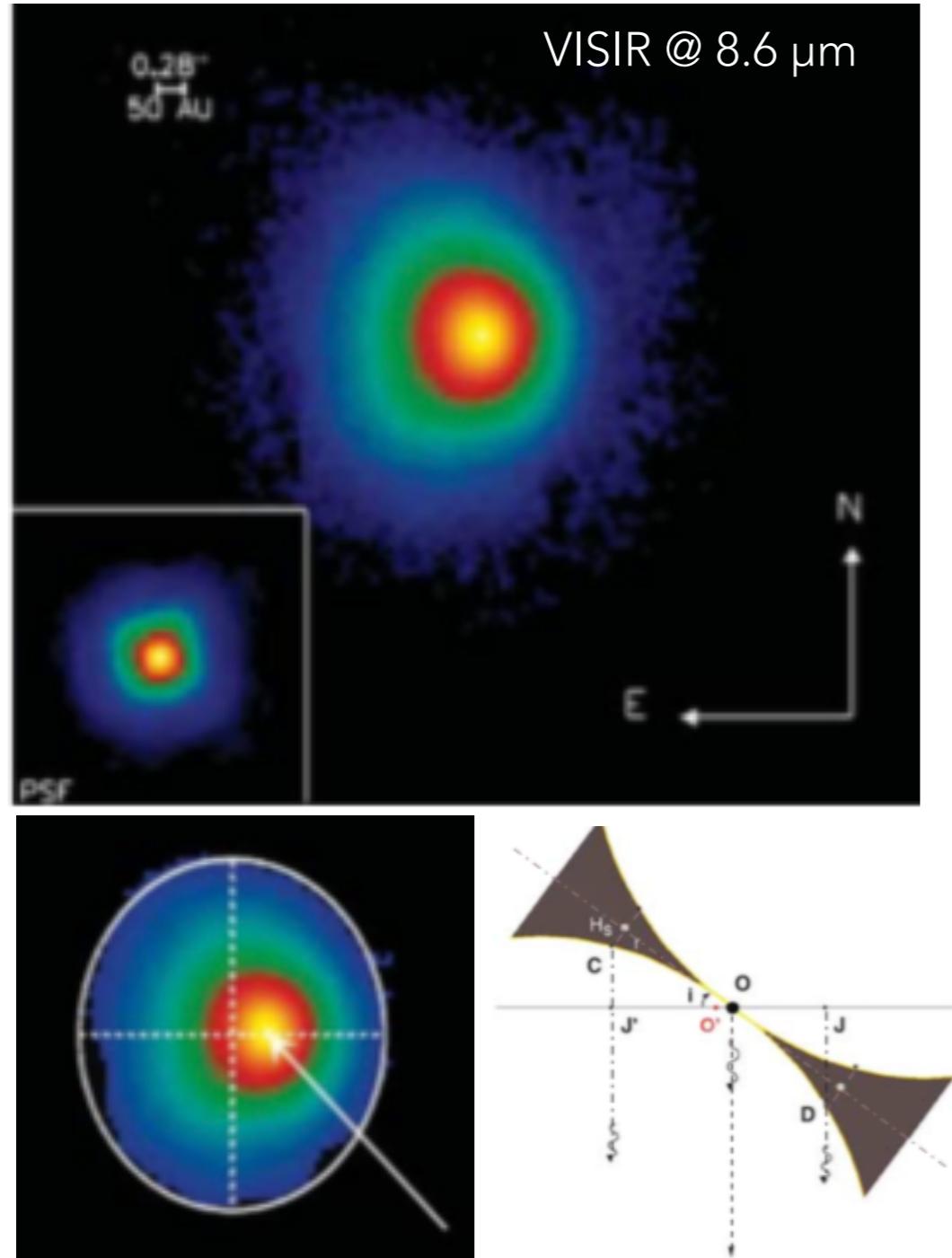


Disk emission is “flat” at (sub)mm wavelengths: evidence for settling

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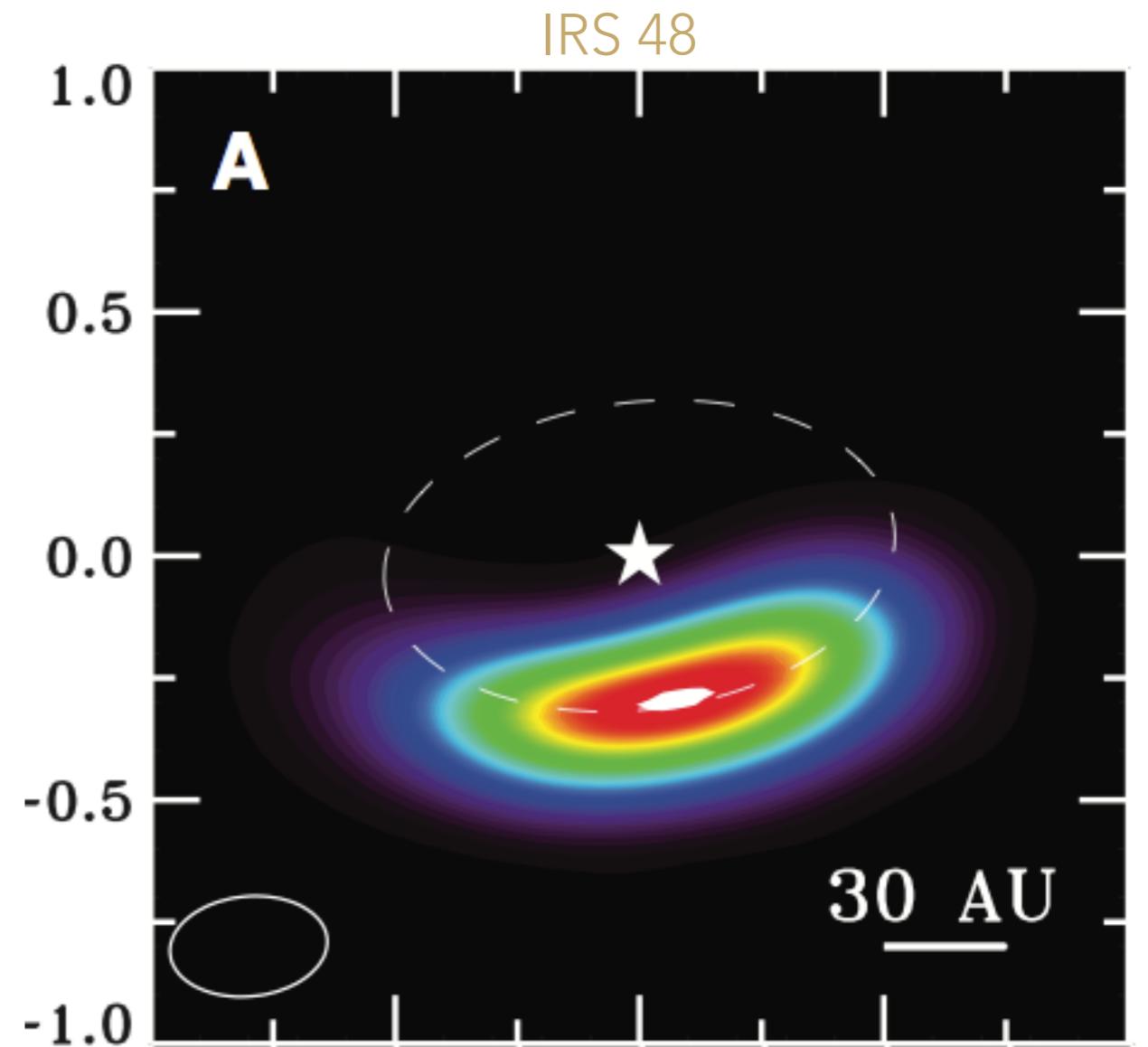
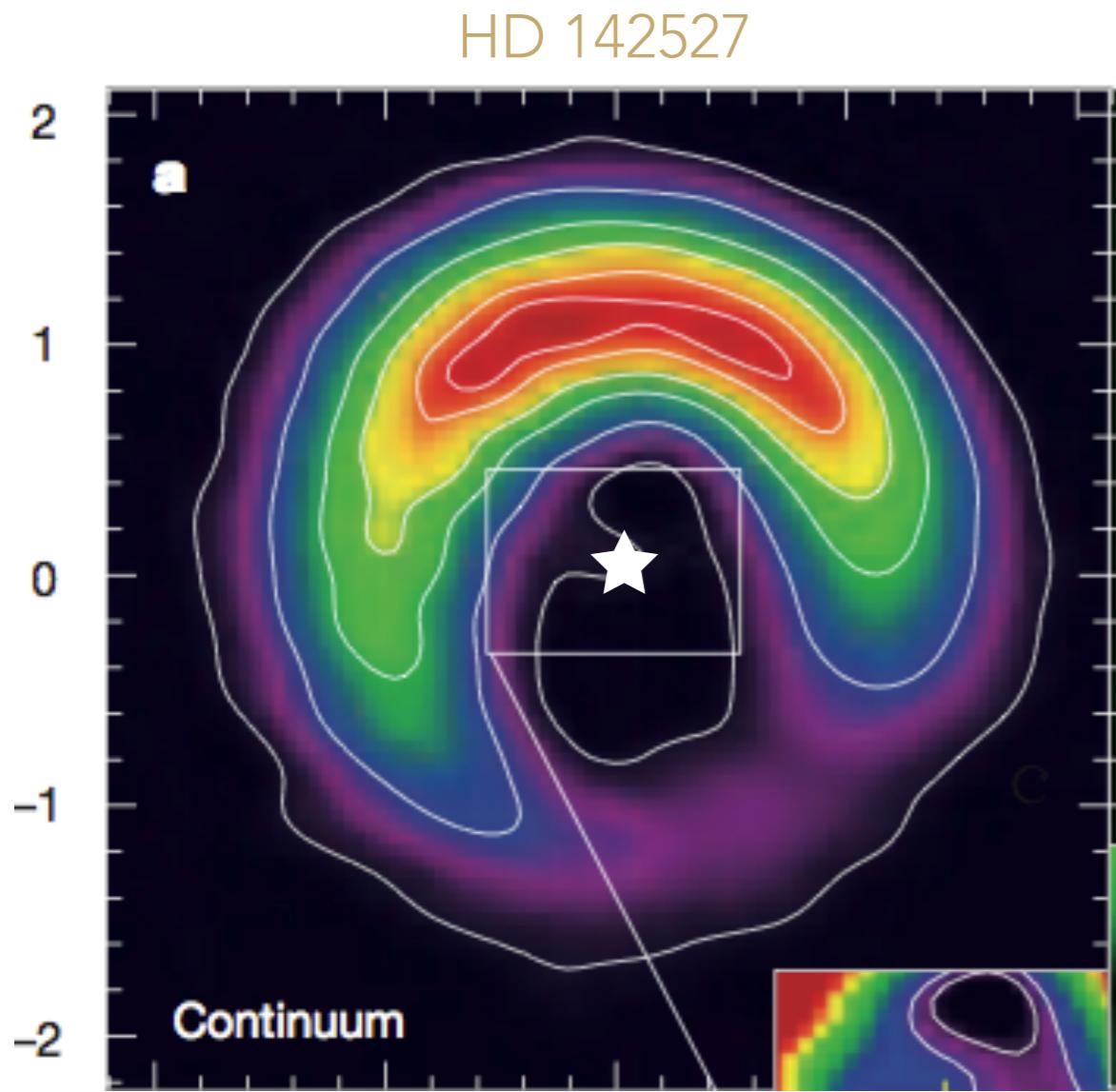
HD 97048: a group I Herbig Ae disk



Disk emission is “flat” at (sub)mm wavelengths: evidence for settling

# Dusty disks in the era of ALMA

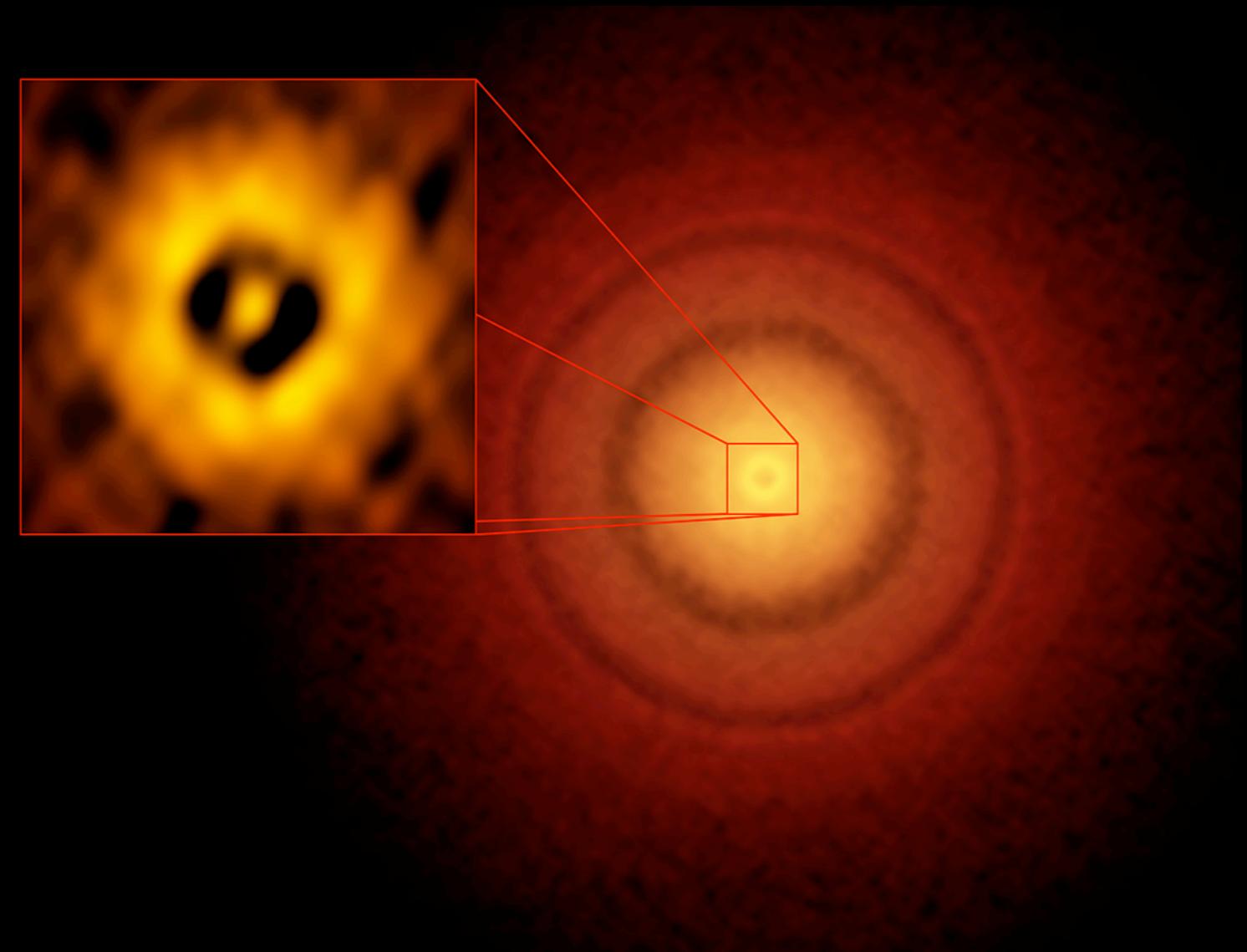
Highly asymmetric (sub)mm dust emission attributed to dust trapping in vortices potentially created by forming planets



Are these the exception rather than the norm? Both are A-type stars

# Dusty disks in the era of ALMA

Highly symmetric and concentric rings attributed to various mechanisms, including dust traps, sintering, condensation fronts, ...

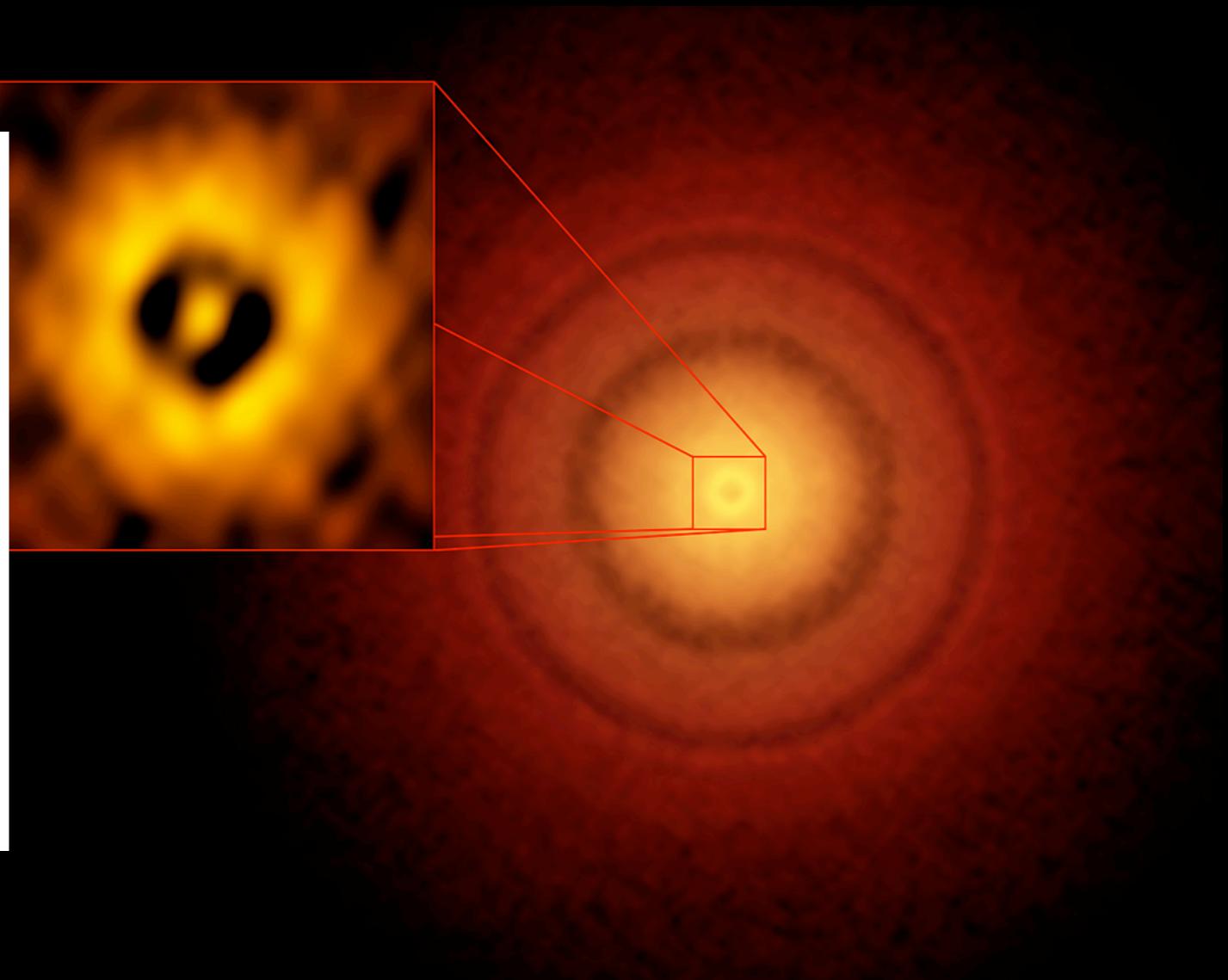
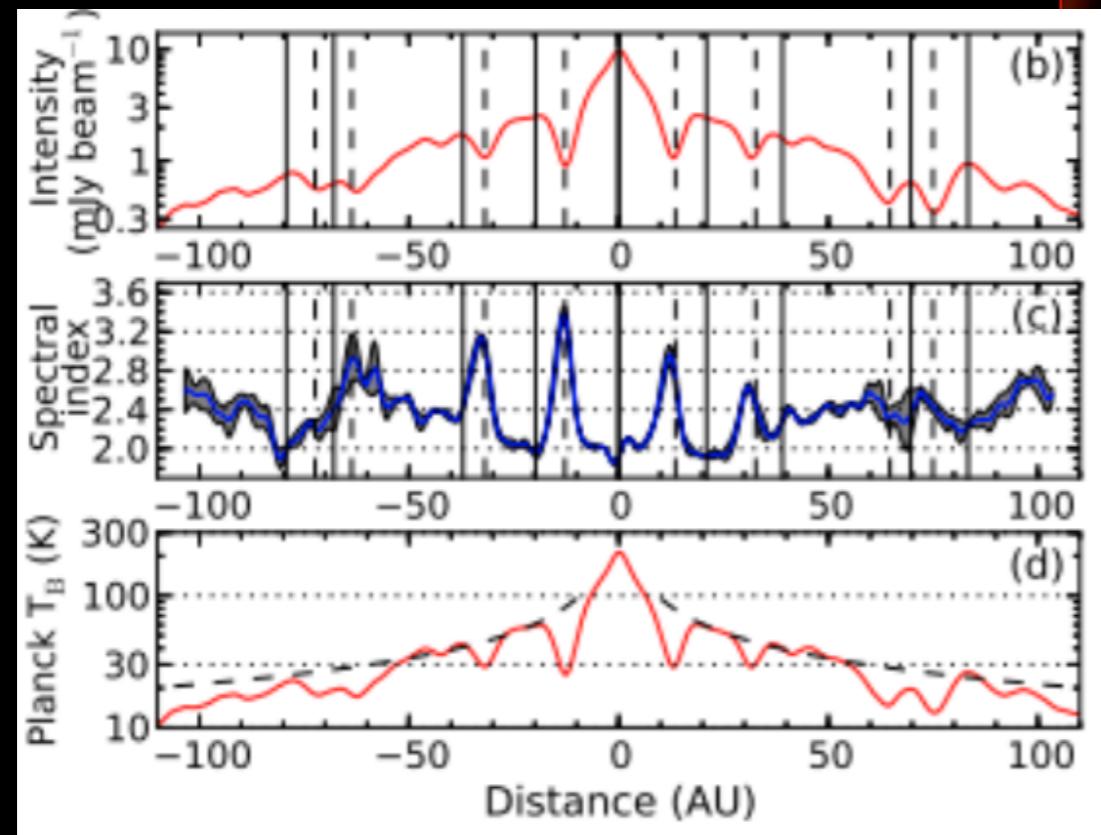


HL Tau

TW Hya

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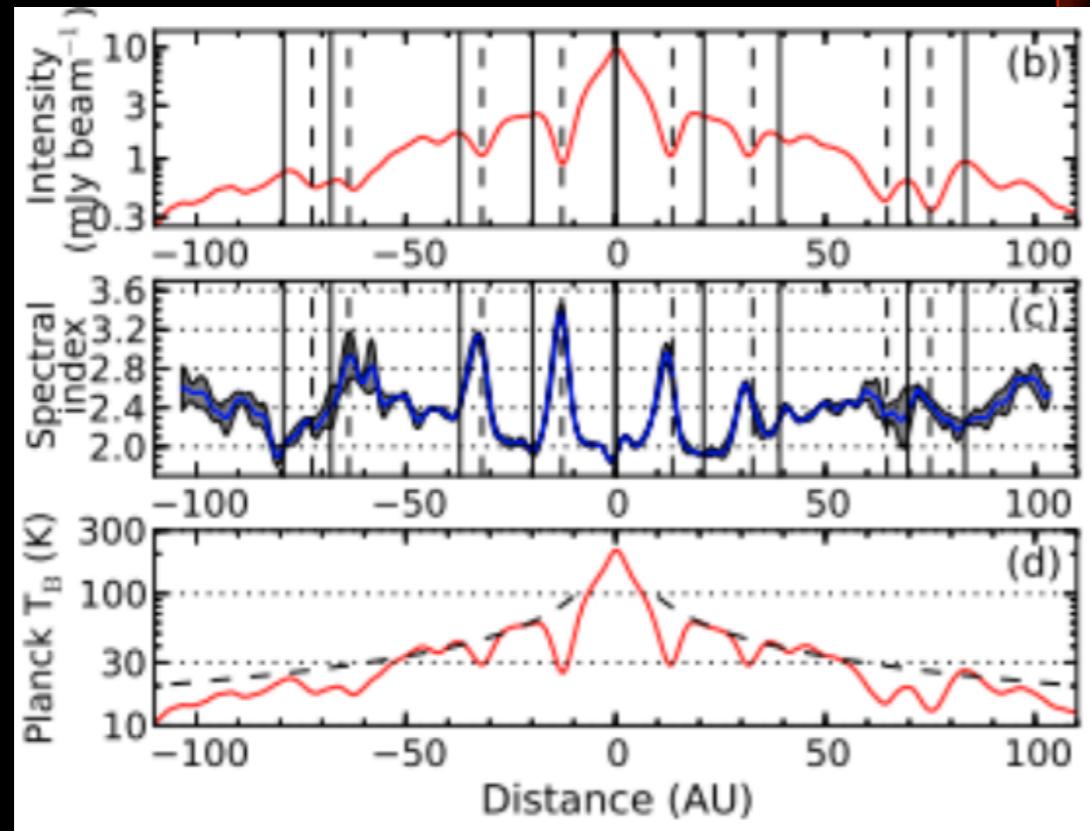


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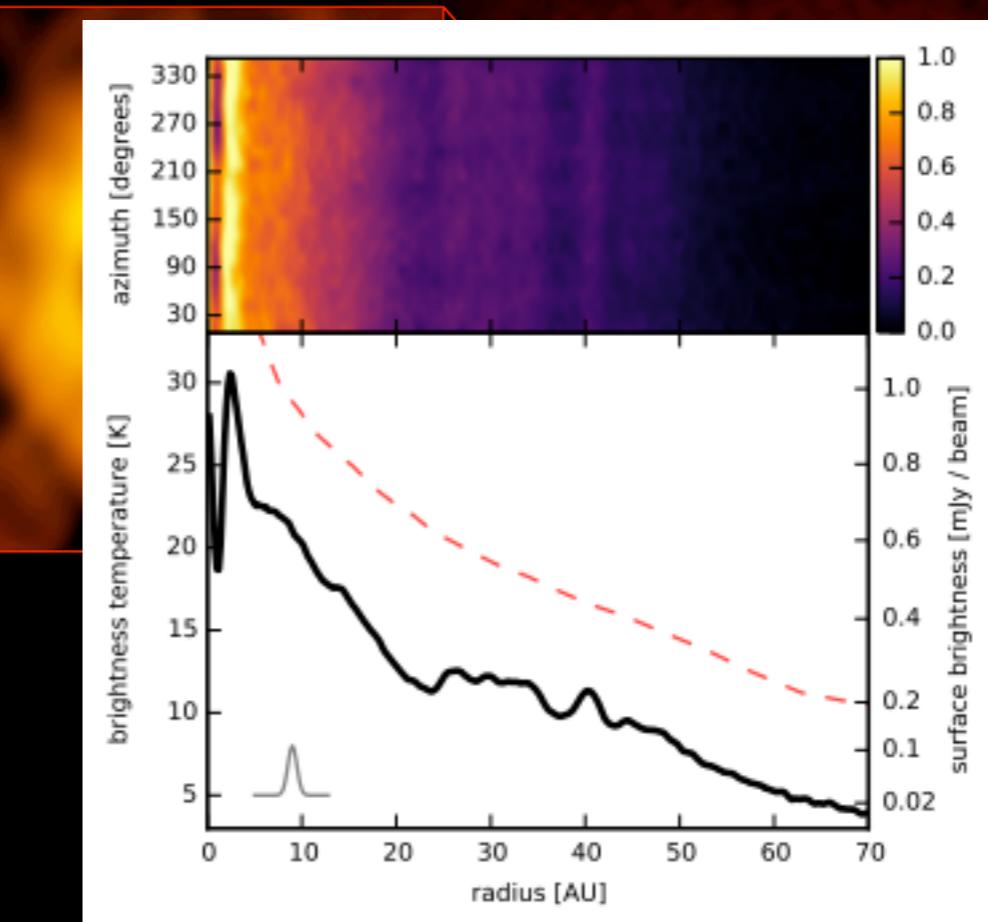
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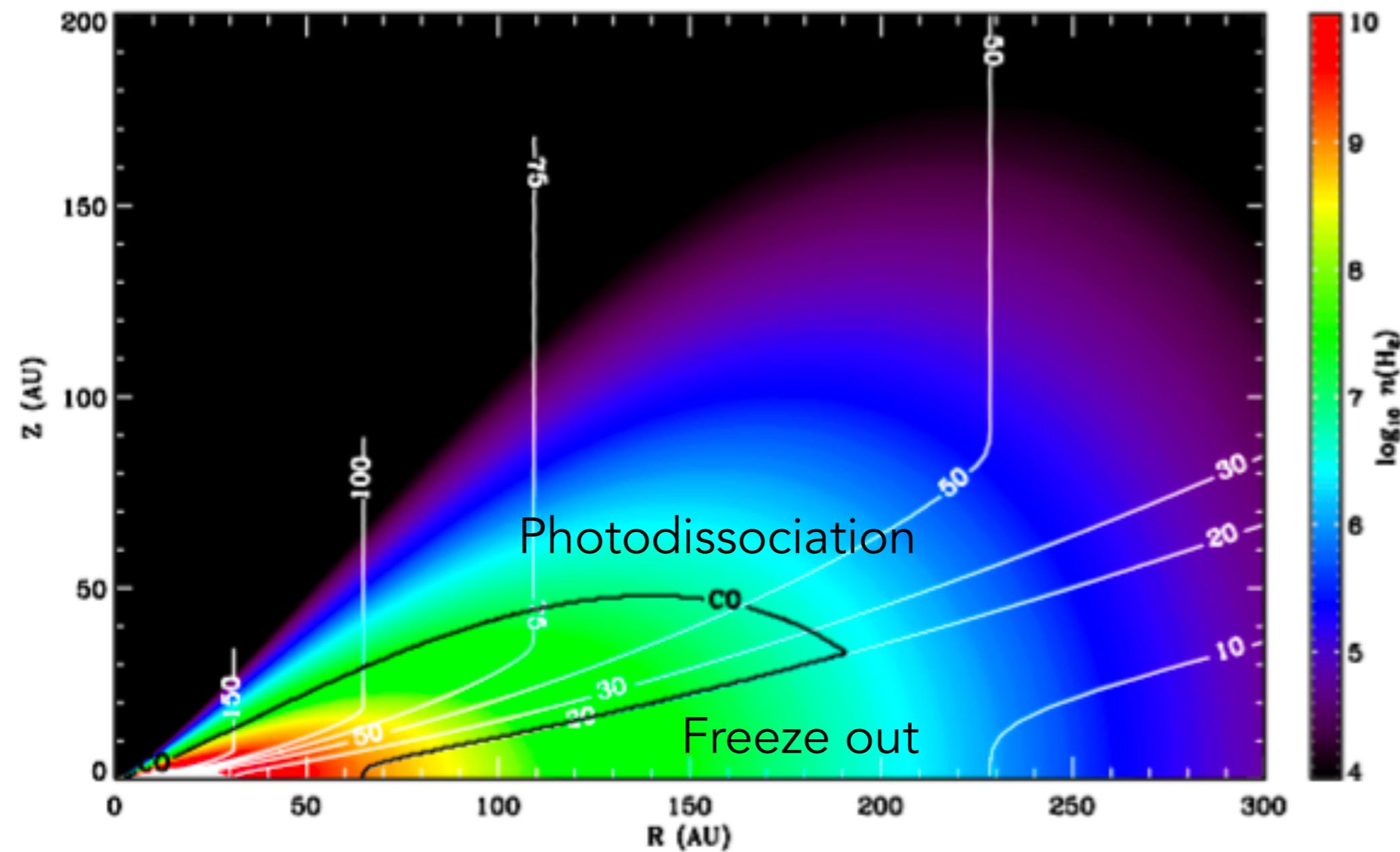
HL Tau



TW Hya

# What about the gas?

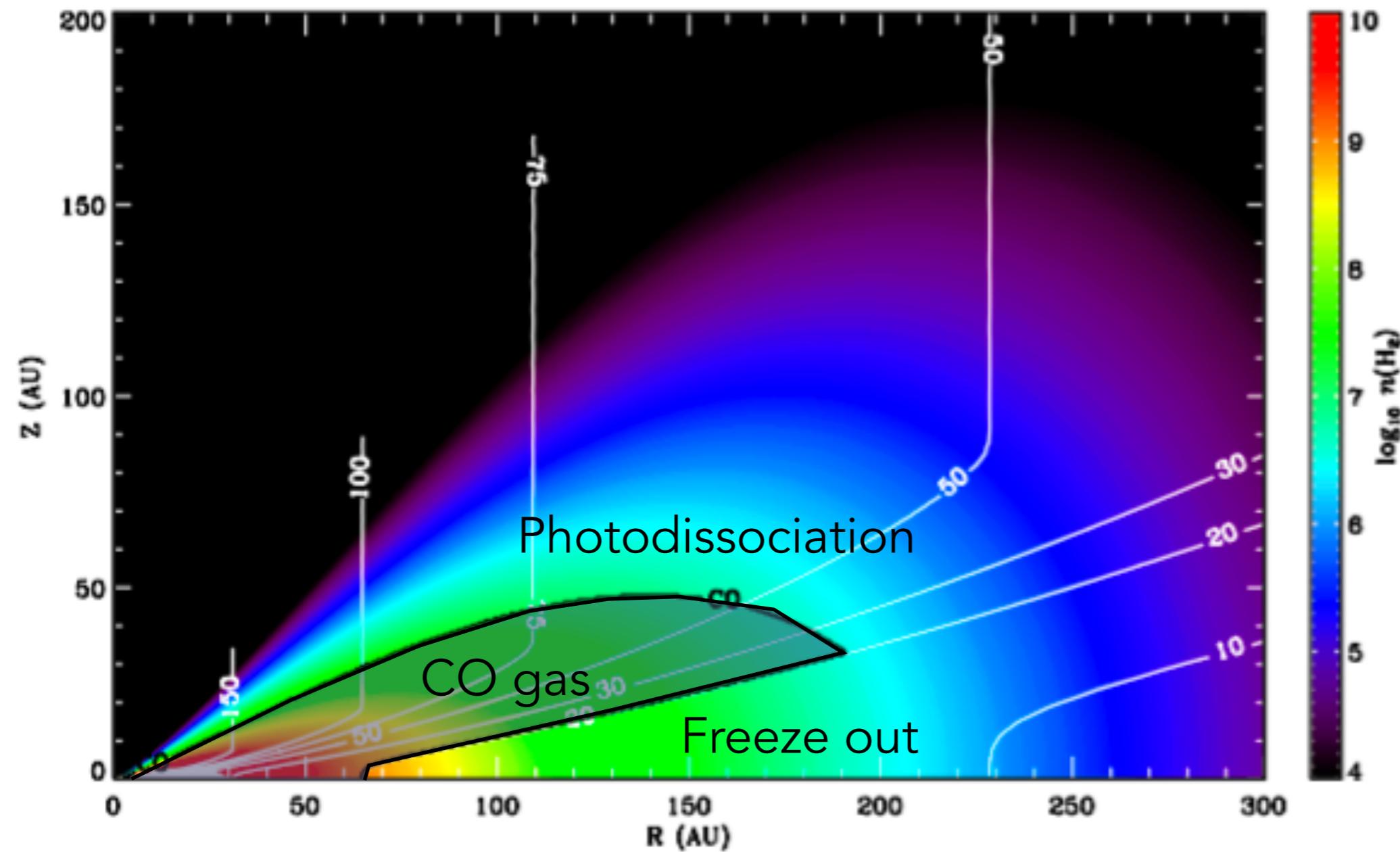
Gas is mainly H<sub>2</sub> (90%) and He (10%) which are difficult to observe: CO (~0.01%) is used as a proxy as the second-most abundant molecule



Warning! CO gas can have a complex distribution due to the disk structure

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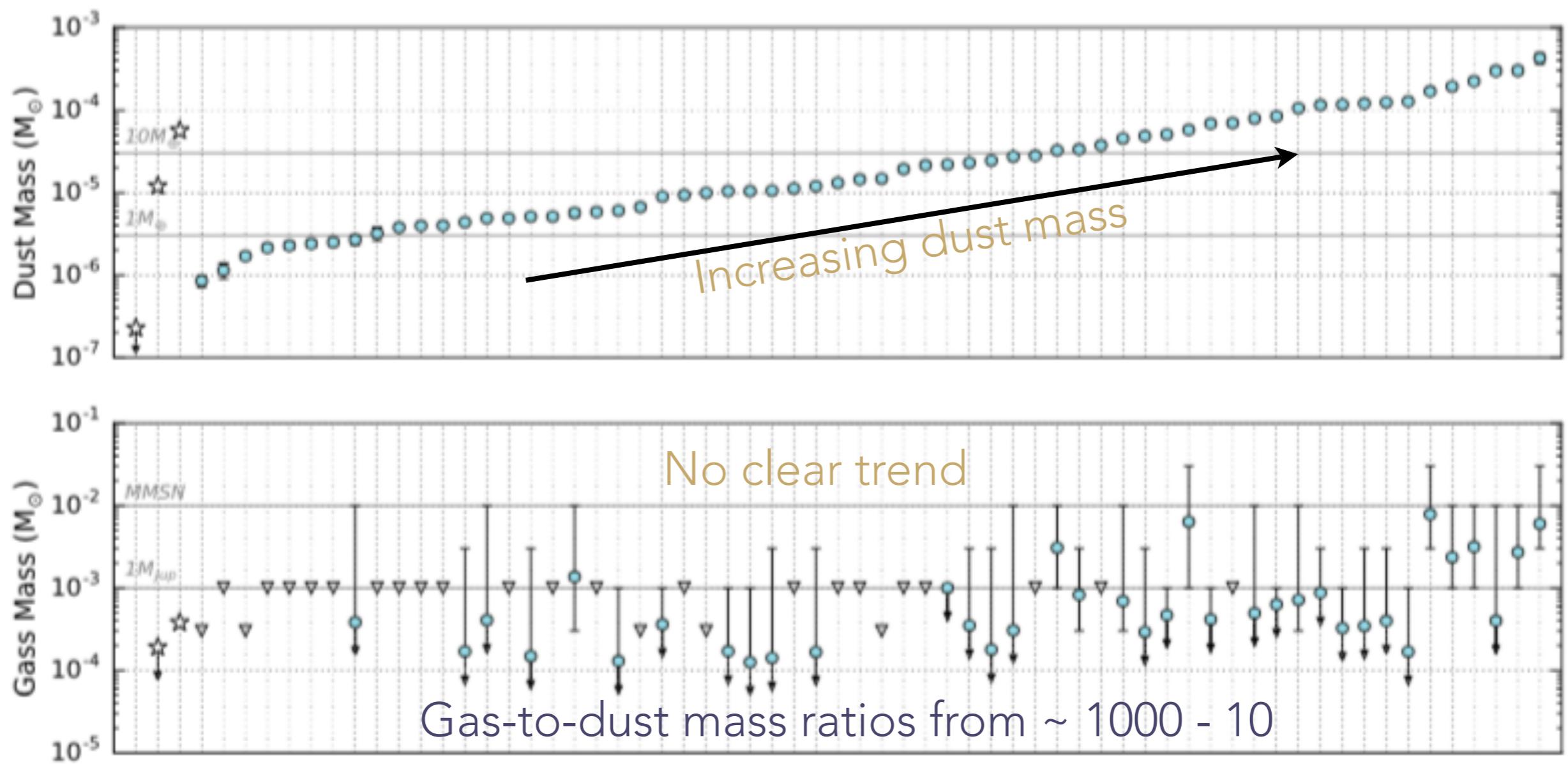
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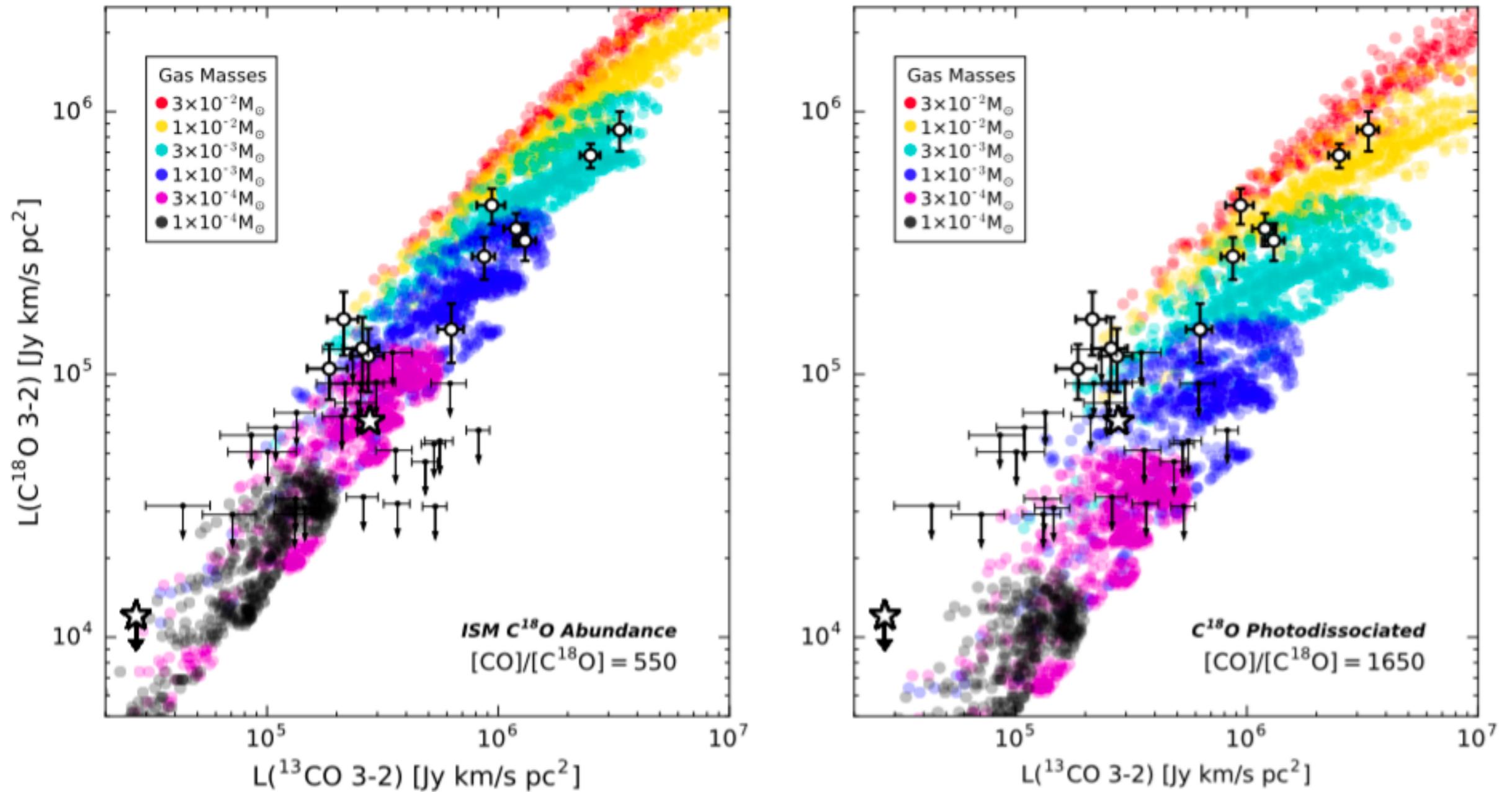
Emission from the main CO isotopologues ( $^{12}\text{CO}$  and  $^{13}\text{CO}$ ) are optically thick, hence less abundant isotopologues are used ( $\text{C}^{18}\text{O}$  and  $\text{C}^{17}\text{O}$ )



Dust and gas masses from the ALMA Survey of Lupus

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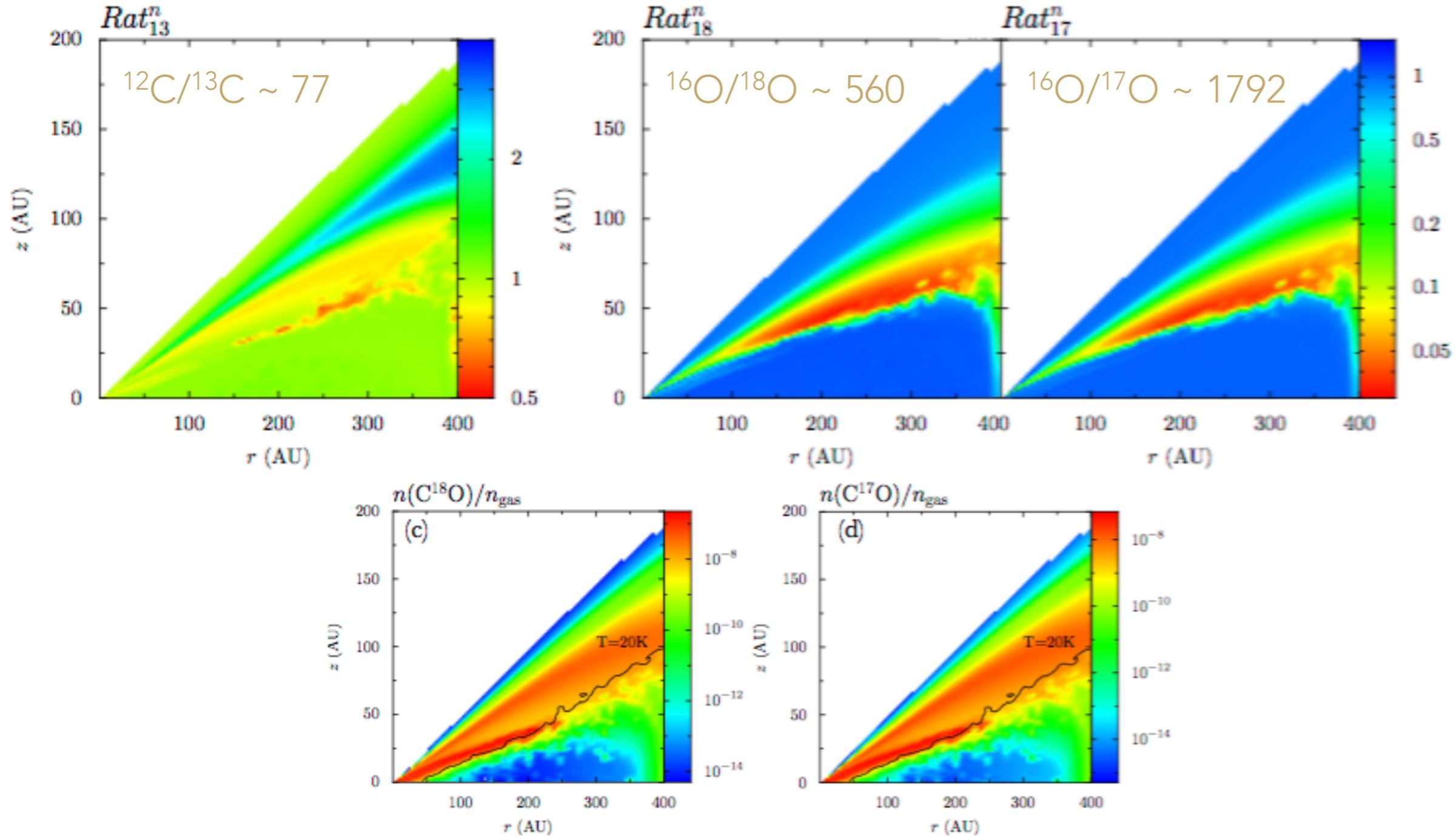
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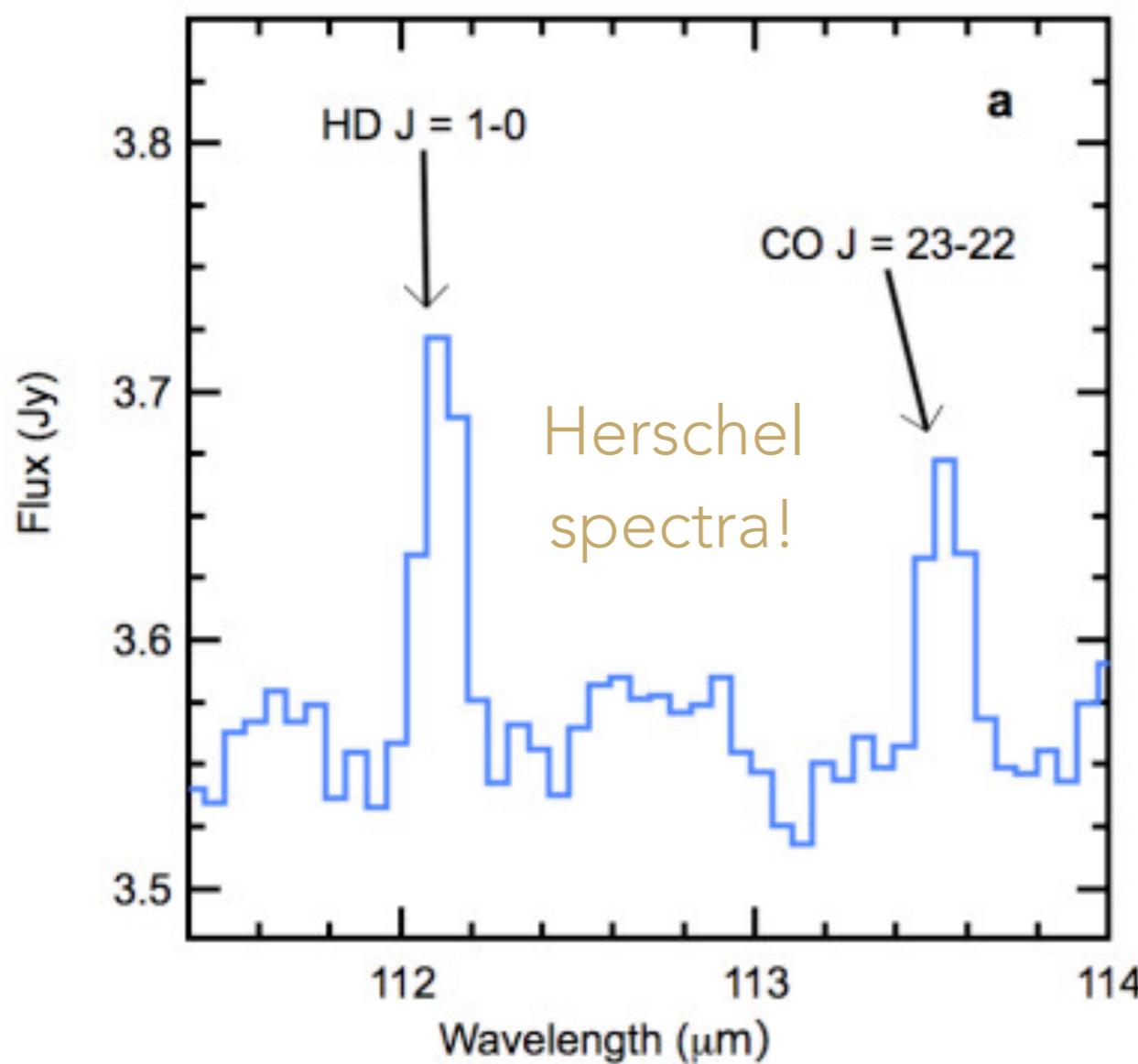
# What about the gas?

Picture is further complicated by chemical effects, namely, isotope-selective photodissociation

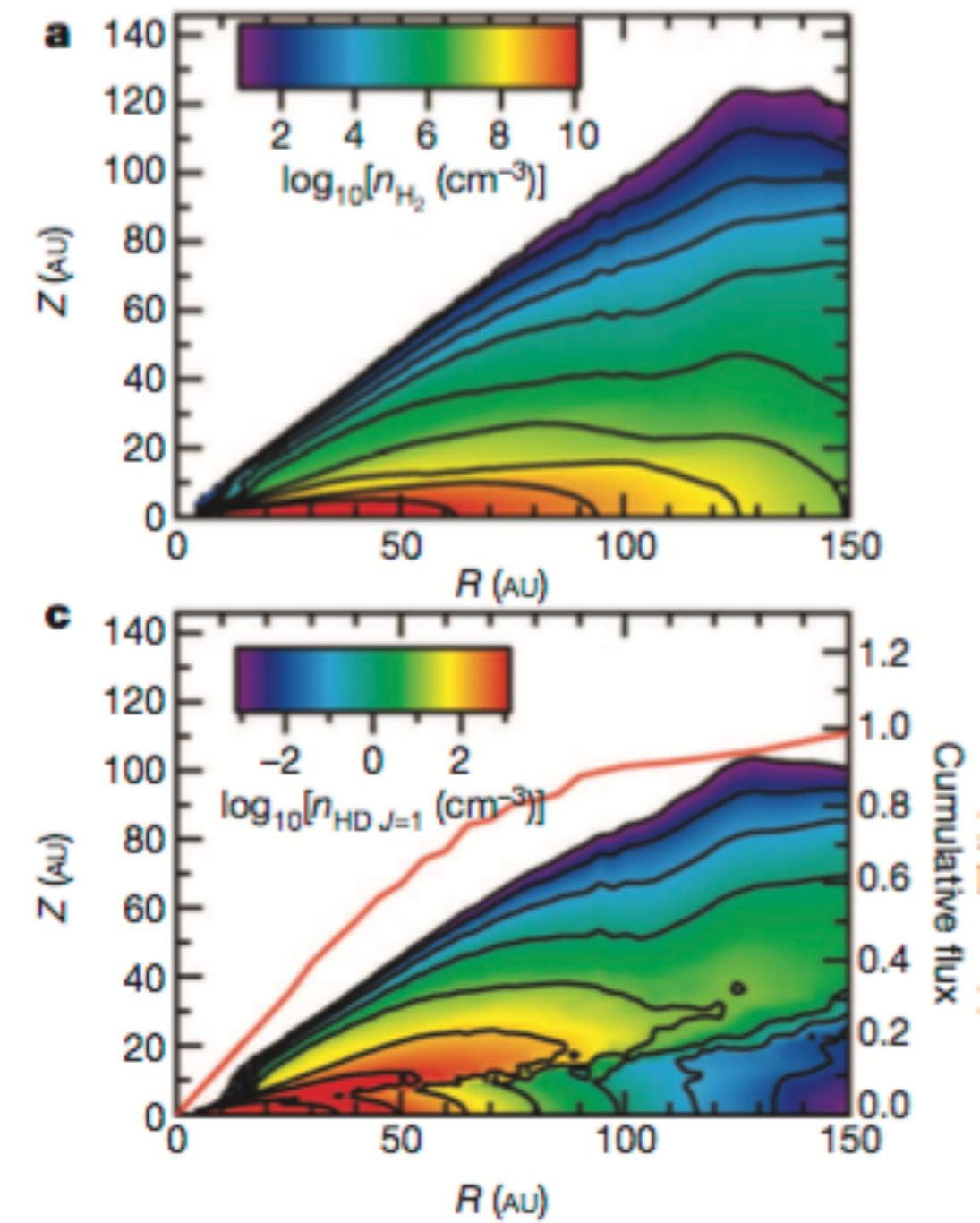


# What about the gas?

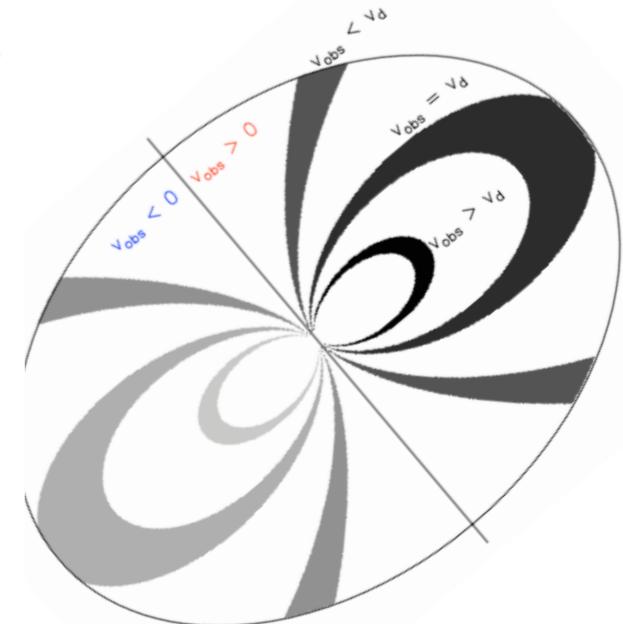
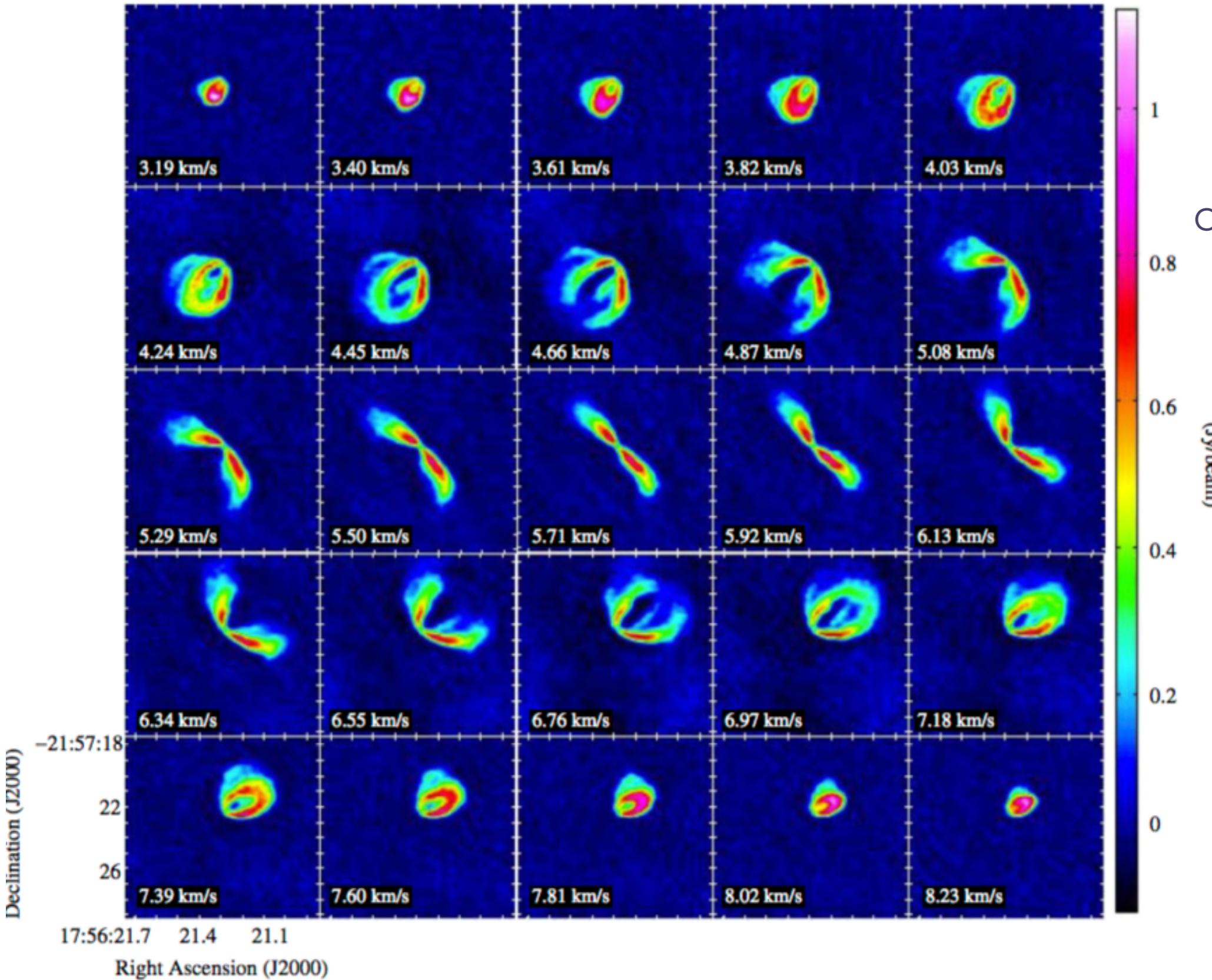
Given the relative complexity of interpreting CO observations, HD ( $\sim 0.001\%$  of H<sub>2</sub>) is proposed as an alternative tracer of the gas mass



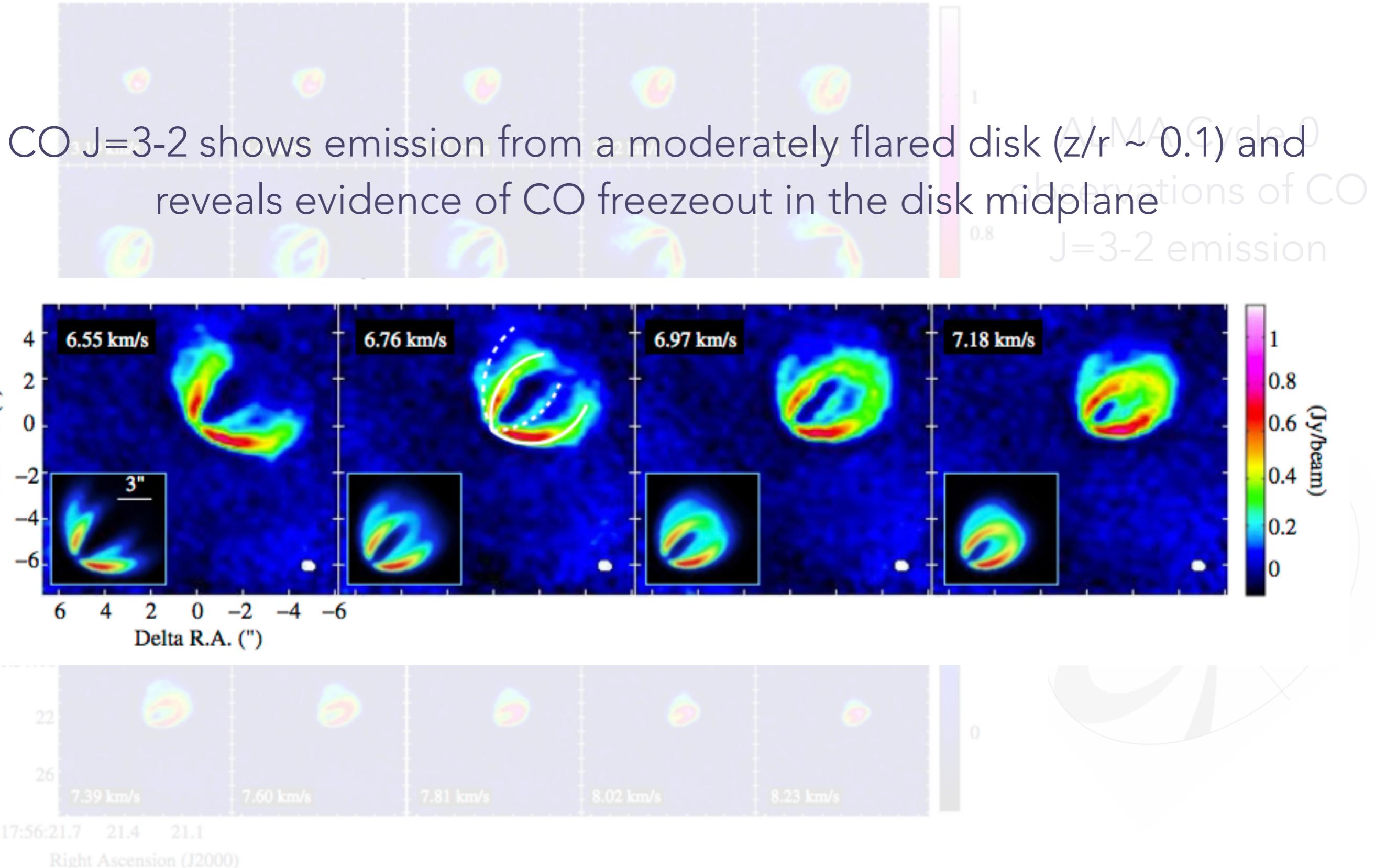
Disk mass (TW Hya)  $> 0.05 M_{\text{sol}}$



# CO line emission in the ALMA era

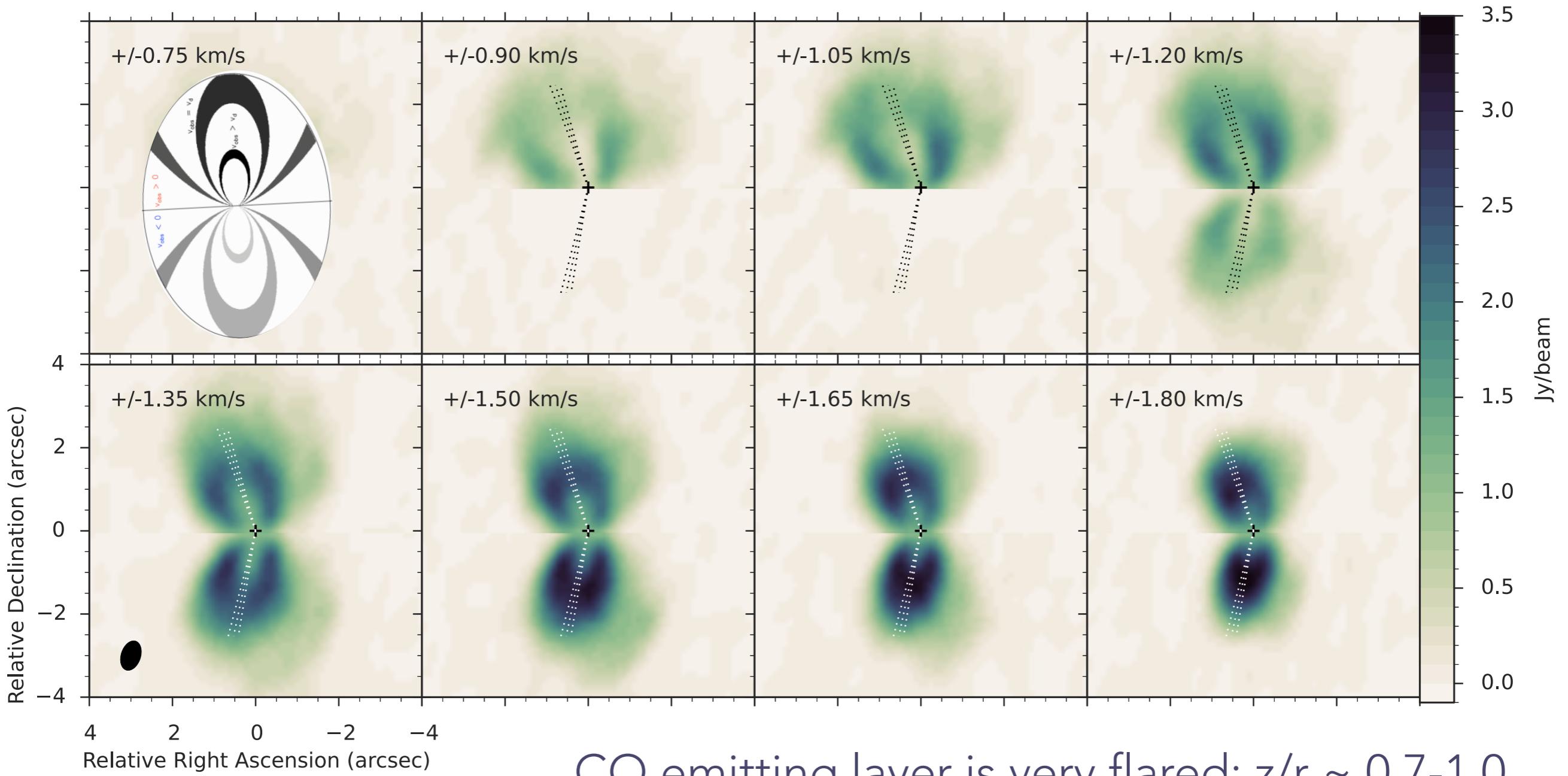


# CO line emission in the ALMA era



# CO line emission in the ALMA era

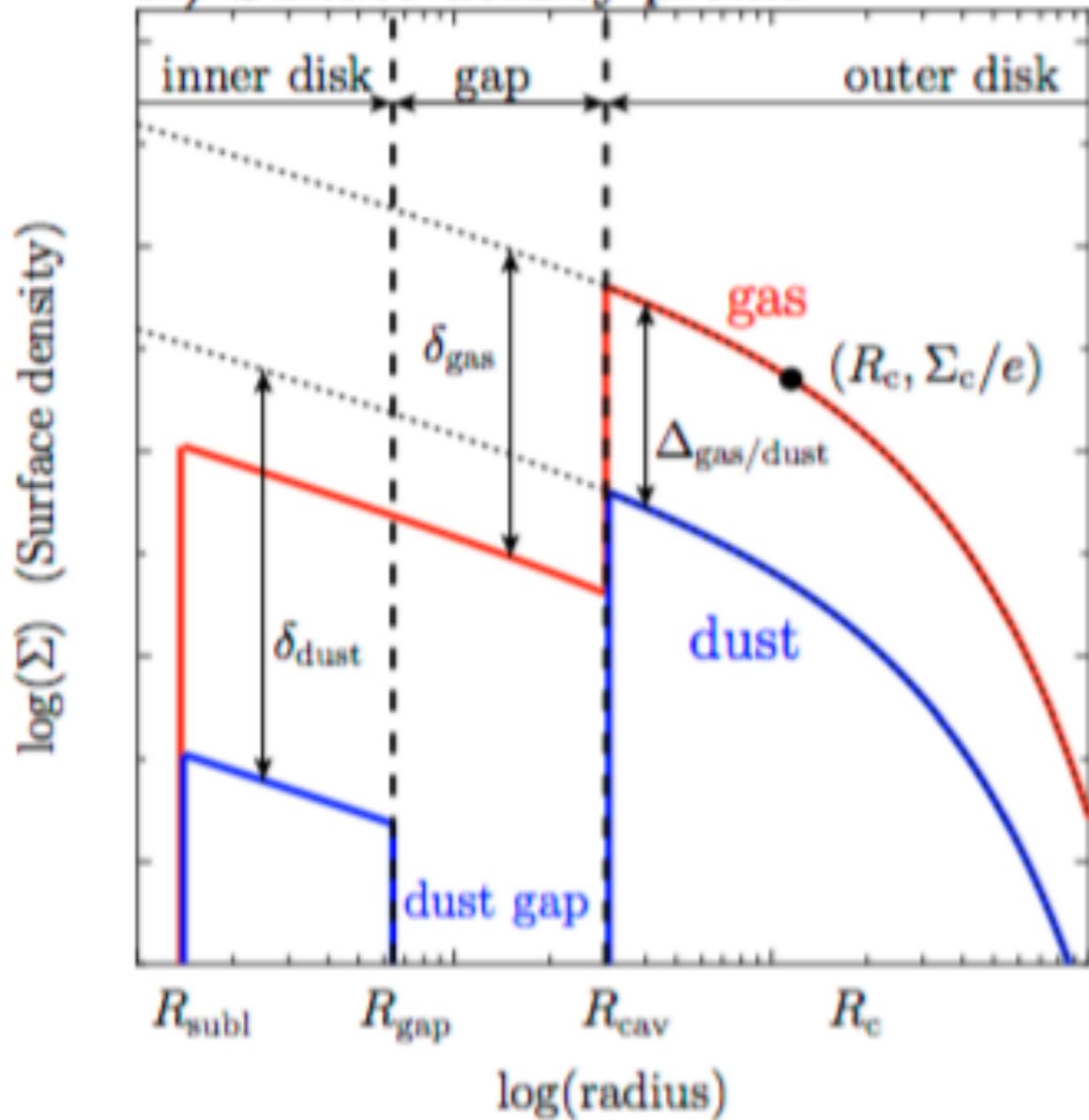
ALMA Cycle 0 observations of CO J=3-2 emission from HD 97048



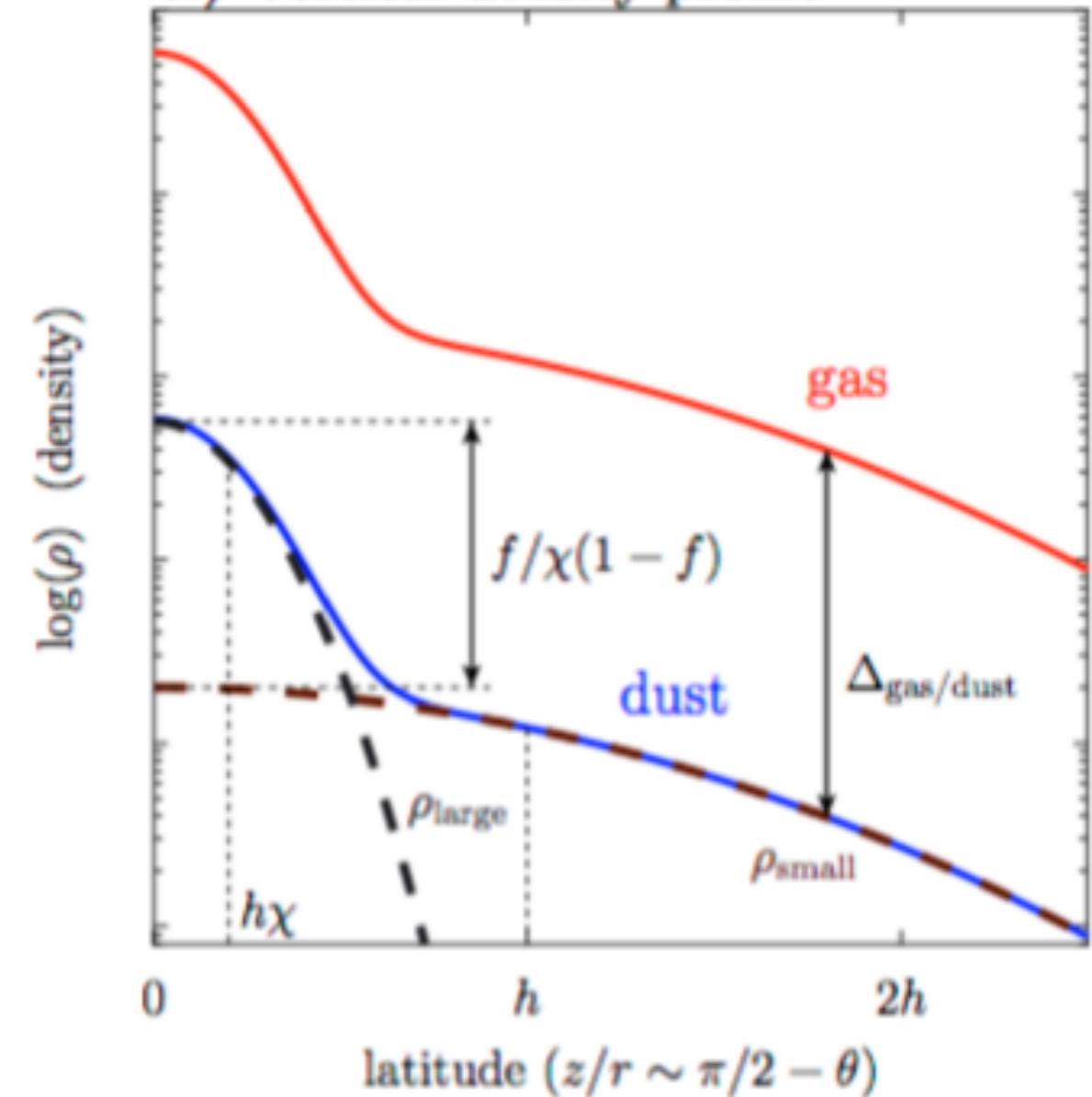
# Modelling the gas and dust in tandem

Gas surface density still assumed to follow the dust: gas-to-dust mass ratio is now a “free” (yet still global) parameter

a.) Surface density profile

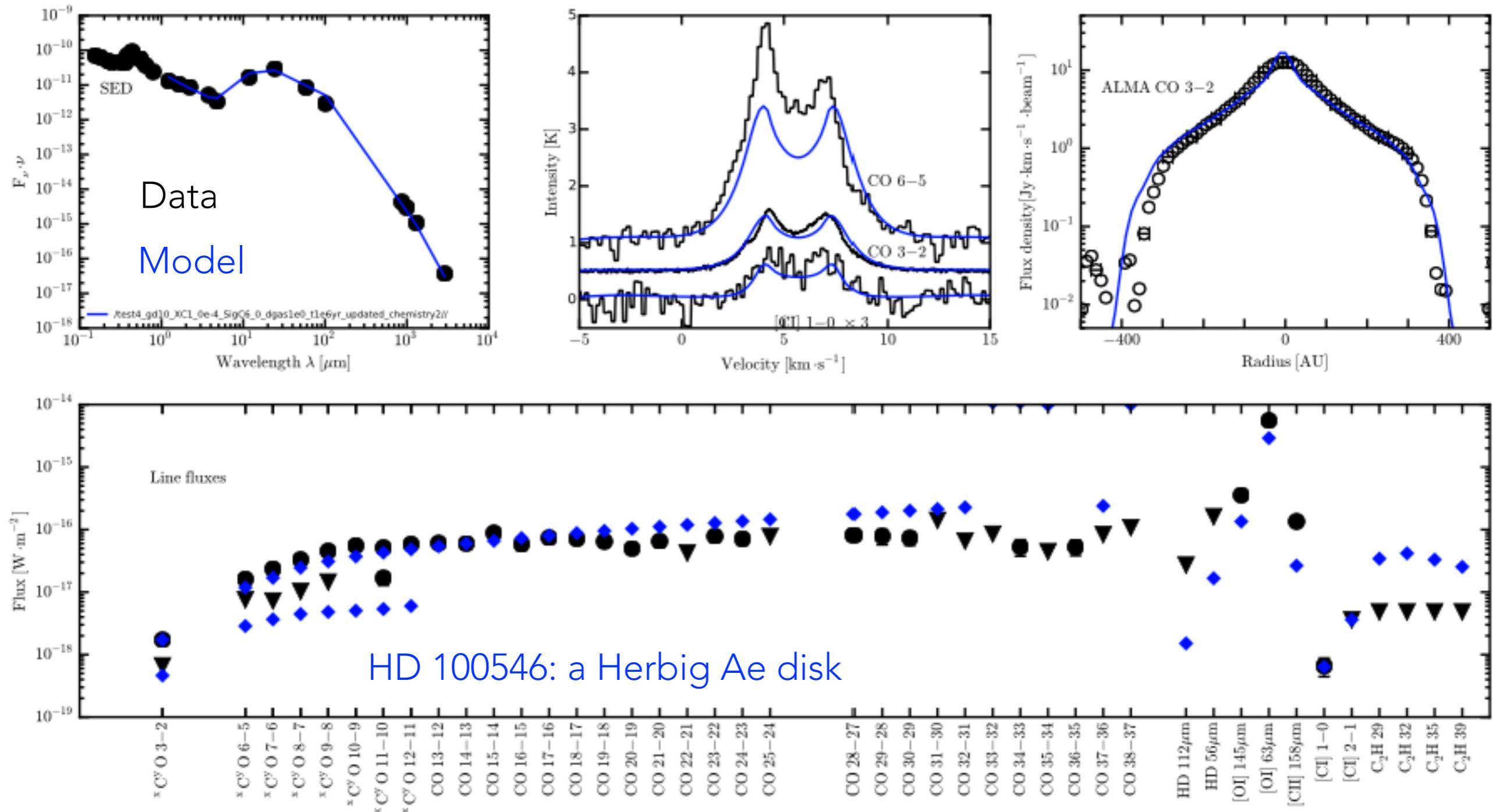


b.) Vertical density profile



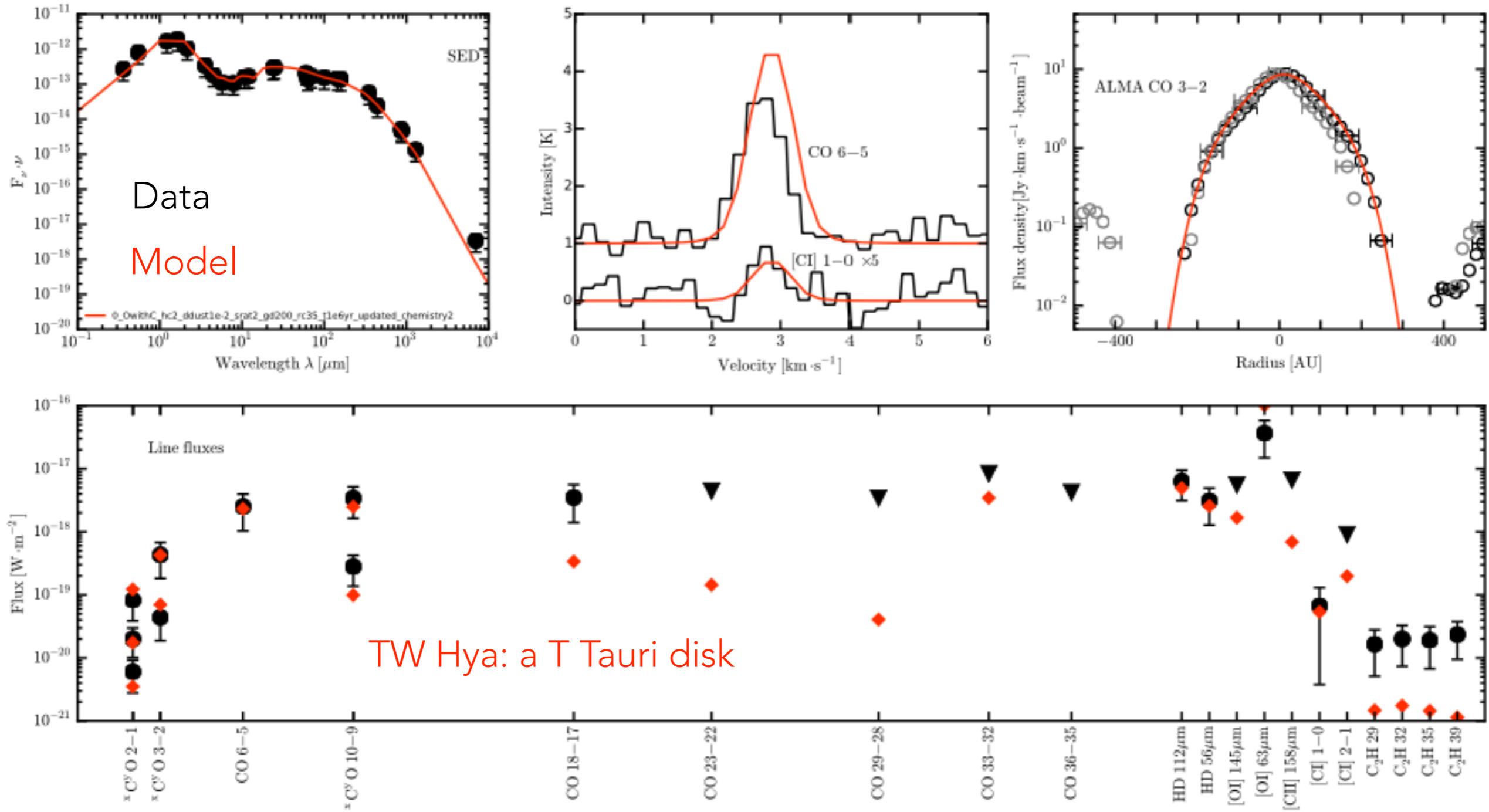
# Towards a global prescription of gas and dust

Modern models now fit the dust SED and spectrally and spatially resolved molecular line observations simultaneously

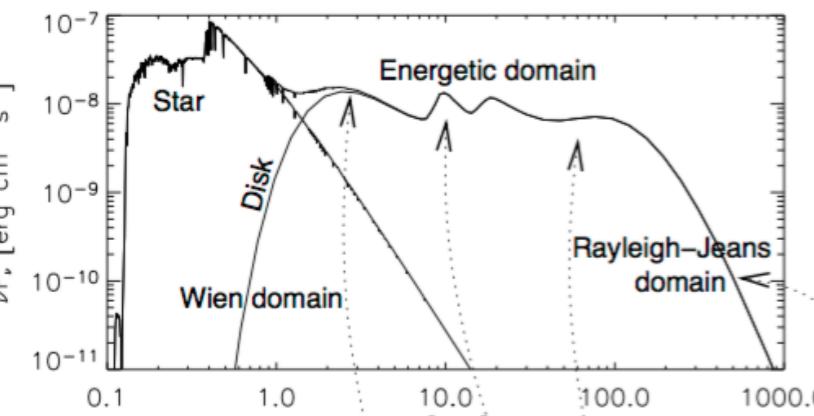


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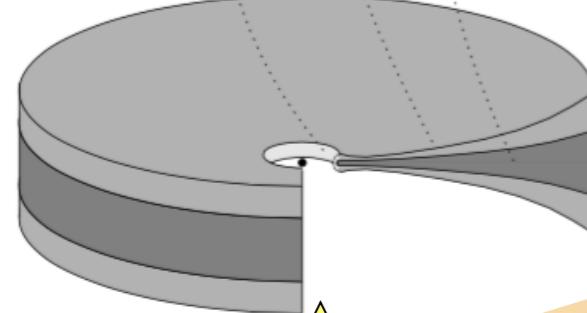
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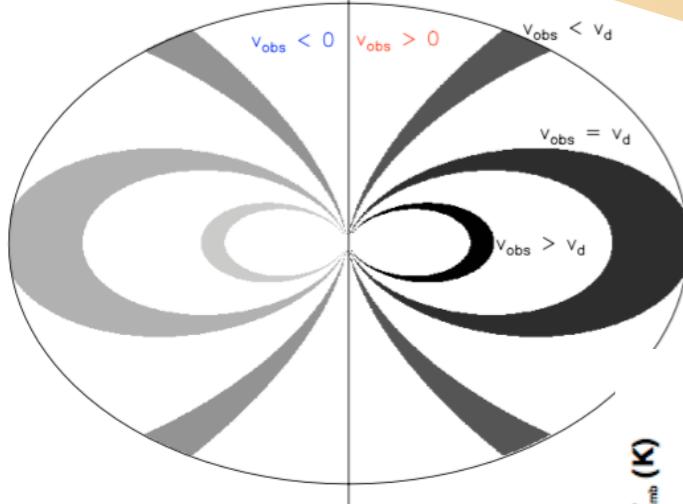
# Observations of protoplanetary disks



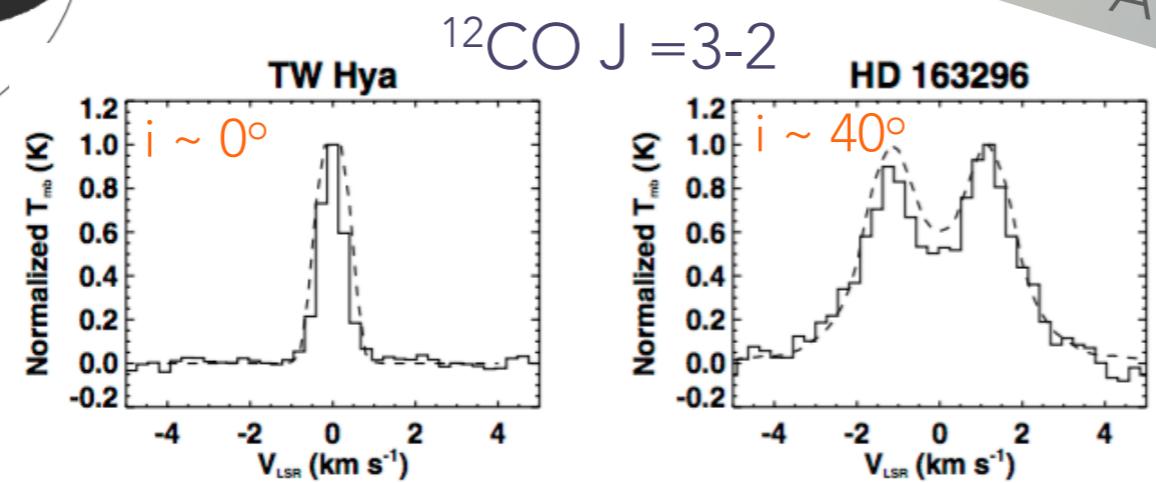
Dust



Gas

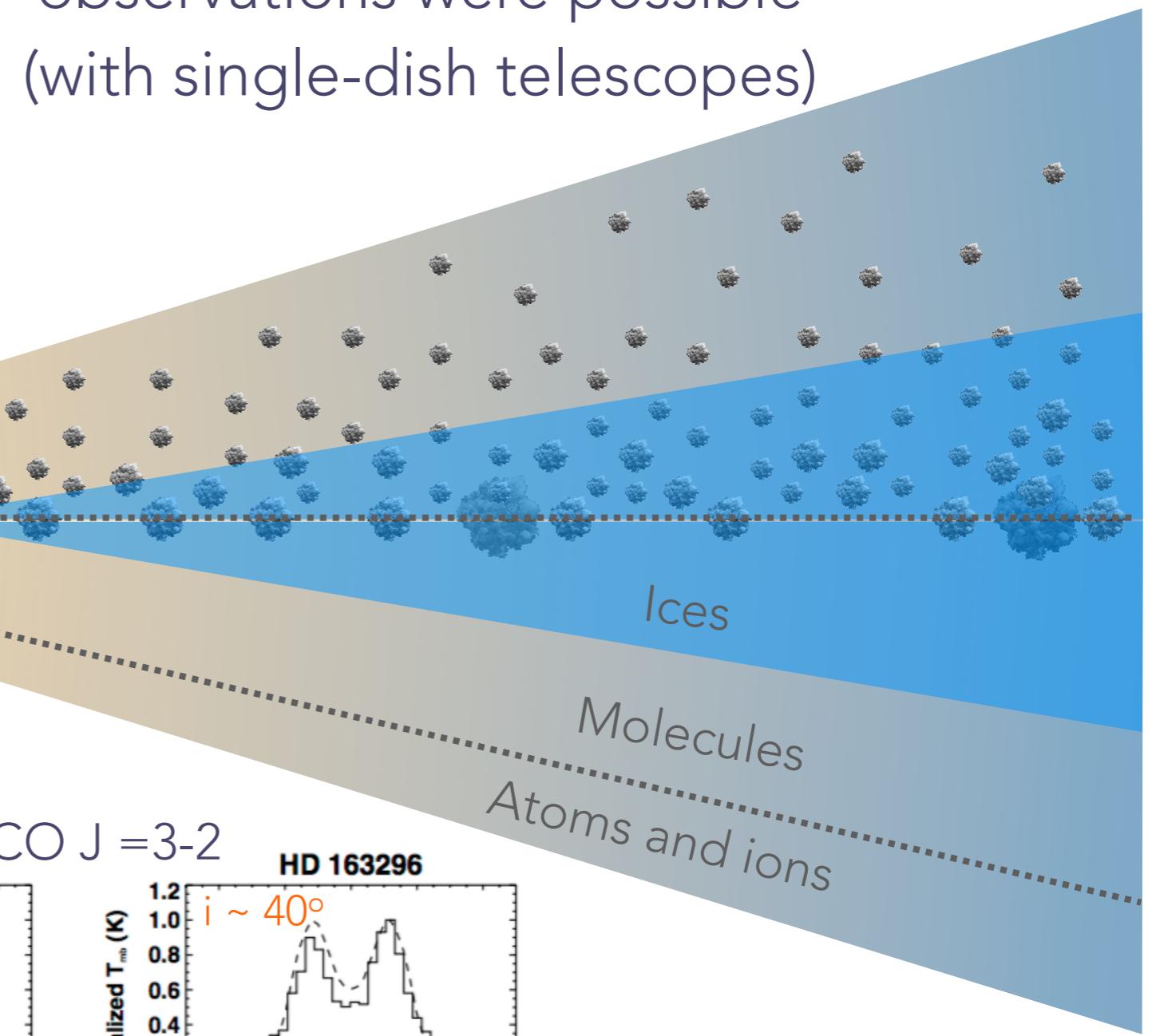


Dust  
spectral  
energy  
distribution

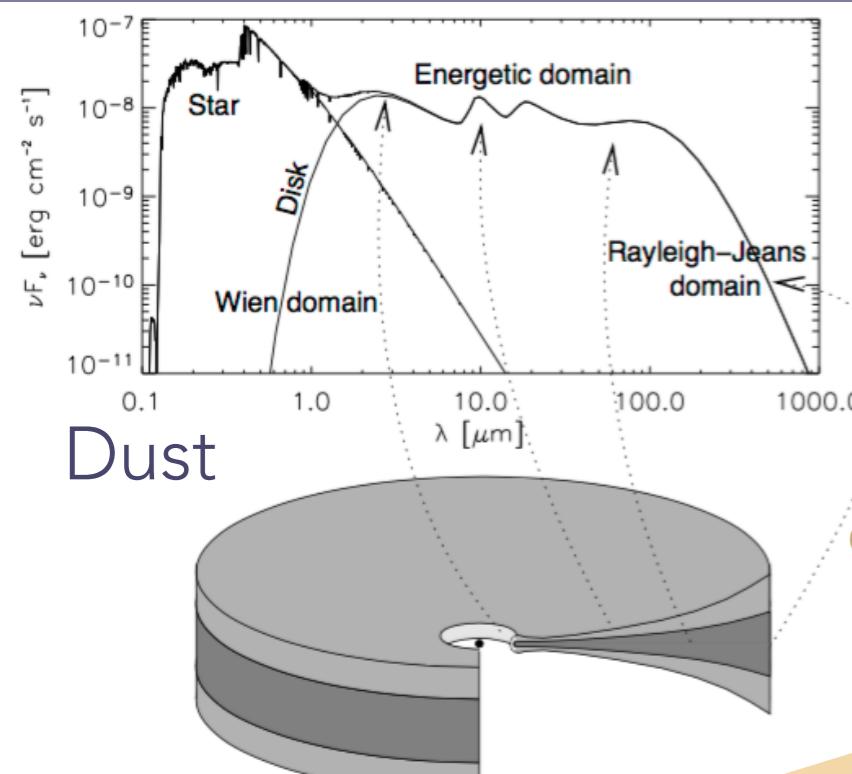


Molecular/atomic  
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Until recently: only spatially-unresolved  
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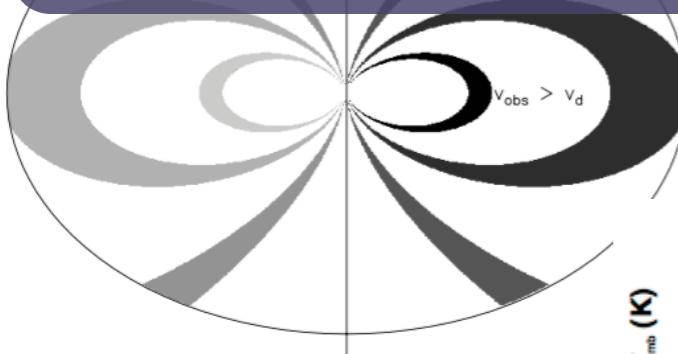


# Observations of protoplanetary disks

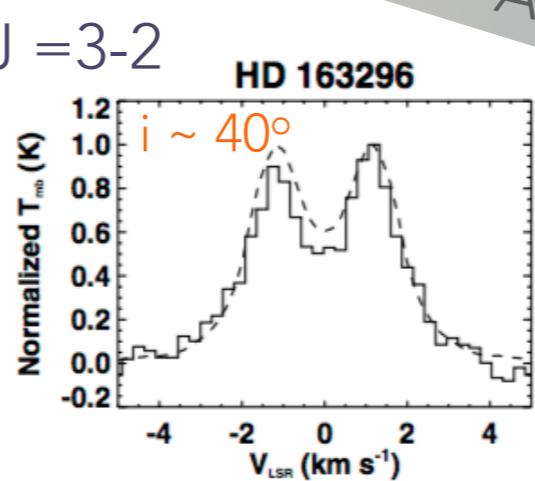
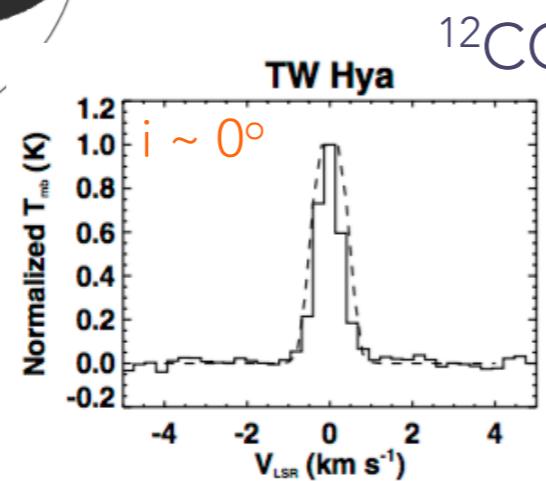


# Gas

What about the chemistry? So far H<sub>2</sub>, HD, and CO only mentioned! Useful probes of disk structure only.



# Molecular/atomic emission lines

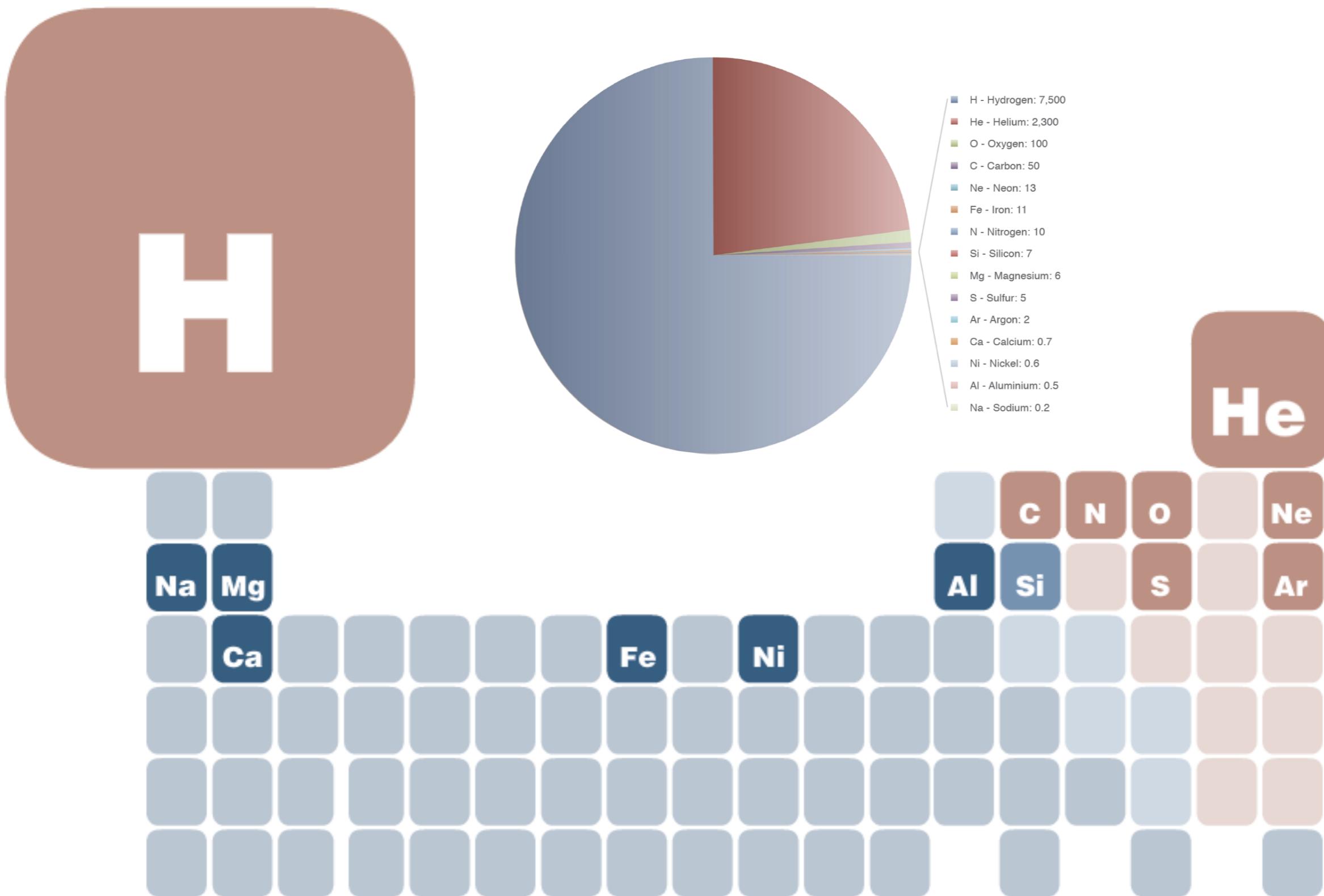


Dust  
spectral  
energy  
distribution

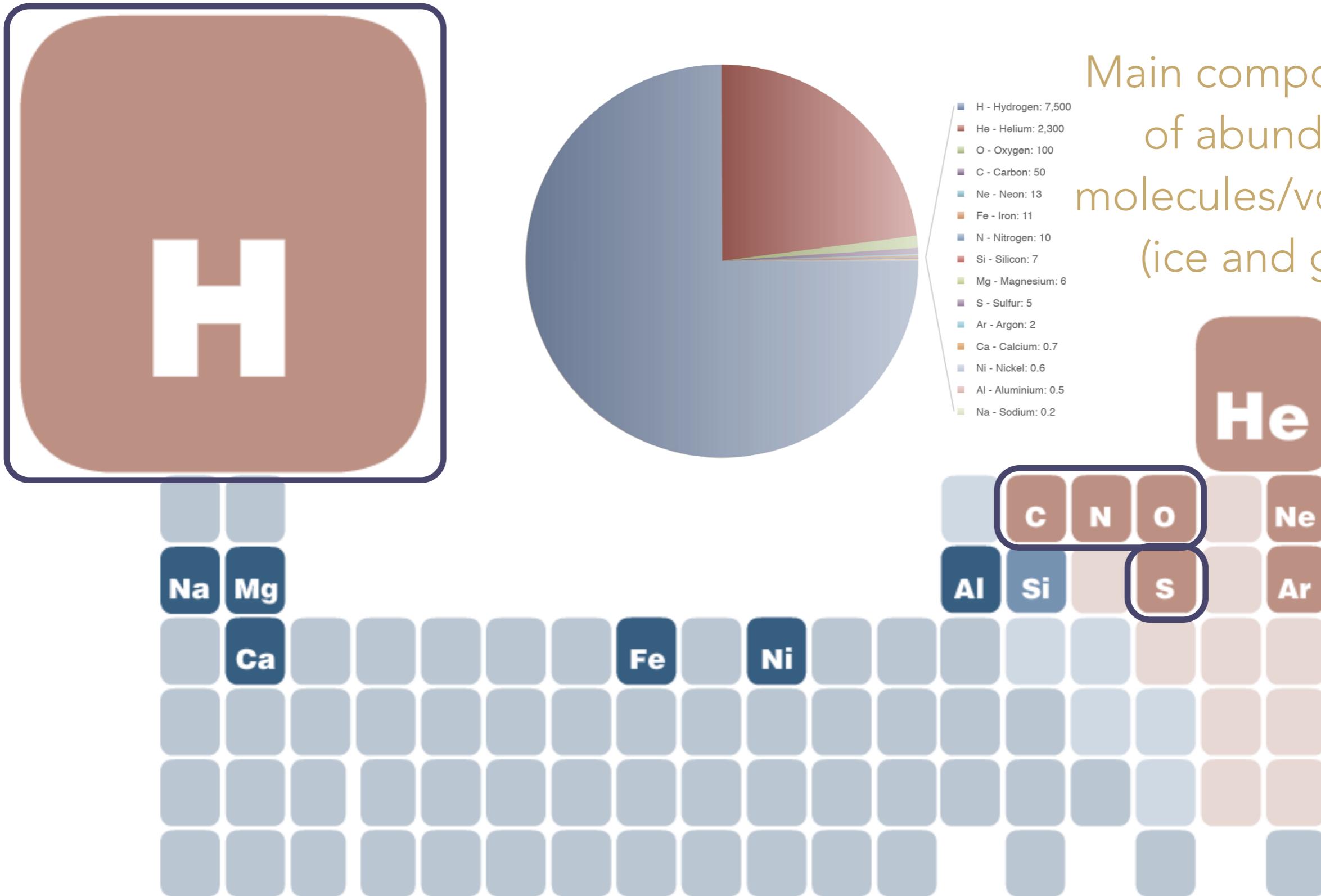
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# Molecules Atoms and ions

# The astronomers' periodic table



# The astronomers' periodic table



# Molecules in space

2 atoms		3 atoms		4 atoms	5 atoms	6 atoms	7 atoms	8 atoms	9 atoms	≥ 10 atoms
H <sub>2</sub>	HD	C <sub>3</sub>	AlNC	c-C <sub>3</sub> H	C <sub>5</sub>	C <sub>5</sub> H	C <sub>6</sub> H	CH <sub>3</sub> C <sub>3</sub> N	CH <sub>3</sub> C <sub>4</sub> H	CH <sub>3</sub> C <sub>5</sub> N
AlF	FeO ?	C <sub>2</sub> H	SiNC	I-C <sub>3</sub> H	C <sub>4</sub> H	I-H <sub>2</sub> C <sub>4</sub>	CH <sub>2</sub> CHCN	HCOOCH <sub>3</sub>	CH <sub>3</sub> CH <sub>2</sub> CN	(CH <sub>3</sub> ) <sub>2</sub> CO
AlCl	O <sub>2</sub>	C <sub>2</sub> O	HCP	C <sub>3</sub> N	C <sub>4</sub> Si	C <sub>2</sub> H <sub>4</sub>	CH <sub>3</sub> C <sub>2</sub> H	CH <sub>3</sub> COOH	(CH <sub>3</sub> ) <sub>2</sub> O	(CH <sub>2</sub> OH) <sub>2</sub>
C <sub>2</sub>	CF <sup>+</sup>	C <sub>2</sub> S	CCP	C <sub>3</sub> O	I-C <sub>3</sub> H <sub>2</sub>	CH <sub>3</sub> CN	HC <sub>5</sub> N	C <sub>7</sub> H	CH <sub>3</sub> CH <sub>2</sub> OH	CH <sub>3</sub> CH <sub>2</sub> CHO
CH	SiH ?	CH <sub>2</sub>	AlOH	C <sub>3</sub> S	c-C <sub>3</sub> H <sub>2</sub>	CH <sub>3</sub> NC	CH <sub>3</sub> CHO	C <sub>6</sub> H <sub>2</sub>	HC <sub>7</sub> N	HC <sub>9</sub> N
CH <sup>+</sup>	PO	HCN	H <sub>2</sub> O <sup>+</sup>	C <sub>2</sub> H <sub>2</sub>	H <sub>2</sub> CCN	CH <sub>3</sub> OH	CH <sub>3</sub> NH <sub>2</sub>	CH <sub>2</sub> OHCHO	C <sub>8</sub> H	CH <sub>3</sub> C <sub>6</sub> H
CN	AlO	HCO	H <sub>2</sub> Cl <sup>+</sup>	NH <sub>3</sub>	CH <sub>4</sub>	CH <sub>3</sub> SH	c-C <sub>2</sub> H <sub>4</sub> O	I-HC <sub>6</sub> H	CH <sub>3</sub> CONH <sub>2</sub>	C <sub>2</sub> H <sub>5</sub> OCHO
CO	OH <sup>+</sup>	HCO <sup>+</sup>	KCN	HCCN	HC <sub>3</sub> N	HC <sub>3</sub> NH <sup>+</sup>	H <sub>2</sub> CCHOH	CH <sub>2</sub> CHCHO ?	C <sub>8</sub> H <sup>-</sup>	CH <sub>3</sub> OCOCH <sub>3</sub>
CO <sup>+</sup>	CN <sup>-</sup>	HCS <sup>+</sup>	FeCN	HCNH <sup>+</sup>	HC <sub>2</sub> NC	HC <sub>2</sub> CHO	C <sub>6</sub> H <sup>-</sup>	CH <sub>2</sub> CCHCN	C <sub>3</sub> H <sub>6</sub>	c-C <sub>6</sub> H <sub>6</sub>
CP	SH <sup>+</sup>	HOC <sup>+</sup>	O <sub>2</sub> H	HNCO	HCOOH	NH <sub>2</sub> CHO		H <sub>2</sub> NCH <sub>2</sub> CN	CH <sub>3</sub> CH <sub>2</sub> SH	n-C <sub>3</sub> H <sub>7</sub> CN
SiC	SH	H <sub>2</sub> O	TiO <sub>2</sub>	HNCS	H <sub>2</sub> CNH	C <sub>5</sub> N		CH <sub>3</sub> CHNH		i-C <sub>3</sub> H <sub>7</sub> CN
HCl	HCl <sup>+</sup>	H <sub>2</sub> S	C <sub>2</sub> N	HOCO <sup>+</sup>	H <sub>2</sub> C <sub>2</sub> O	I-HC <sub>4</sub> H				HC <sub>11</sub> N
KCl	TiO	HNC	Si <sub>2</sub> C	H <sub>2</sub> CO	H <sub>2</sub> N CN	I-HC <sub>4</sub> N				C <sub>60</sub>
NH	ArH <sup>+</sup>	HNO		H <sub>2</sub> CN	HNC <sub>3</sub>	c-H <sub>2</sub> C <sub>3</sub> O				C <sub>70</sub>
NO	NO <sup>+</sup> ?	MgCN		H <sub>2</sub> CS	SiH <sub>4</sub>	H <sub>2</sub> CCNH ?				
NS		MgNC		H <sub>3</sub> O <sup>+</sup>	H <sub>2</sub> COH <sup>+</sup>	C <sub>5</sub> N <sup>-</sup>	✳ Cations (positively-charged)			
NaCl		N <sub>2</sub> H <sup>+</sup>		c-SiC <sub>3</sub>	C <sub>4</sub> H <sup>-</sup>	HNCHCN	✳ Anions (negatively-charged)			
OH		N <sub>2</sub> O		CH <sub>3</sub>	HCOCN		✳ Radicals (unpaired electrons)			
PN		NaCN		C <sub>3</sub> N <sup>-</sup>	HNCNH		✳ Unsaturated carbon chains			
SO		OCS		PH <sub>3</sub>	CH <sub>3</sub> O		✳ Structural isomers			
SO <sup>+</sup>		SO <sub>2</sub>		HCNO	NH <sub>4</sub> <sup>+</sup>		✳ Complex organic molecules			
SiN		c-SiC <sub>2</sub>		HSCN	H <sub>2</sub> NCO <sup>+</sup> ?		✳ Many isotolologues			
SiO		CO <sub>2</sub>		H <sub>2</sub> O <sub>2</sub>	HCCNH <sup>+</sup>		✳ > 180 and counting ...			
SiS		NH <sub>2</sub>		C <sub>3</sub> H <sup>+</sup>						
CS		H <sub>3</sub> <sup>+</sup>		HMgNC						
HF		SiCN		HCCO						

# Molecules in space

2 atoms	3 atoms	4 atoms	5 atoms	6 atoms	7 atoms	8 atoms	9 atoms	≥ 10 atoms
H <sub>2</sub>	HD	C <sub>3</sub>	AlNC	c-C <sub>3</sub> H	C <sub>5</sub>	C <sub>5</sub> H	C <sub>6</sub> H	CH <sub>3</sub> C <sub>3</sub> N
AlF	FeO ?	C <sub>2</sub> H	SiNC	I-C <sub>3</sub> H	C <sub>4</sub> H	I-H <sub>2</sub> C <sub>4</sub>	CH <sub>2</sub> CHCN	HCOOCH <sub>3</sub>
AlCl	O <sub>2</sub>	C <sub>2</sub> O	HCP	C <sub>3</sub> N	C <sub>4</sub> Si	C <sub>2</sub> H <sub>4</sub>	CH <sub>3</sub> C <sub>2</sub> H	CH <sub>3</sub> COOH
C <sub>2</sub>	CF <sup>+</sup>	C <sub>2</sub> S	CCP	C <sub>3</sub> O	I-C <sub>3</sub> H <sub>2</sub>	CH <sub>3</sub> CN	HC <sub>5</sub> N	C <sub>7</sub> H
CH	SiH ?	CH <sub>2</sub>	AlOH	C <sub>3</sub> S	c-C <sub>3</sub> H <sub>2</sub>	CH <sub>3</sub> NC	CH <sub>3</sub> CHO	C <sub>6</sub> H <sub>2</sub>
CH <sup>+</sup>	PO	HCN	H <sub>2</sub> O <sup>+</sup>	C <sub>2</sub> H <sub>2</sub>	H <sub>2</sub> CCN	CH <sub>3</sub> OH	CH <sub>2</sub> OHCHO	C <sub>8</sub> H
CN	AlO	HCO	H <sub>2</sub> Cl <sup>+</sup>	NH <sub>3</sub>	CH <sub>4</sub>	CH <sub>3</sub> SH	c-C <sub>2</sub> H <sub>4</sub> O	I-CH <sub>6</sub> H
CO	OH <sup>+</sup>	HCO <sup>+</sup>	KCN	HCCN	HC <sub>3</sub> N	HC <sub>3</sub> NH <sup>+</sup>	H <sub>2</sub> CCHOH	CH <sub>2</sub> CHCHO ?
CO <sup>+</sup>	CN <sup>-</sup>	HCS <sup>+</sup>	FeCN	HCNH <sup>+</sup>	HC <sub>2</sub> NC	HC <sub>2</sub> CHO	C <sub>6</sub> H <sup>-</sup>	CH <sub>2</sub> CCHCN
CP	SH <sup>+</sup>	HOC <sup>+</sup>	O <sub>2</sub> H	HNCO	HCOOH	NH <sub>2</sub> CHO		C <sub>3</sub> H <sub>6</sub>
SiC	SH	H <sub>2</sub> O	TiO <sub>2</sub>	HNCS	H <sub>2</sub> CNH	C <sub>5</sub> N		CH <sub>3</sub> CHNH
HCl	HCl <sup>+</sup>	H <sub>2</sub> S	C <sub>2</sub> N	HO <sub>2</sub> C	H <sub>2</sub> C <sub>2</sub> O	I-CH <sub>4</sub> H		HC <sub>11</sub> N
KCl	TiO	HNC	Si <sub>2</sub> C	H <sub>2</sub> CO	H <sub>2</sub> NCN	I-CH <sub>4</sub> N		C <sub>60</sub>
NH	ArH <sup>+</sup>	HNO		H <sub>2</sub> CN	HNC <sub>3</sub>	c-H <sub>2</sub> C <sub>3</sub> O		C <sub>70</sub>
NO	NO <sup>+</sup> ?	MgCN		H <sub>2</sub> CS	SiH <sub>4</sub>	H <sub>2</sub> CCNH ?		
NS		MgNC		H <sub>3</sub> O <sup>+</sup>	H <sub>2</sub> COH <sup>+</sup>	C <sub>5</sub> N <sup>-</sup>		
NaCl		N <sub>2</sub> H <sup>+</sup>		c-SiC <sub>3</sub>	C <sub>4</sub> H <sup>-</sup>	HNCHCN		
OH		N <sub>2</sub> O		CH <sub>3</sub>	HCOCN			
PN		NaCN		C <sub>3</sub> N <sup>-</sup>	HNCNH			
SO		OCS		PH <sub>3</sub>	CH <sub>3</sub> O			
SO <sup>+</sup>		SO <sub>2</sub>		HCNO	NH <sub>4</sub> <sup>+</sup>			
SiN		c-SiC <sub>2</sub>		HSCN	H <sub>2</sub> NCO <sup>+</sup> ?			
SiO		CO <sub>2</sub>		H <sub>2</sub> O <sub>2</sub>	HCCNH <sup>+</sup>			
SiS		NH <sub>2</sub>		C <sub>3</sub> H <sup>+</sup>				
CS		H <sub>3</sub> <sup>+</sup>		HMgNC				
HF		SiCN		HCCO				

# Molecules in space

2 atoms	3 atoms	4 atoms	5 atoms	6 atoms	7 atoms	8 atoms	9 atoms	≥ 10 atoms		
H <sub>2</sub>	HD	C <sub>3</sub>	AlNC	c-C <sub>3</sub> H	C <sub>5</sub>	C <sub>5</sub> H	CH <sub>3</sub> C <sub>3</sub> N	CH <sub>3</sub> C <sub>4</sub> H	CH <sub>3</sub> C <sub>5</sub> N	
AlF	FeO ?	C <sub>2</sub> H	SiNC	I-C <sub>3</sub> H	C <sub>4</sub> H	I-H <sub>2</sub> C <sub>4</sub>	CH <sub>2</sub> CHCN	HCOOCH <sub>3</sub>	CH <sub>3</sub> CH <sub>2</sub> CN	(CH <sub>3</sub> ) <sub>2</sub> CO
AlCl	O <sub>2</sub>	C <sub>2</sub> O	HCP	C <sub>3</sub> N	C <sub>4</sub> Si	C <sub>2</sub> H <sub>4</sub>	CH <sub>3</sub> C <sub>2</sub> H	CH <sub>3</sub> COOH	(CH <sub>3</sub> ) <sub>2</sub> O	(CH <sub>2</sub> OH) <sub>2</sub>
C <sub>2</sub>	CF <sup>+</sup>	C <sub>2</sub> S	CCP	C <sub>3</sub> O	I-C <sub>3</sub> H <sub>2</sub>	CH <sub>3</sub> CN	HC <sub>5</sub> N	C <sub>7</sub> H	CH <sub>3</sub> CH <sub>2</sub> OH	CH <sub>3</sub> CH <sub>2</sub> CHO
CH	SiH ?	CH <sub>2</sub>	AlOH	C <sub>3</sub> S	c-C <sub>3</sub> H <sub>2</sub>	CH <sub>3</sub> NC	CH <sub>3</sub> CHO	C <sub>6</sub> H <sub>2</sub>	HC <sub>7</sub> N	HC <sub>9</sub> N
CH <sup>+</sup>	PO	HCN	H <sub>2</sub> O <sup>+</sup>	C <sub>2</sub> H <sub>2</sub>	H <sub>2</sub> CCN	CH <sub>3</sub> OH	CH <sub>3</sub> NH <sub>2</sub>	CH <sub>2</sub> OHCHO	C <sub>8</sub> H	CH <sub>3</sub> C <sub>6</sub> H
CN	AlO	HCO	H <sub>2</sub> Cl <sup>+</sup>	NH <sub>3</sub>	CH <sub>4</sub>	CH <sub>3</sub> SH	c-C <sub>2</sub> H <sub>4</sub> O	I-HC <sub>6</sub> H	CH <sub>3</sub> CONH <sub>2</sub>	C <sub>2</sub> H <sub>5</sub> OCHO
CO	OH <sup>+</sup>	HCO <sup>+</sup>	KCN	HCCN	HC <sub>3</sub> N	HC <sub>3</sub> NH <sup>+</sup>	H <sub>2</sub> CCHOH	CH <sub>2</sub> CHCHO ?	C <sub>8</sub> H <sup>-</sup>	CH <sub>3</sub> OCOCH <sub>3</sub>
CO <sup>+</sup>	CN <sup>-</sup>	HCS <sup>+</sup>	FeCN	HCNH <sup>+</sup>	HC <sub>2</sub> NC	HC <sub>2</sub> CHO	C <sub>6</sub> H <sup>-</sup>	CH <sub>2</sub> CCHCN	C <sub>3</sub> H <sub>6</sub>	c-C <sub>6</sub> H <sub>6</sub>
CP	SH <sup>+</sup>	HOC <sup>+</sup>	O <sub>2</sub> H	HNCO	HCOOH	NH <sub>2</sub> CHO		H <sub>2</sub> NCH <sub>2</sub> CN	CH <sub>3</sub> CH <sub>2</sub> SH	n-C <sub>3</sub> H <sub>7</sub> CN
SiC	SH	H <sub>2</sub> O	TiO <sub>2</sub>	HNCS	H <sub>2</sub> CNH	C <sub>5</sub> N		CH <sub>3</sub> CHNH		i-C <sub>3</sub> H <sub>7</sub> CN
HCl	HCl <sup>+</sup>	H <sub>2</sub> S	C <sub>2</sub> N	HO <sub>2</sub> C	H <sub>2</sub> C <sub>2</sub> O	I-HC <sub>4</sub> H				HC <sub>11</sub> N
KCl	TiO	HNC	Si <sub>2</sub> C	H <sub>2</sub> CO	H <sub>2</sub> NCN	I-HC <sub>4</sub> N				C <sub>60</sub>
NH	ArH <sup>+</sup>	HNO		H <sub>2</sub> CN	HNC <sub>3</sub>	c-H <sub>2</sub> C <sub>3</sub> O				C <sub>70</sub>
NO	NO <sup>+</sup> ?	MgCN		H <sub>2</sub> CS	SiH <sub>4</sub>	H <sub>2</sub> CCNH ?				
NS		MgNC		H <sub>3</sub> O <sup>+</sup>	H <sub>2</sub> COH <sup>+</sup>	C <sub>5</sub> N <sup>-</sup>				
NaCl		N <sub>2</sub> H <sup>+</sup>		c-SiC <sub>3</sub>	C <sub>4</sub> H <sup>-</sup>	HNCHCN				
OH		N <sub>2</sub> O		CH <sub>3</sub>	HCOCN					
PN		NaCN		C <sub>3</sub> N <sup>-</sup>	HNCNH					
SO		OCS		PH <sub>3</sub>	CH <sub>3</sub> O					
SO <sup>+</sup>		SO <sub>2</sub>		HCNO	NH <sub>4</sub> <sup>+</sup>					
SiN		c-SiC <sub>2</sub>		HSCN	H <sub>2</sub> NCO <sup>+</sup> ?					
SiO		CO <sub>2</sub>		H <sub>2</sub> O <sub>2</sub>	HCCNH <sup>+</sup>					
SiS		NH <sub>2</sub>		C <sub>3</sub> H <sup>+</sup>						
CS		H <sub>3</sub> <sup>+</sup>		HMgNC						
HF		SiCN		HCCO						

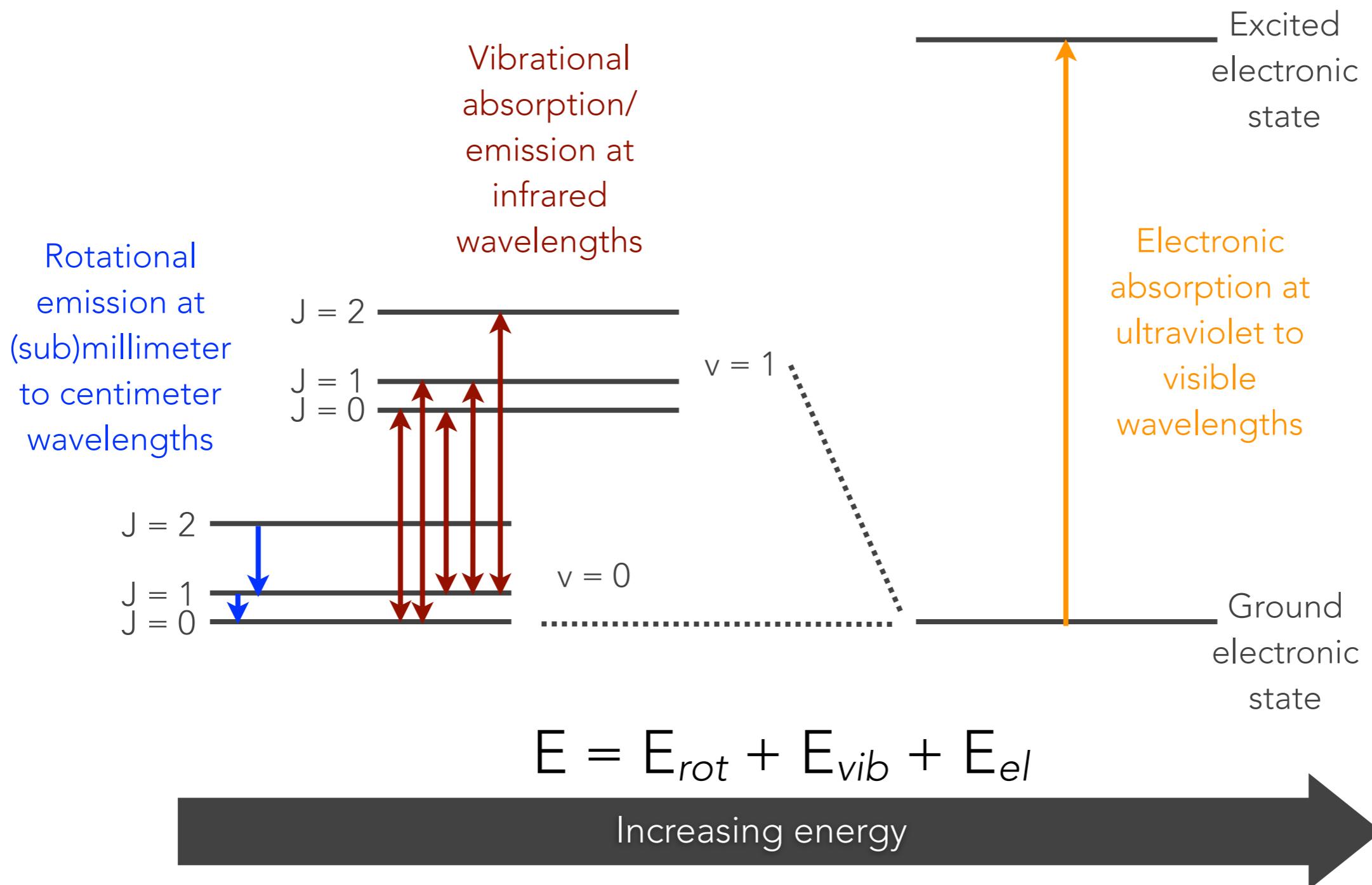
Protoplanetary disk  
molecules/volatiles?

21\* and counting ...

\* not including isotopologues

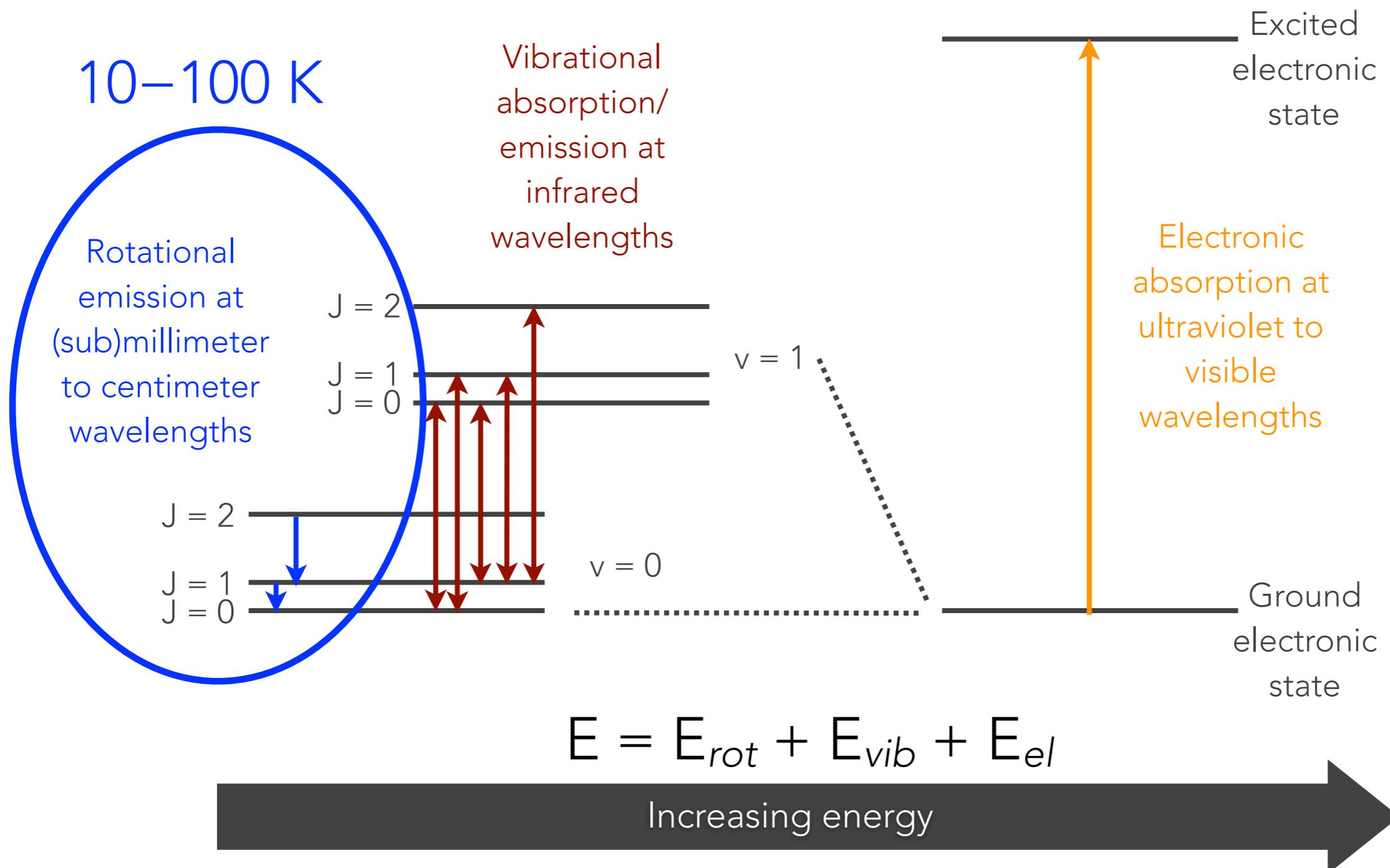
# Detection of molecules

## A crash course in molecular spectroscopy



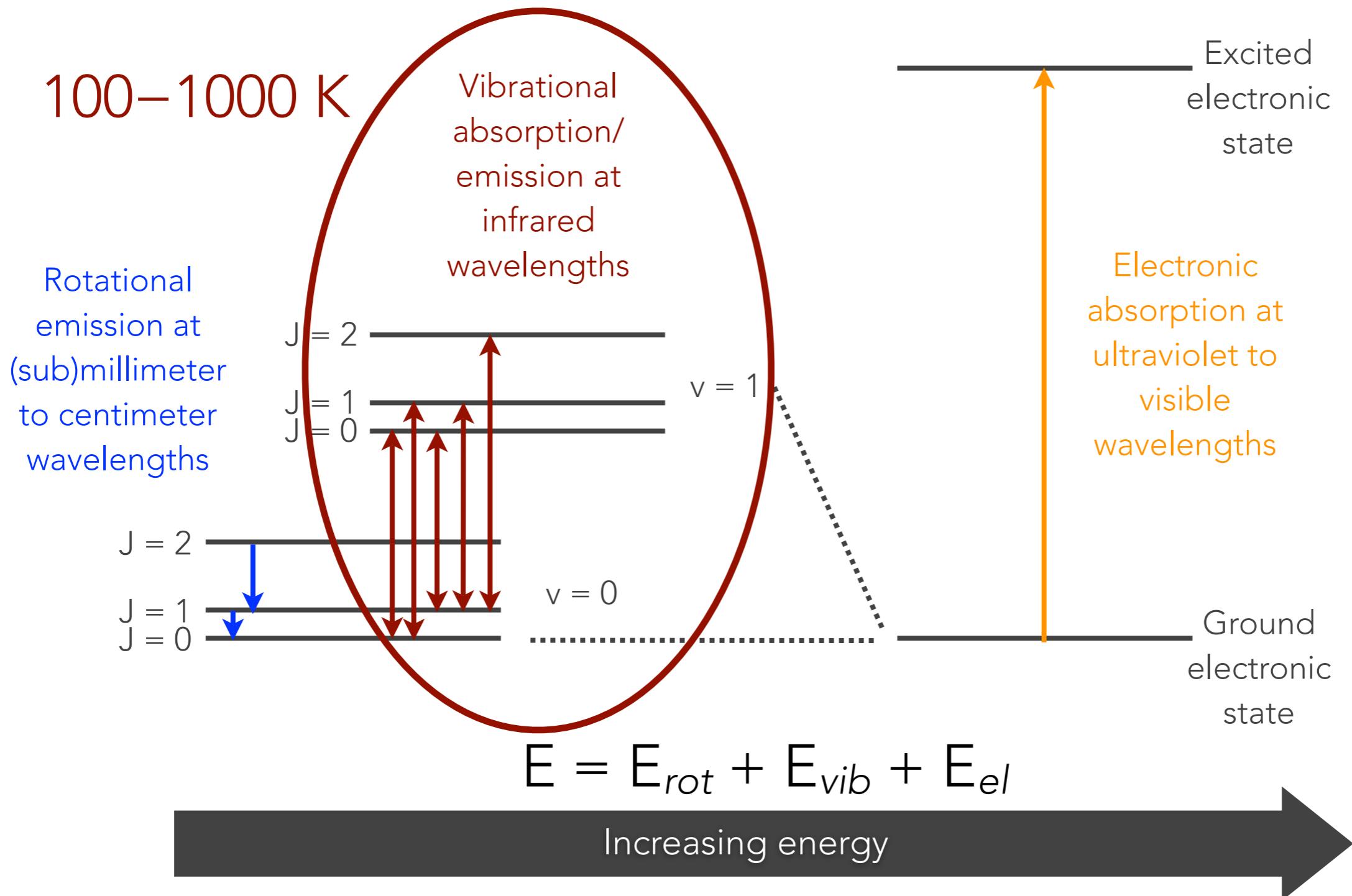
# Detection of molecules

## A crash course in molecular spectroscopy

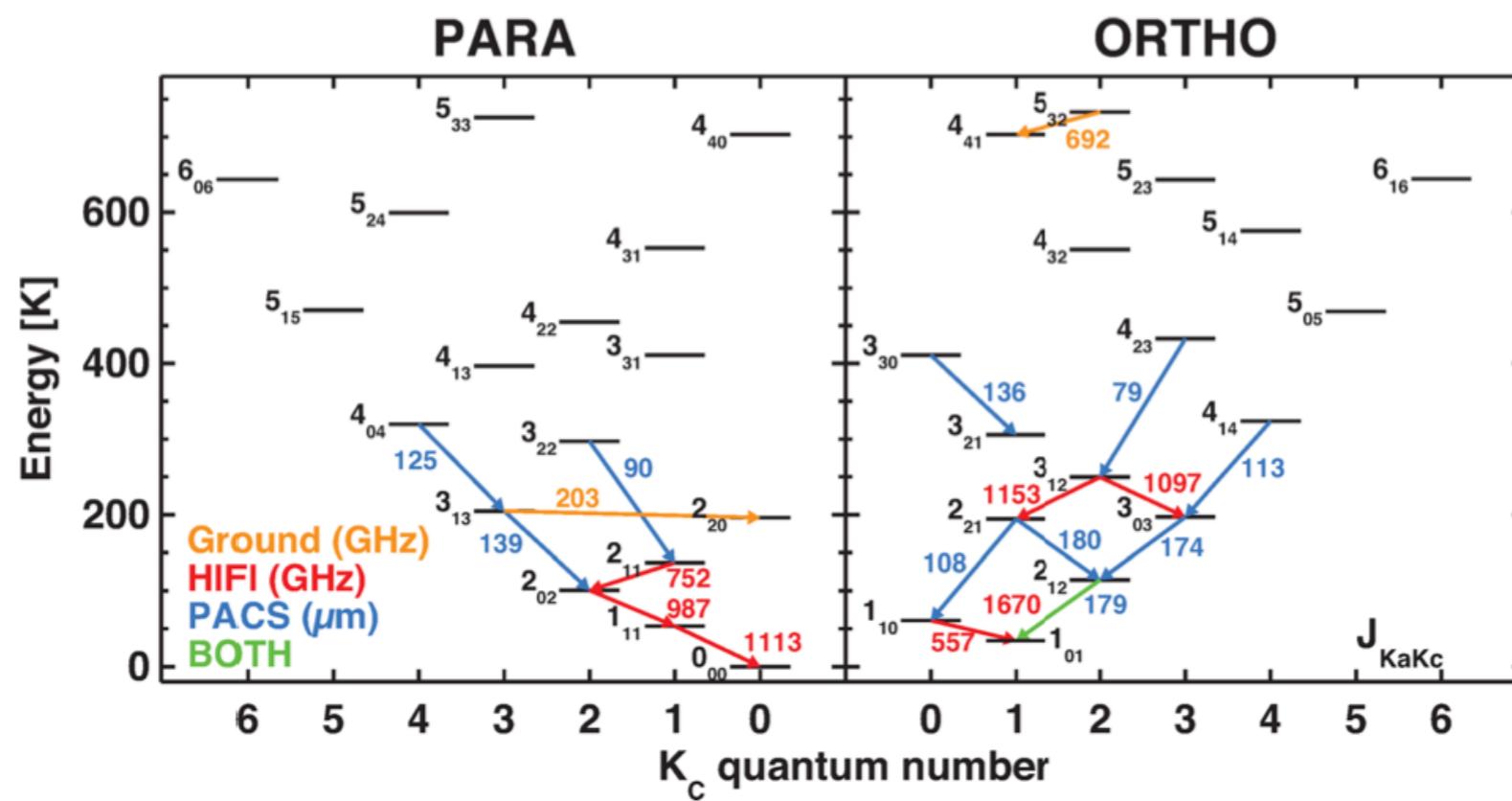
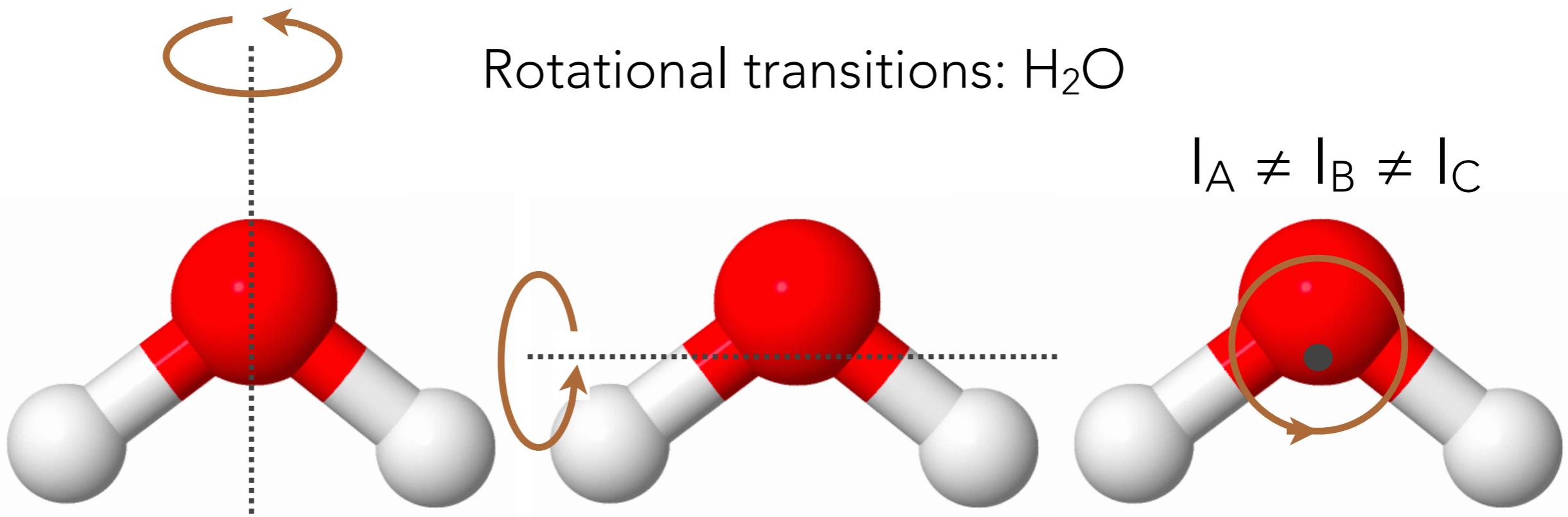


# Detection of molecules

## A crash course in molecular spectroscopy



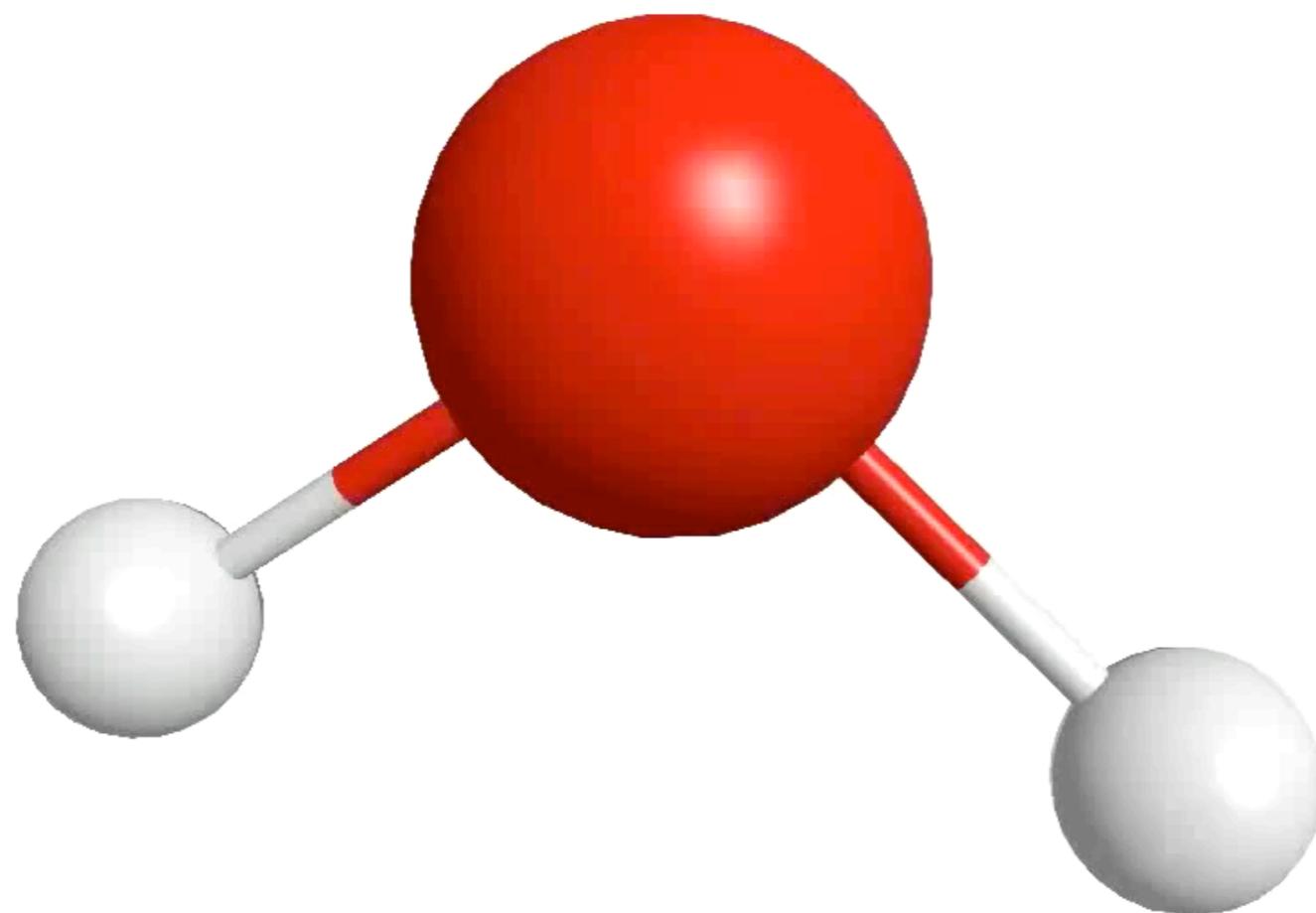
# Detection of molecules



Despite only consisting of 3 atoms,  $\text{H}_2\text{O}$  has a complex rotational spectrum

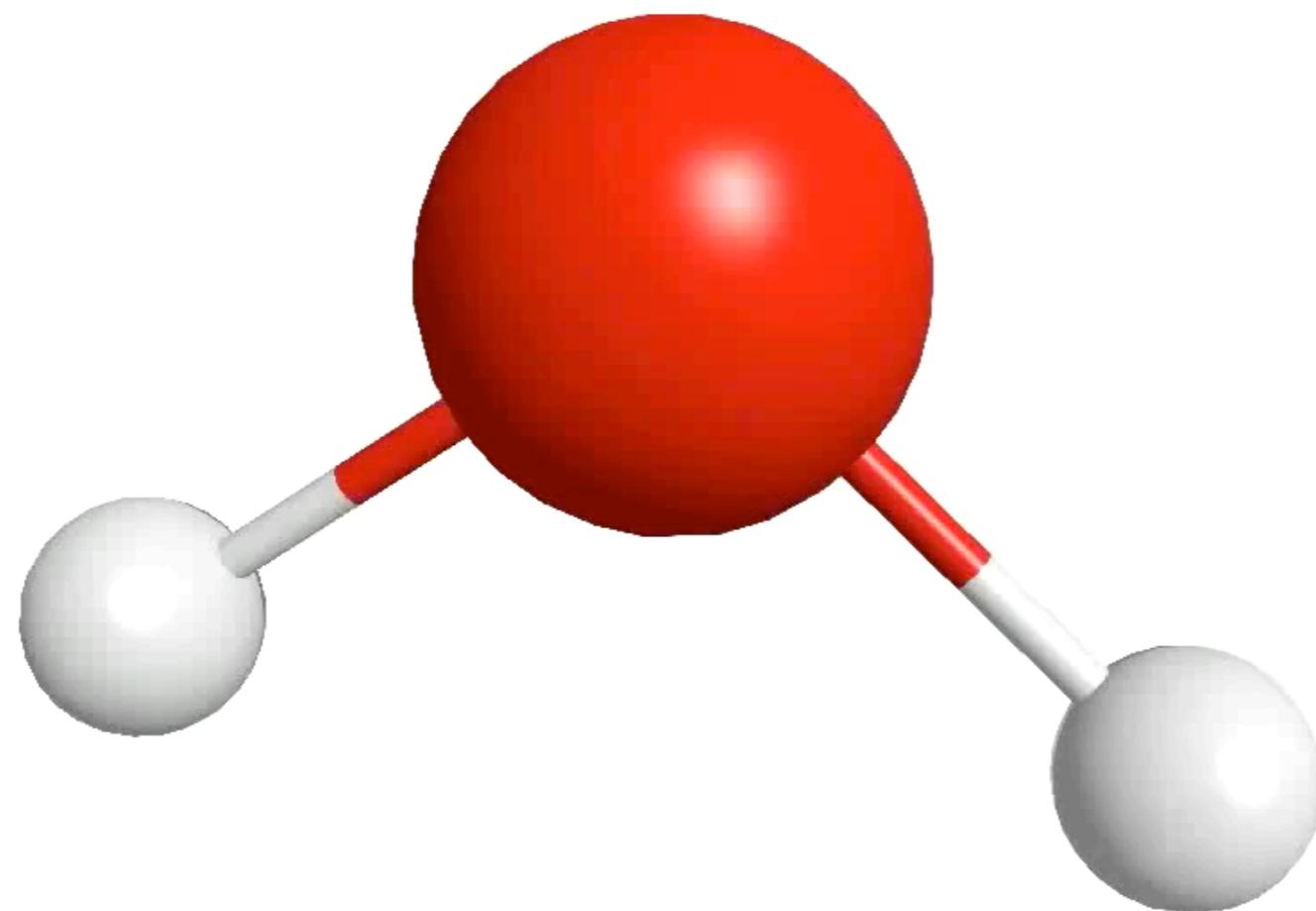
# Detection of molecules

Vibrational transitions: H<sub>2</sub>O



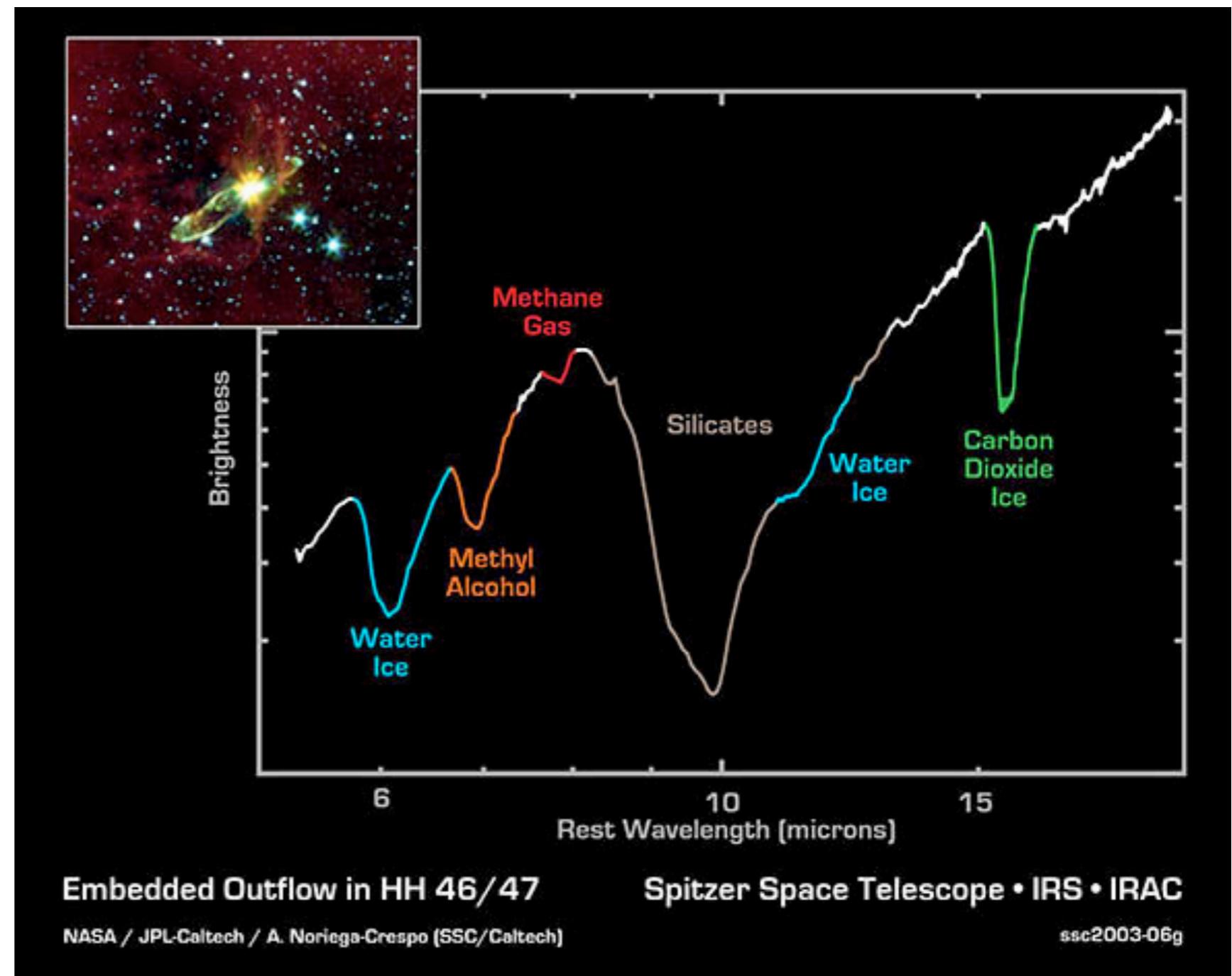
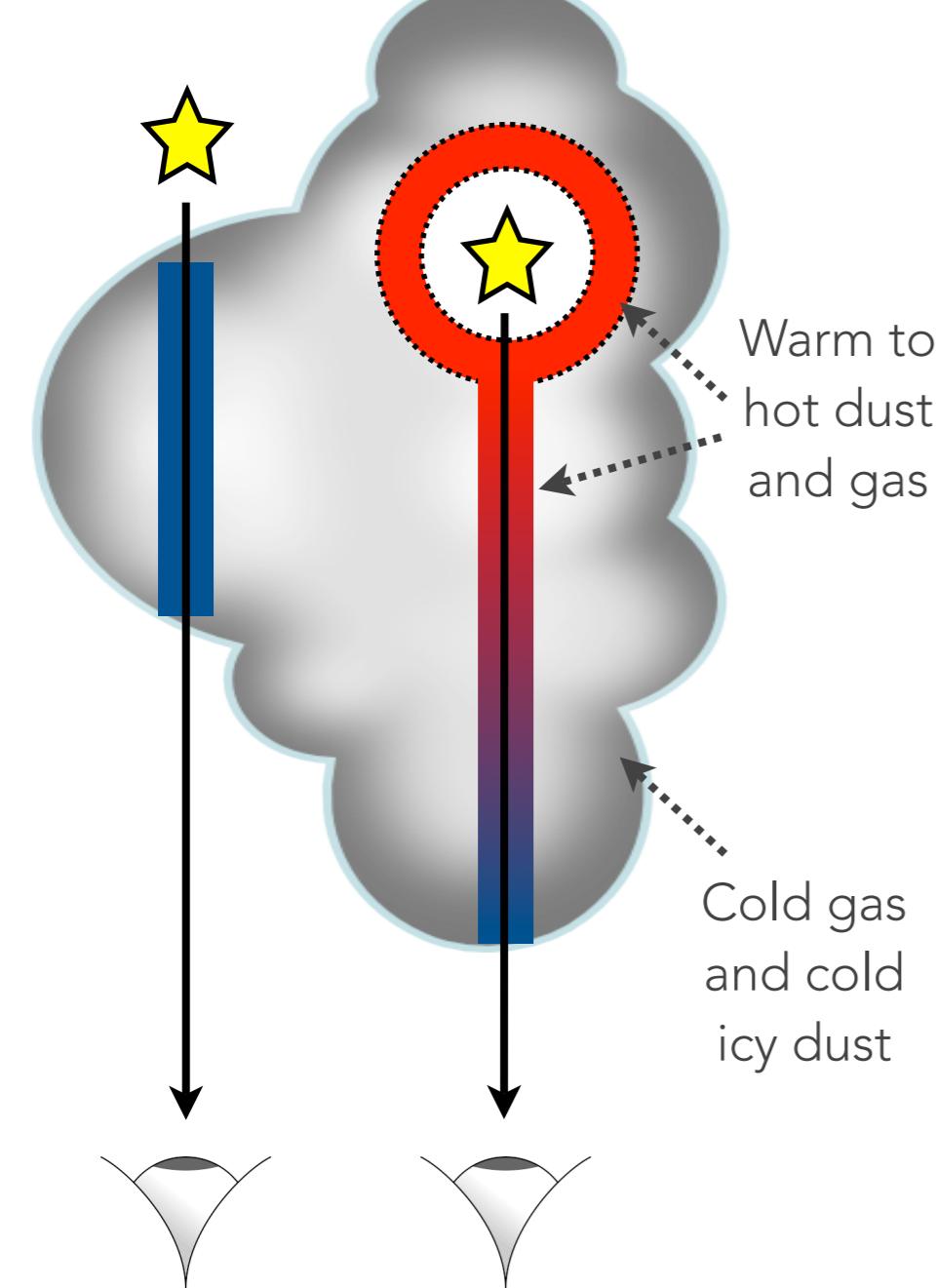
# Detection of molecules

Vibrational transitions: H<sub>2</sub>O



# Detection of molecules

## Infrared wavelengths: absorption

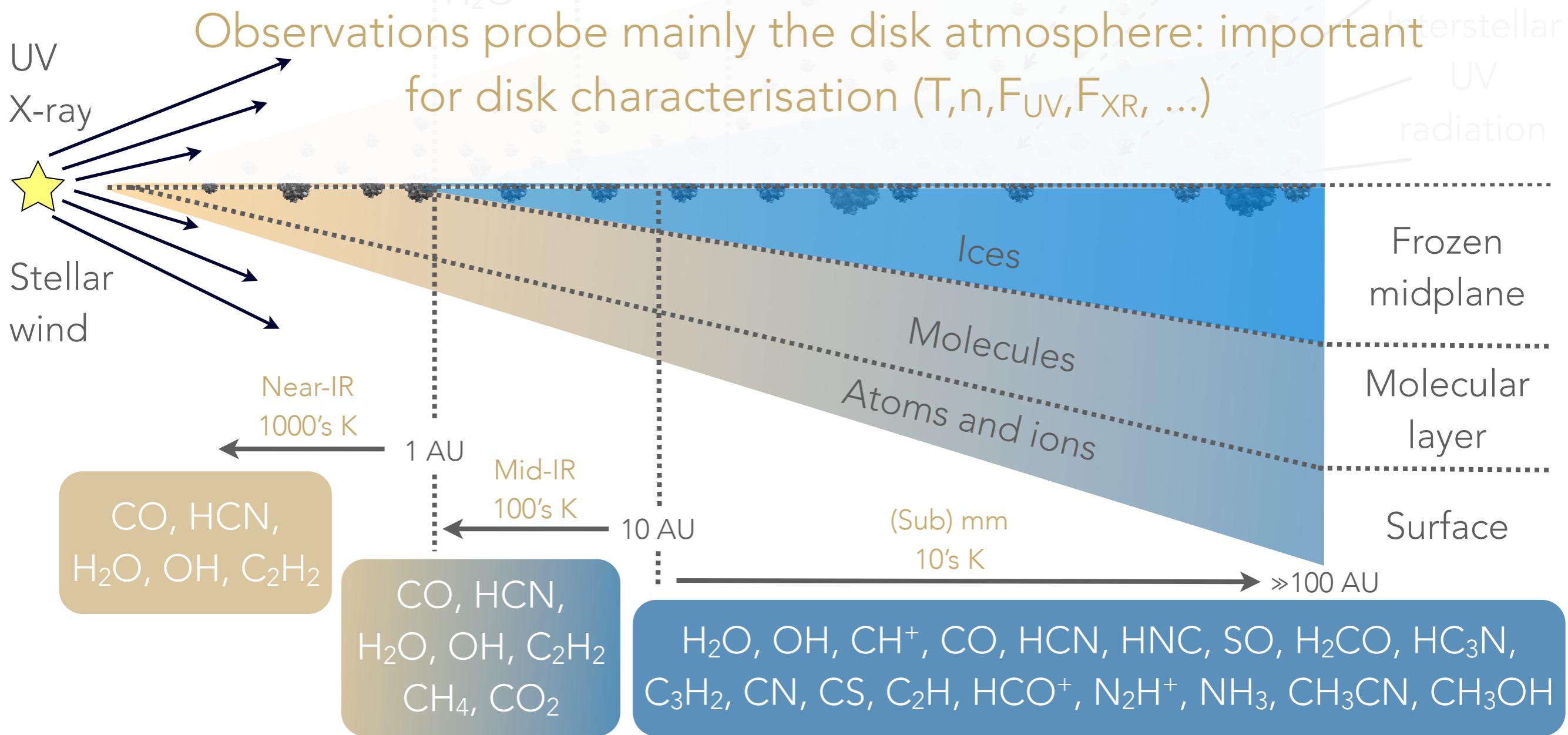


# Detection of molecules

(Sub)millimeter wavelengths: emission

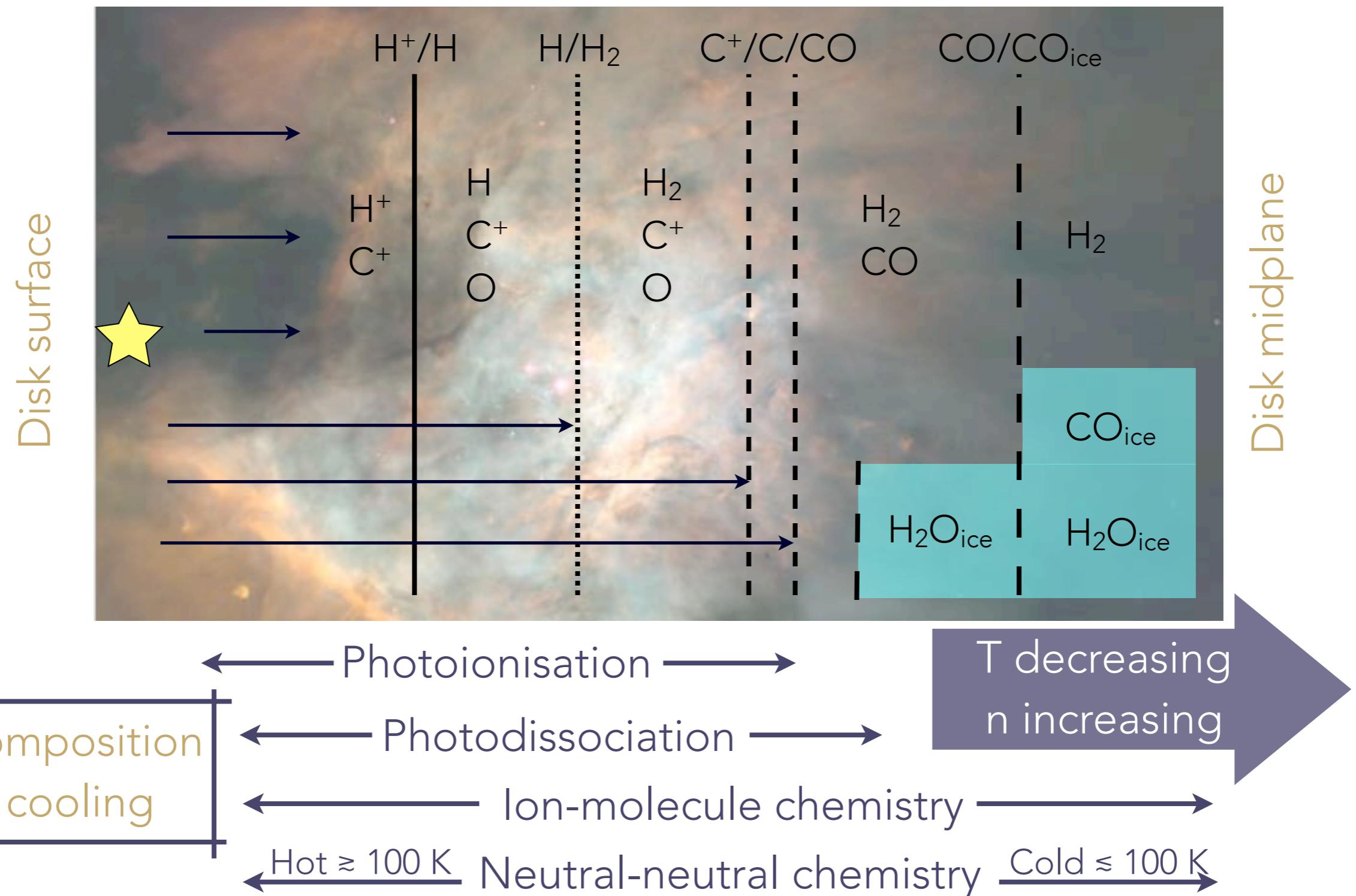
# Chemical anatomy of a protoplanetary disk

Two phases of molecules/volatiles in protoplanetary disks:  
gas ( $H_2$  dominated) and ice ( $H_2O$  dominated)



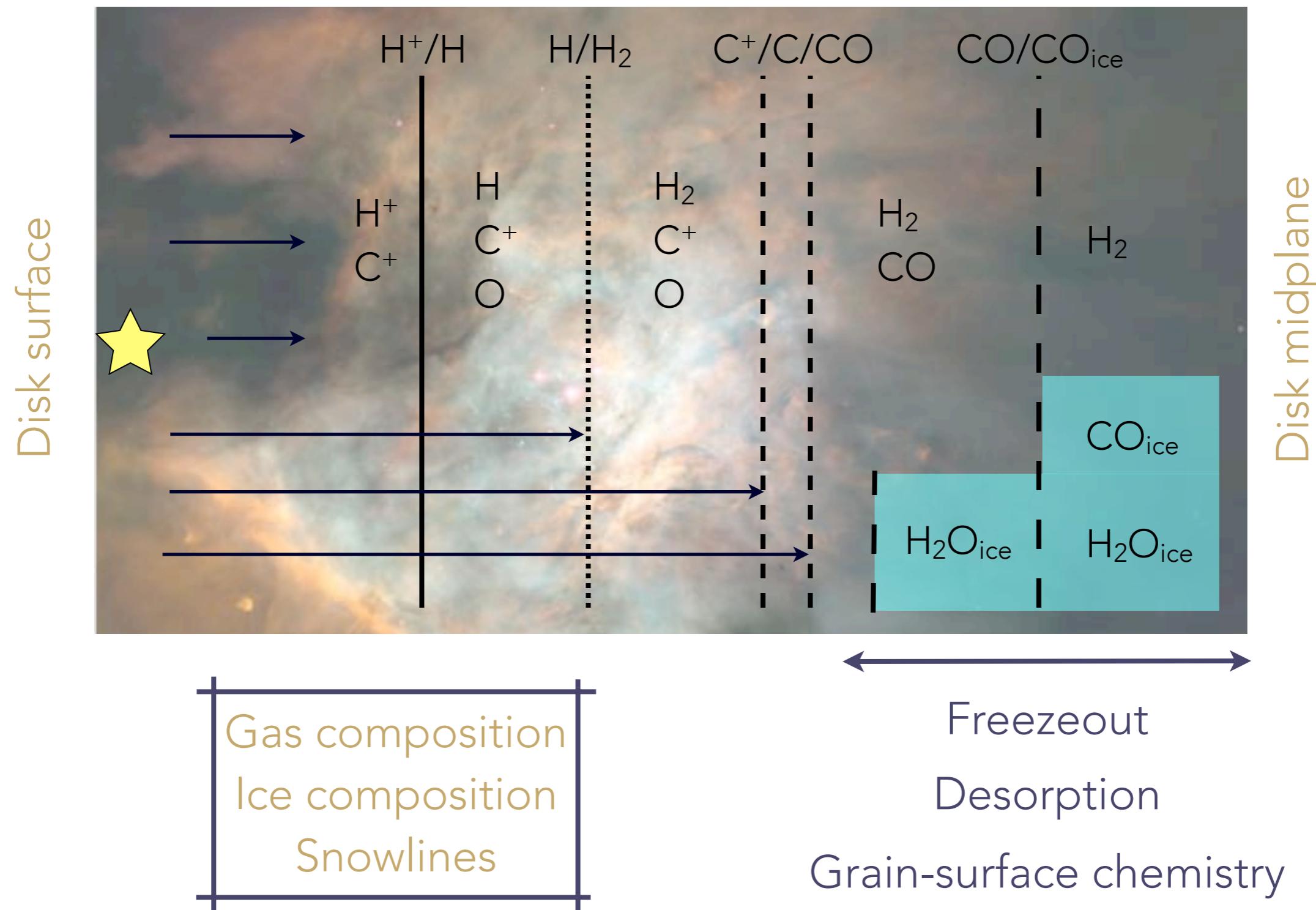
# What chemistry is important where and why?

Protoplanetary disks are essentially 2/3D photon-dominated regions (PDRs)



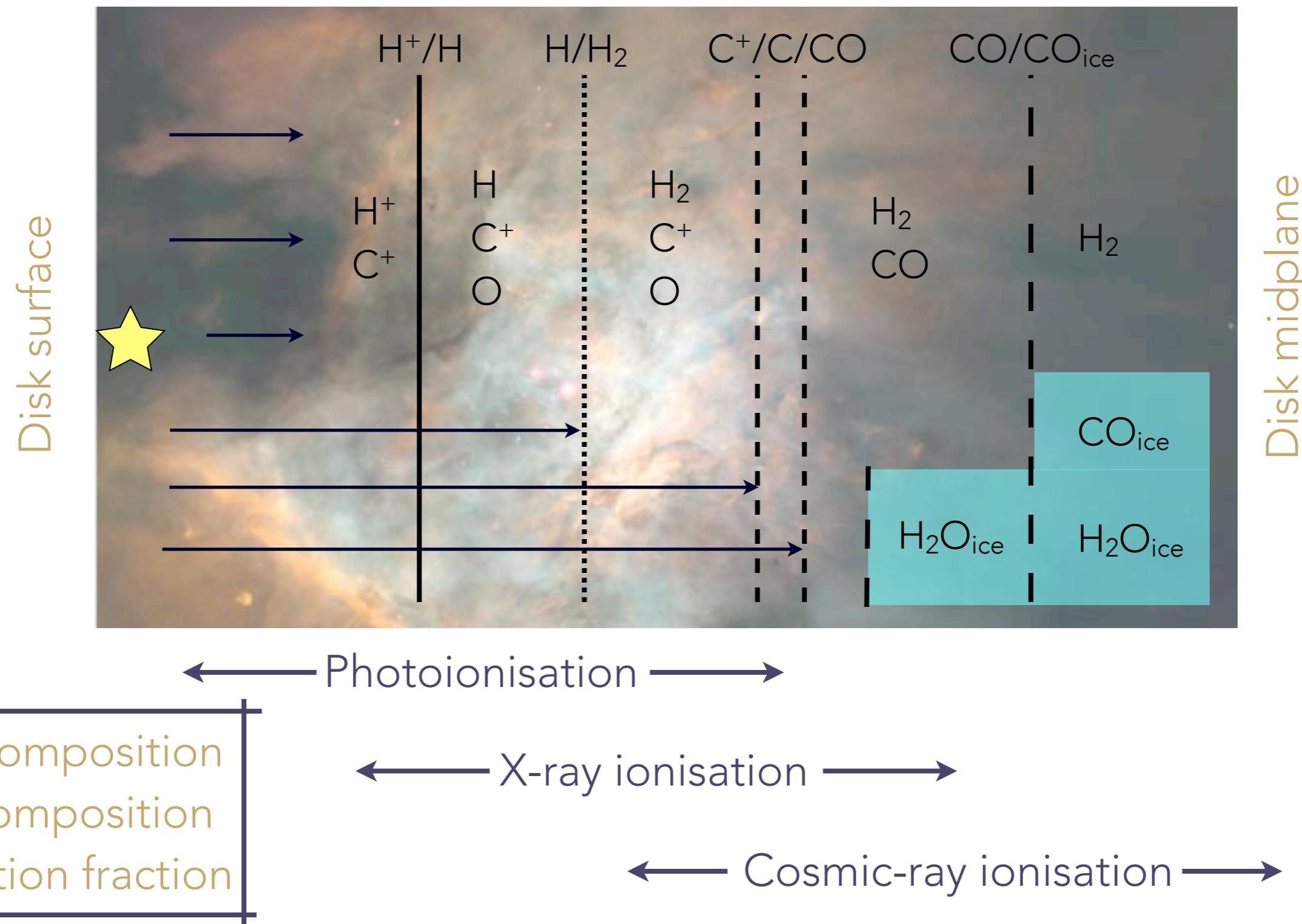
# What chemistry is important where and why?

Protoplanetary disks are essentially 2/3D photon-dominated regions (PDRs)



# What chemistry is important where and why?

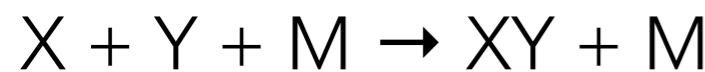
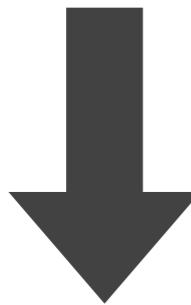
Protoplanetary disks are essentially 2/3D photon-dominated regions (PDRs)



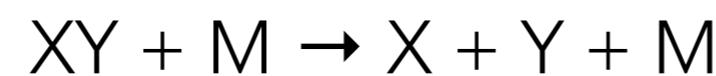
# Formation and destruction of molecules

## Gas-phase chemistry

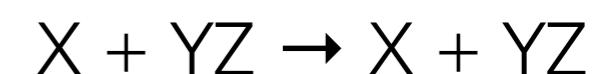
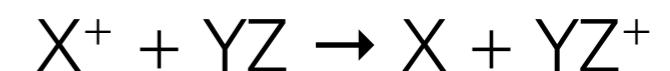
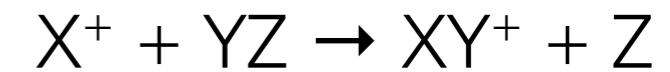
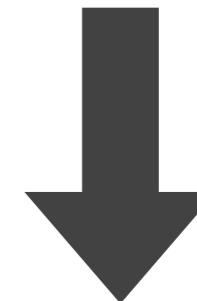
Bond formation



Bond  
destruction



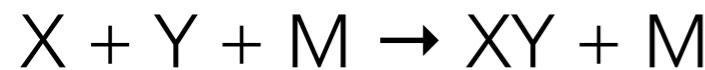
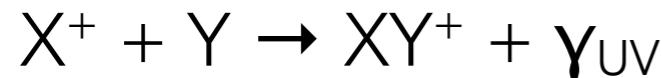
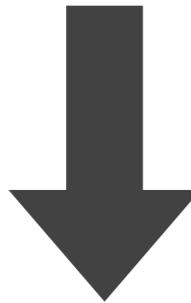
Bond  
rearrangement



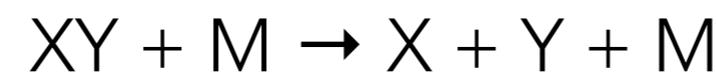
# Formation and destruction of molecules

## Gas-phase chemistry

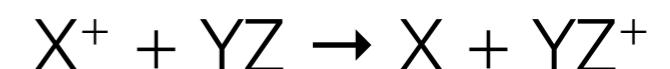
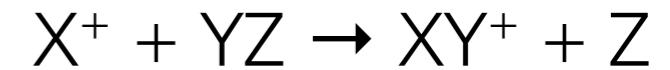
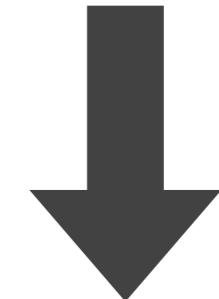
Bond formation



Bond  
destruction



Bond  
rearrangement



Interstellar and circumstellar conditions: chemical kinetics dominate

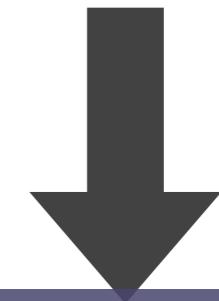
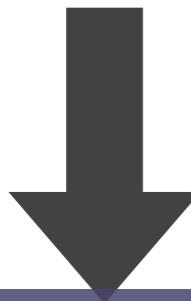
# Formation and destruction of molecules

## Gas-phase chemistry

Bond formation

Bond  
destruction

Bond  
rearrangement

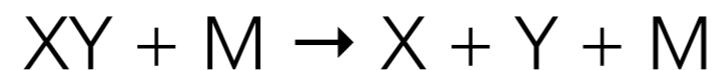
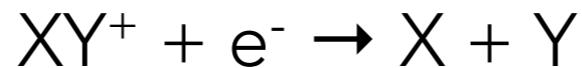
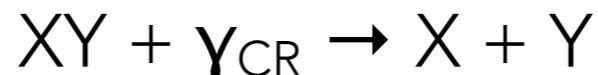


$X^+ + Y^- \rightarrow XY$

$X^- + Y^+ \rightarrow XY$

$X + Y + M \rightarrow XY + M$

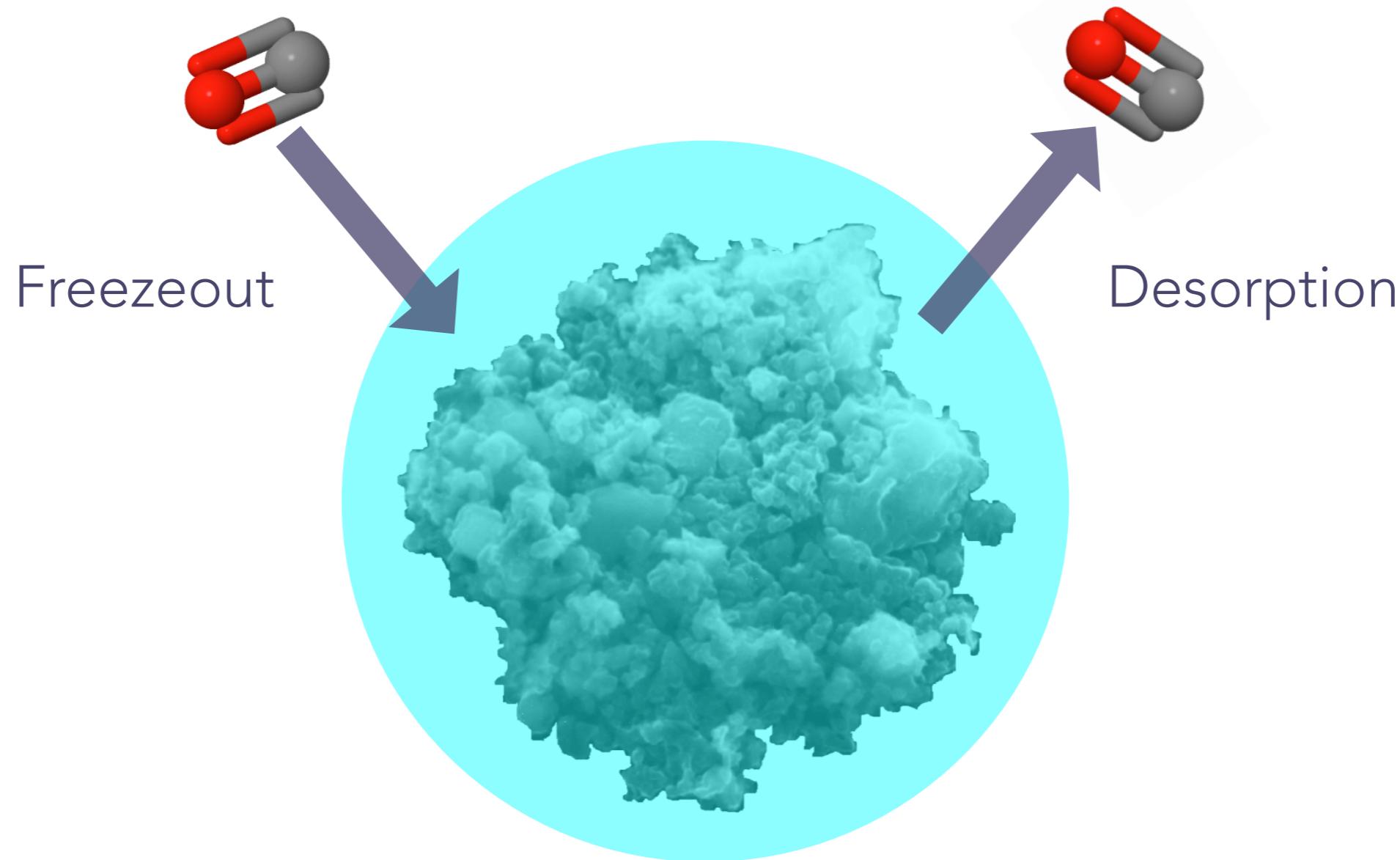
We know ion-molecule chemistry is important in disks because we see cations:  $CH^+$ ,  $HCO^+$ ,  $N_2H^+$ ,  $DCO^+$ ,  $N_2D^+$ ,  $YZ^+$



Interstellar and circumstellar conditions: chemical kinetics dominate

# Formation and destruction of molecules

## Grain-surface processes



Dust grains act as a third body for association reactions

# Formation and destruction of molecules

## Grain-surface processes

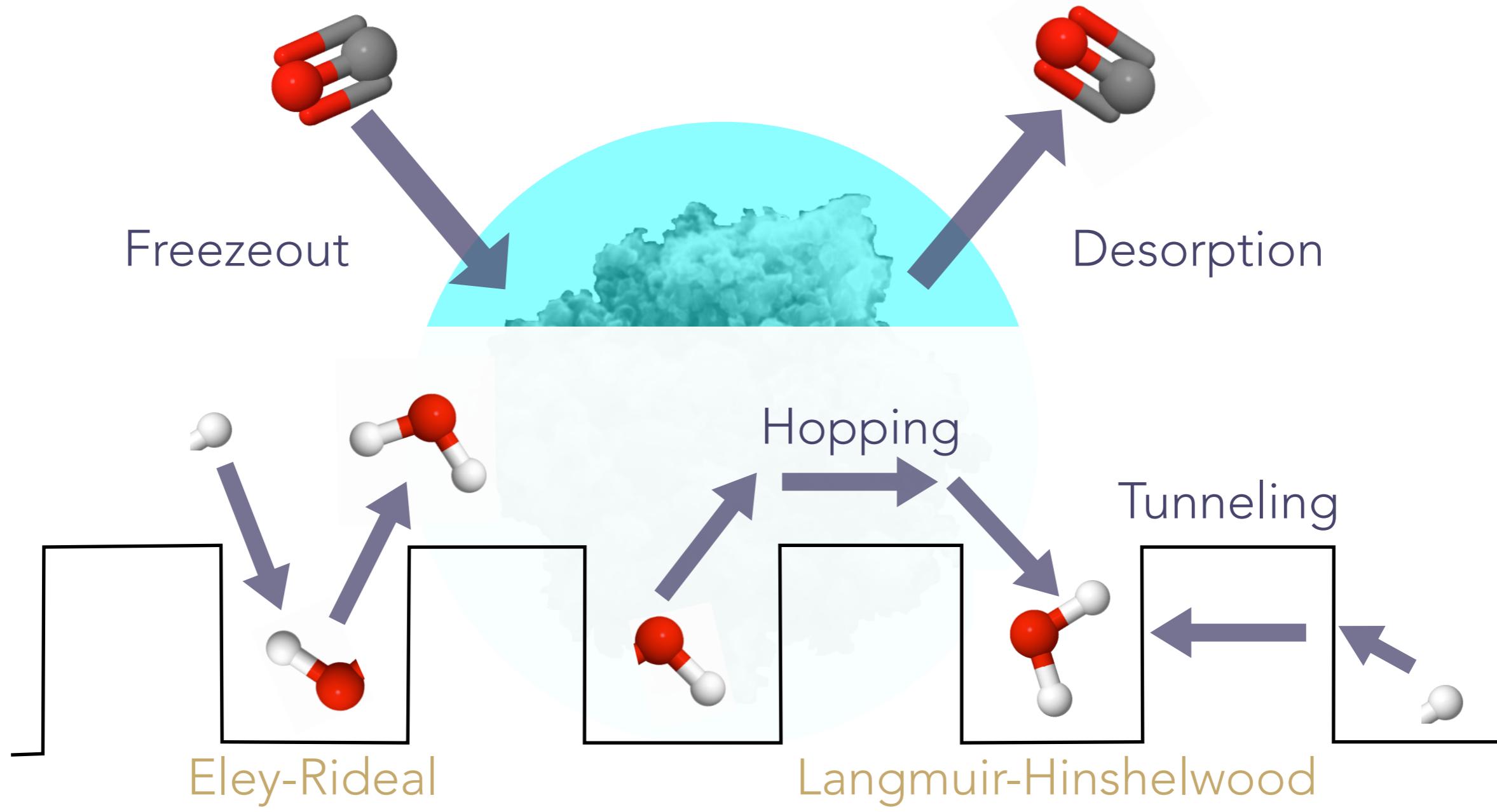


We know freezeout and desorption are important in disks because we see midplane depletion of e.g., CO, and gas-phase molecules present where they would otherwise be ice

Dust grains act as a third body for association reactions

# Formation and destruction of molecules

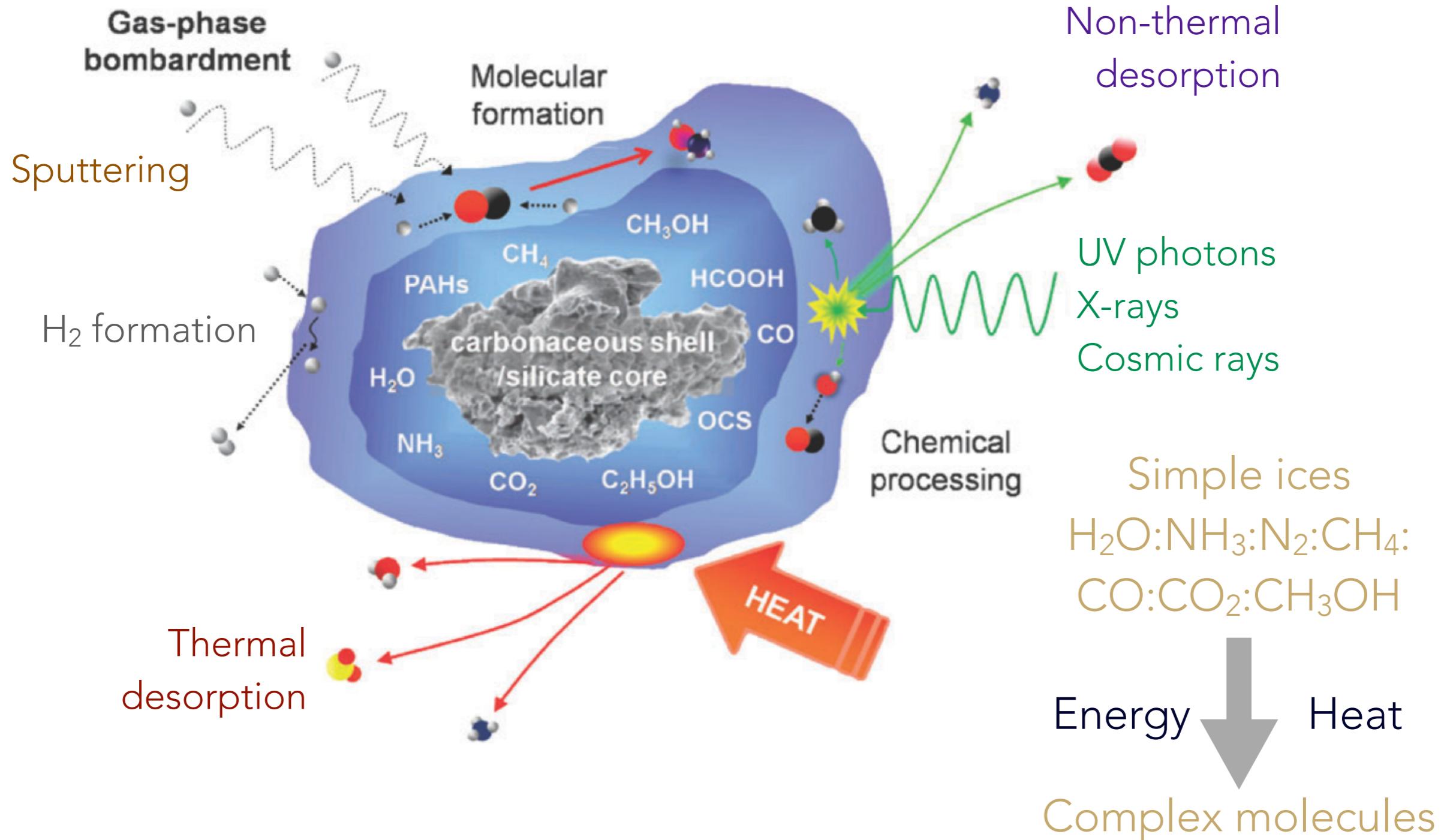
## Grain-surface processes



Dust grains act as a third body for association reactions

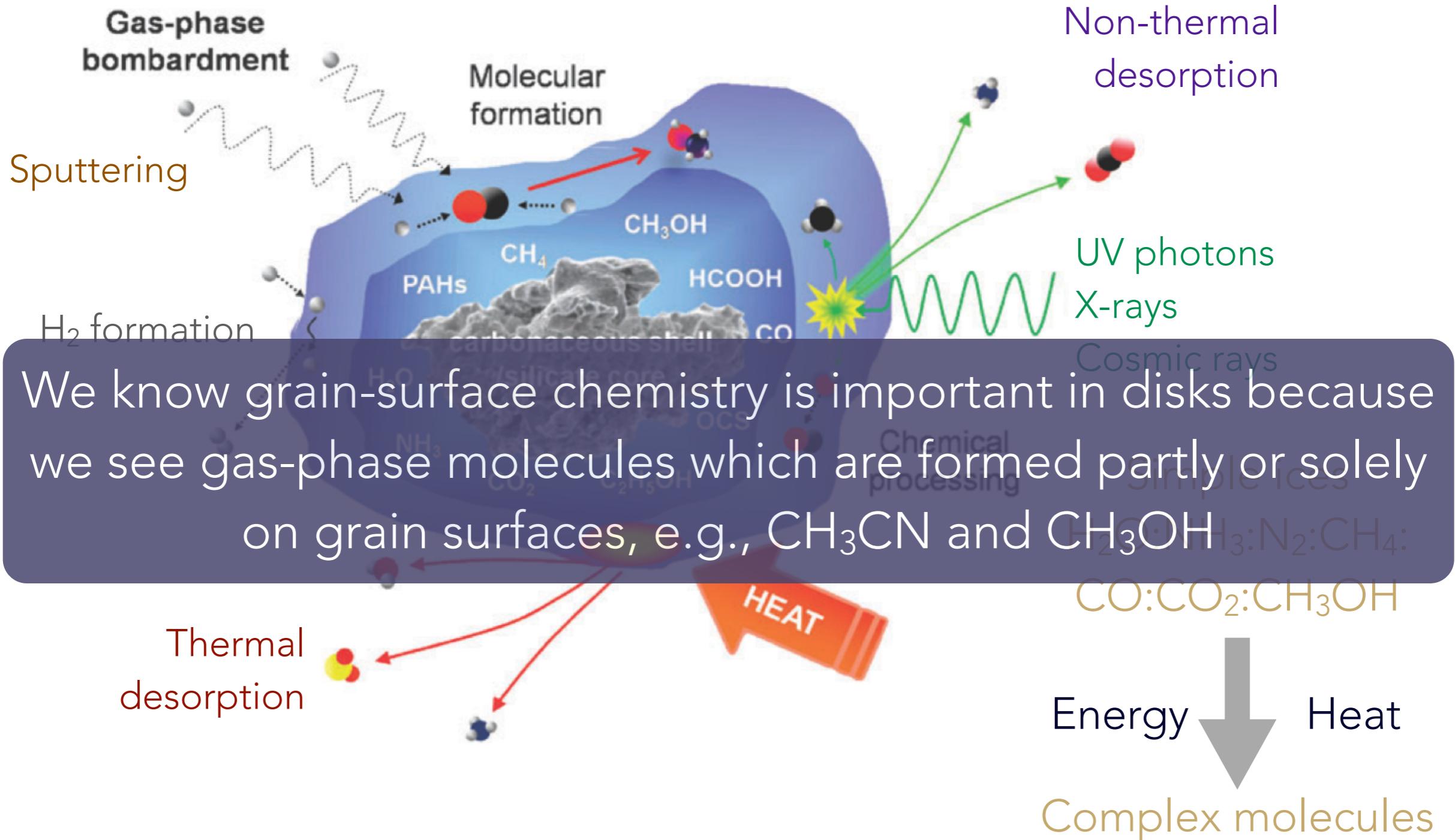
# Formation and destruction of molecules

## Grain-surface processes



# Formation and destruction of molecules

## Grain-surface processes



# Grain-surface chemistry increases complexity

# Grain-surface chemistry increases complexity



# Calculating the chemistry

Molecular abundances are a function of disk conditions and time

$$n_X = F [T_{\text{gas}}, T_{\text{dust}}, n_{\text{gas}}, F_{\text{UV}}(\lambda), F_{\text{XR}}(E_{\text{XR}}), \zeta_{\text{CR}}, \sigma_{\text{dust}}]$$

$$\frac{dn_X}{dt} = F_X - D_X$$

$$\frac{dn_X}{ds} = F_X - D_X$$

$$s = (r, z) \quad \text{or} \quad (\rho, \phi, z)$$

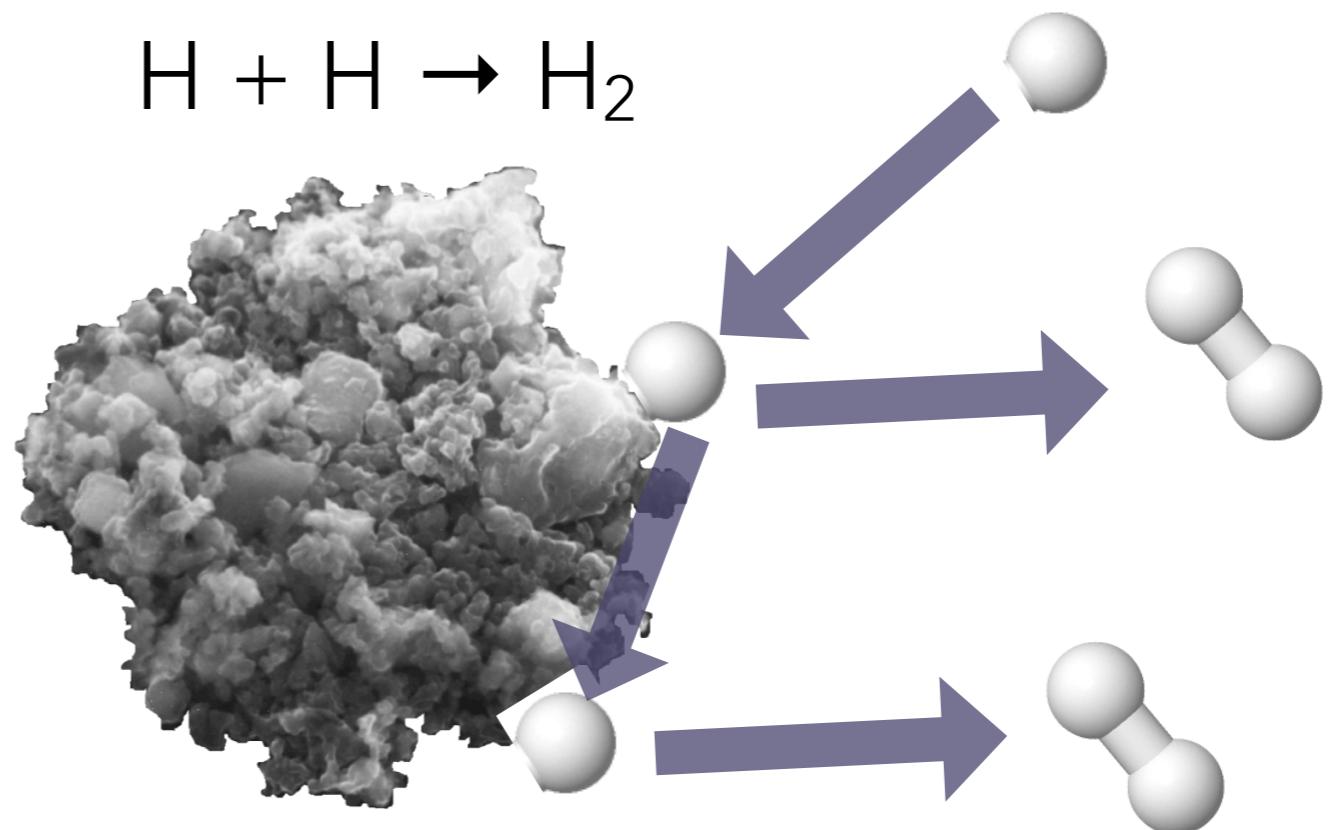
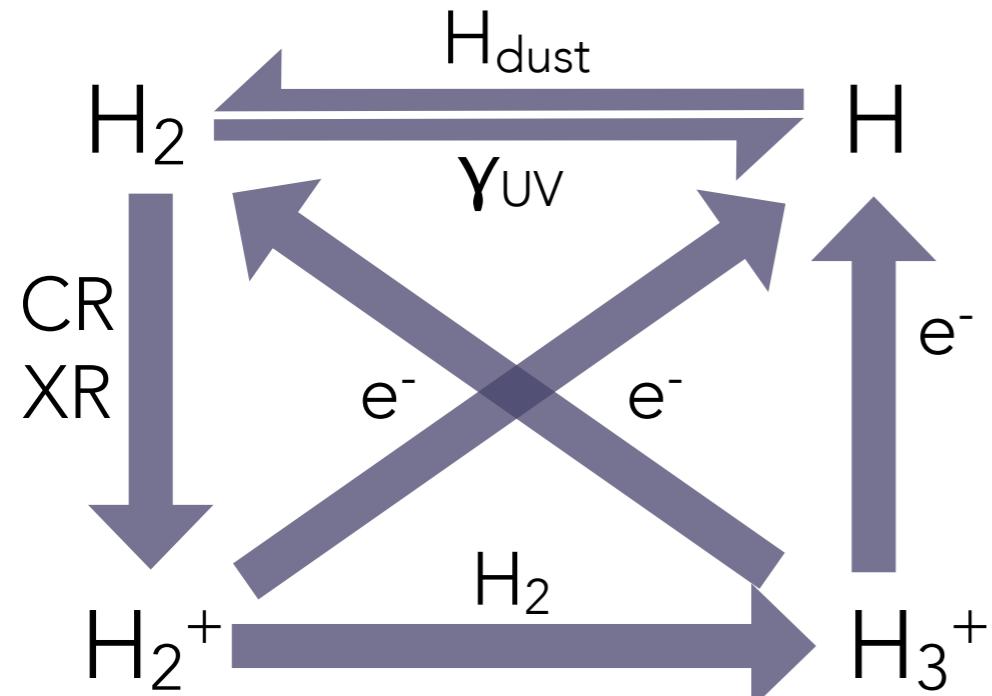
Chemistry in disks is not in equilibrium: steady state is possible

# A “simple” chemical network: H<sub>2</sub>

Dense gas:  
 $A_v \gg 1$  mag



Irradiated gas:  
 $A_v < 1$  mag

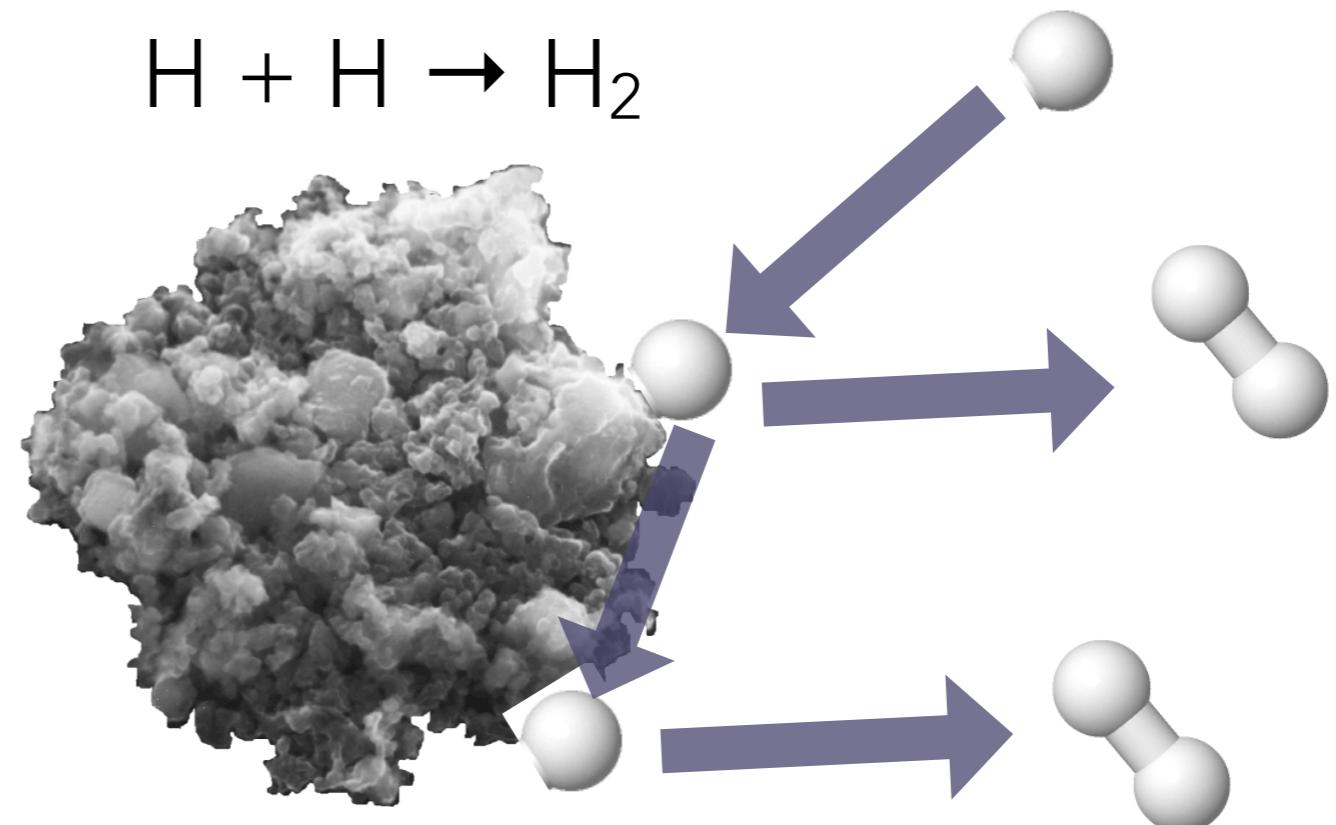
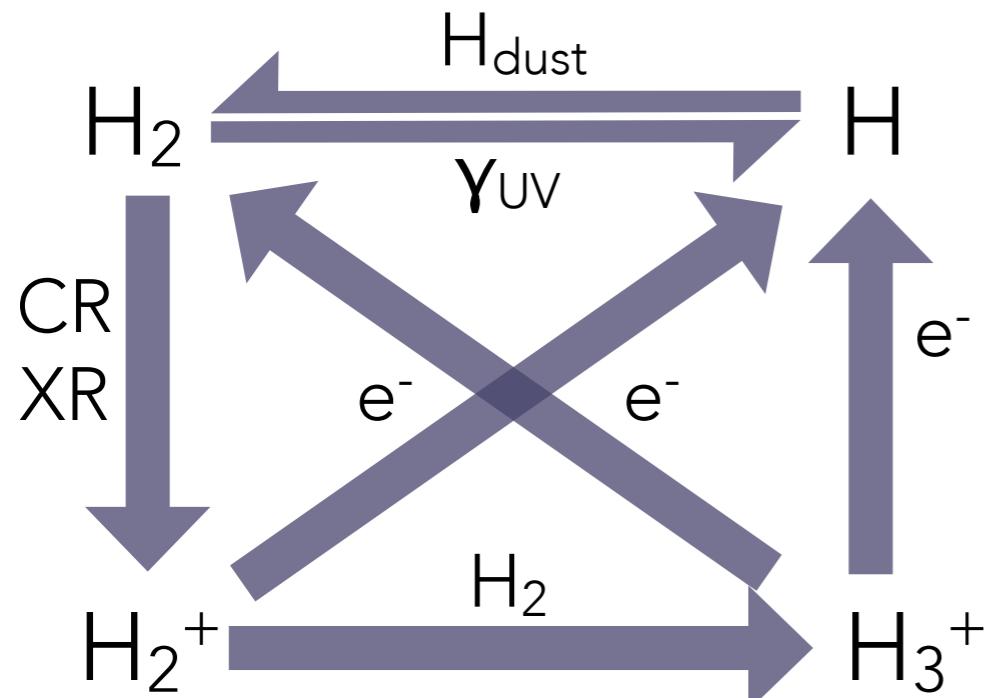


# A “simple” chemical network: H<sub>2</sub>

Dense gas:  
 $A_v \gg 1$  mag

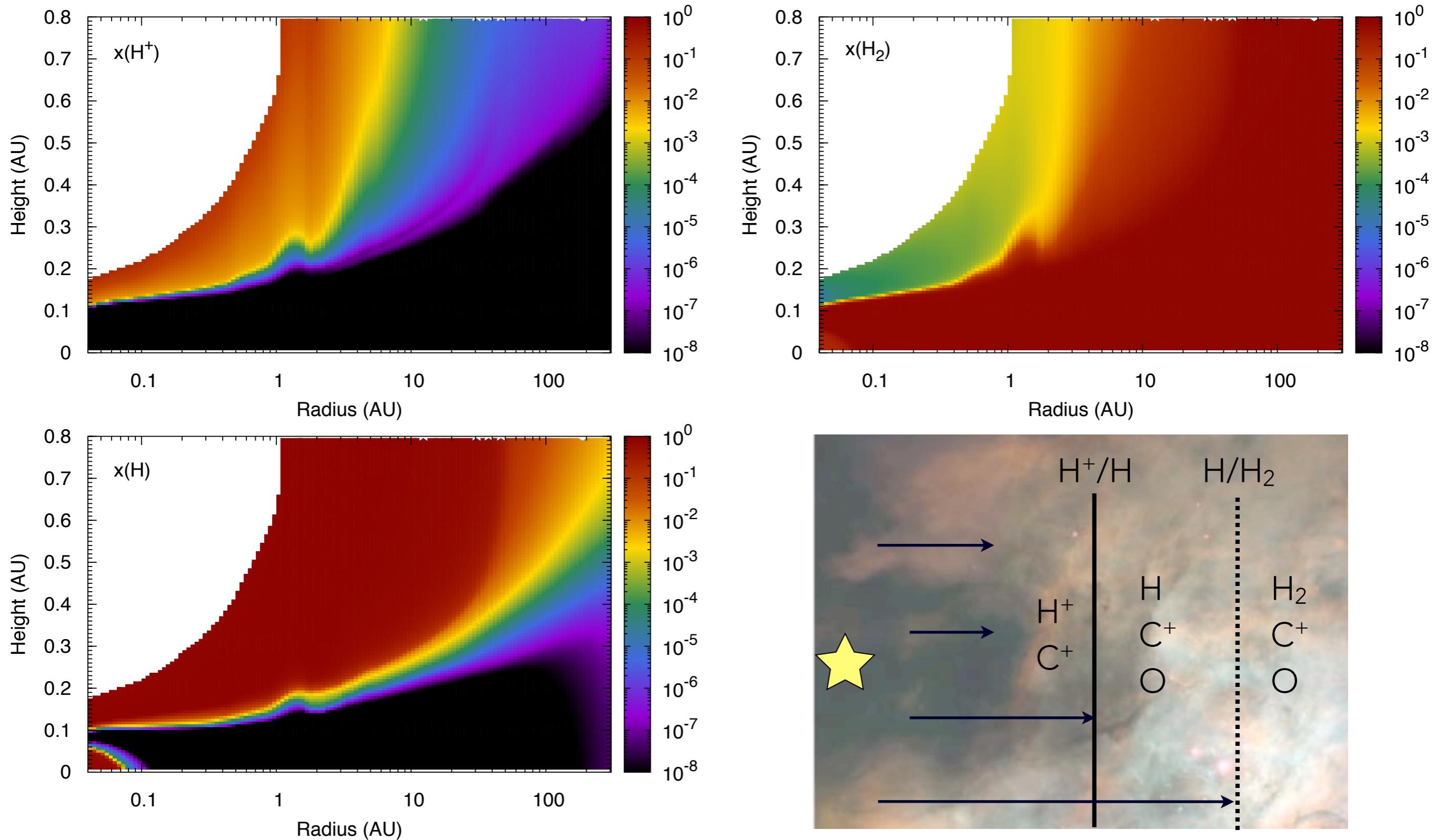


Irradiated gas:  
 $A_v < 1$  mag



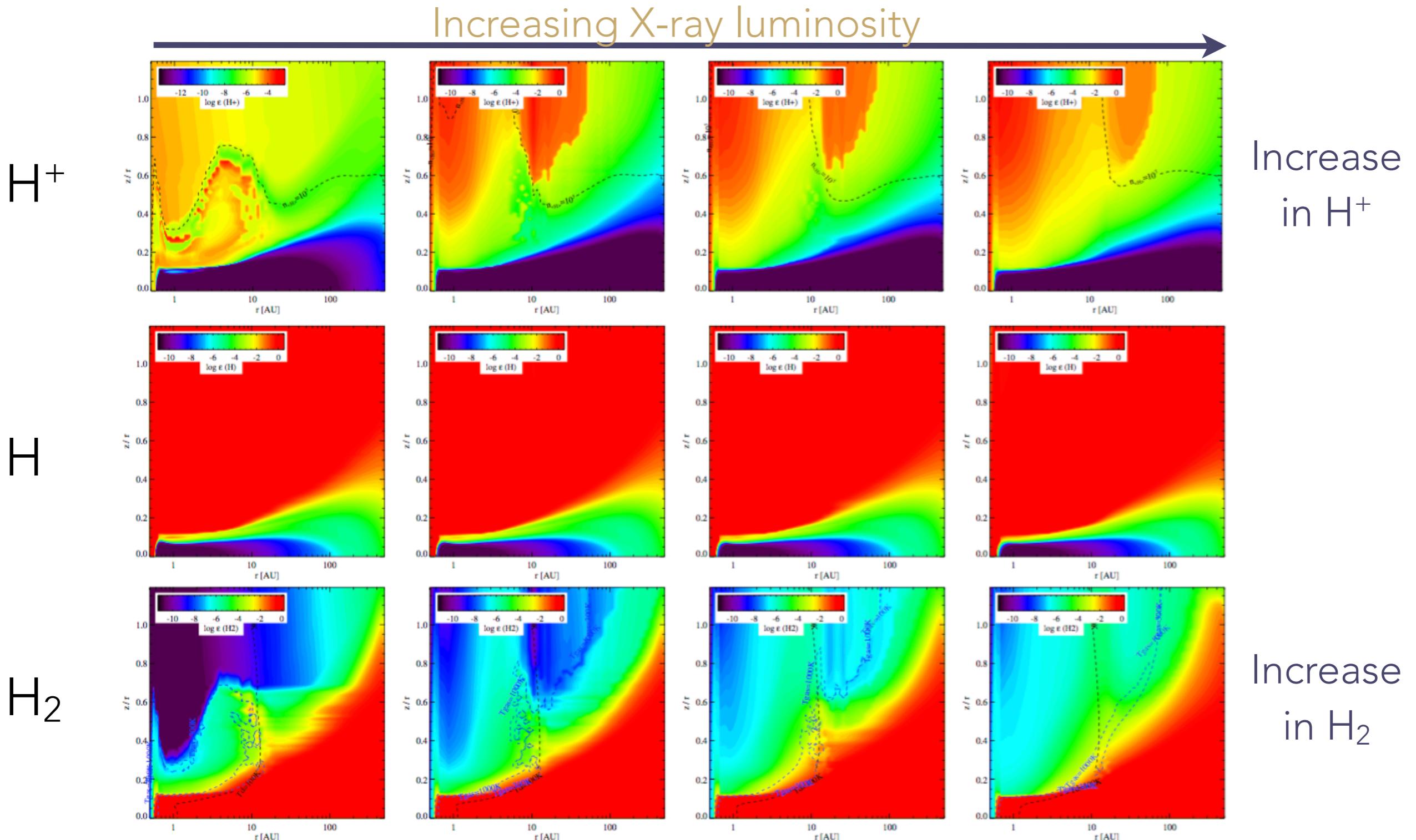
$\text{H}_2$  forms almost exclusively on dust grains

# $H^+/H/H_2$ in protoplanetary disks

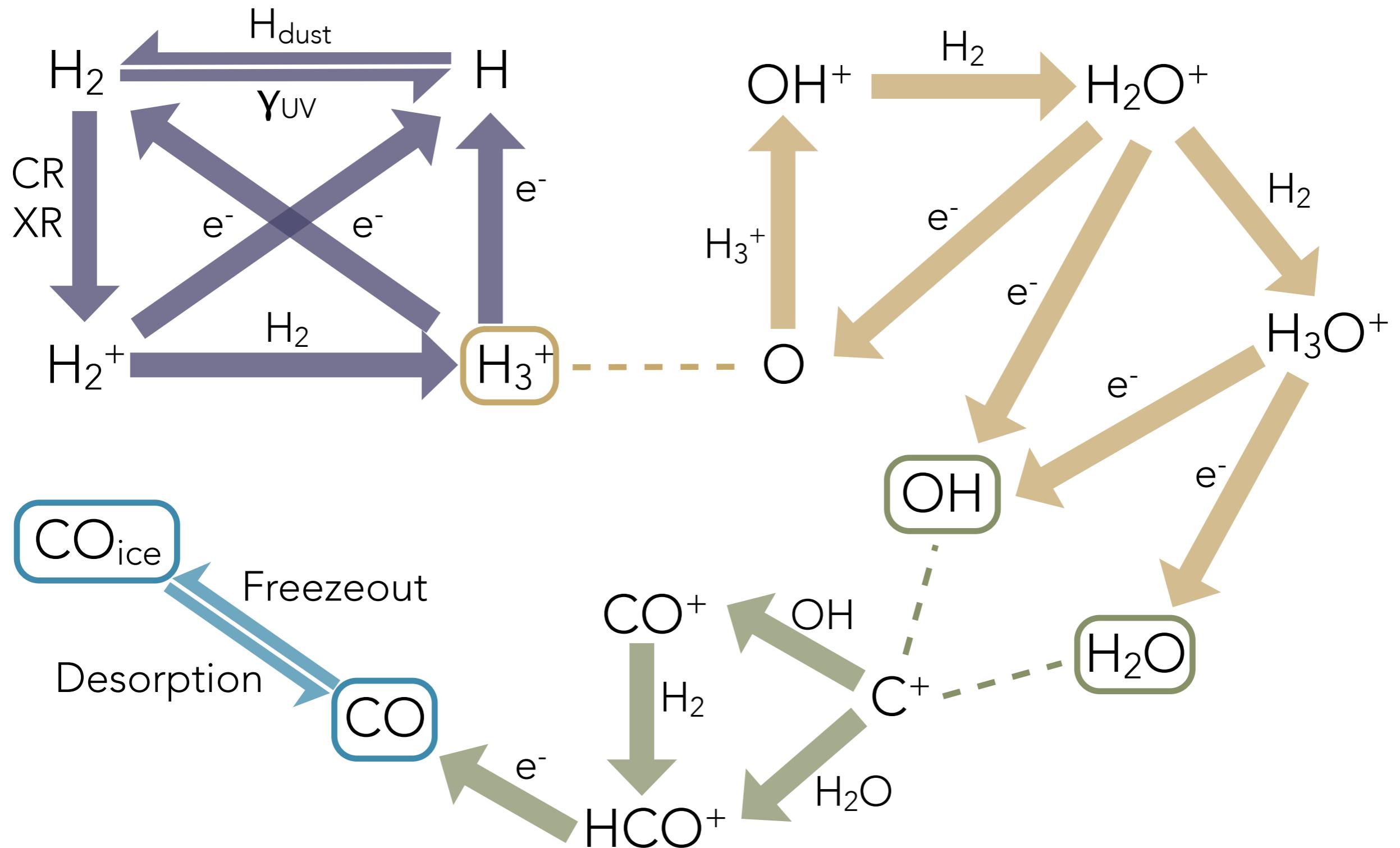


# $H^+/H/H_2$ in protoplanetary disks

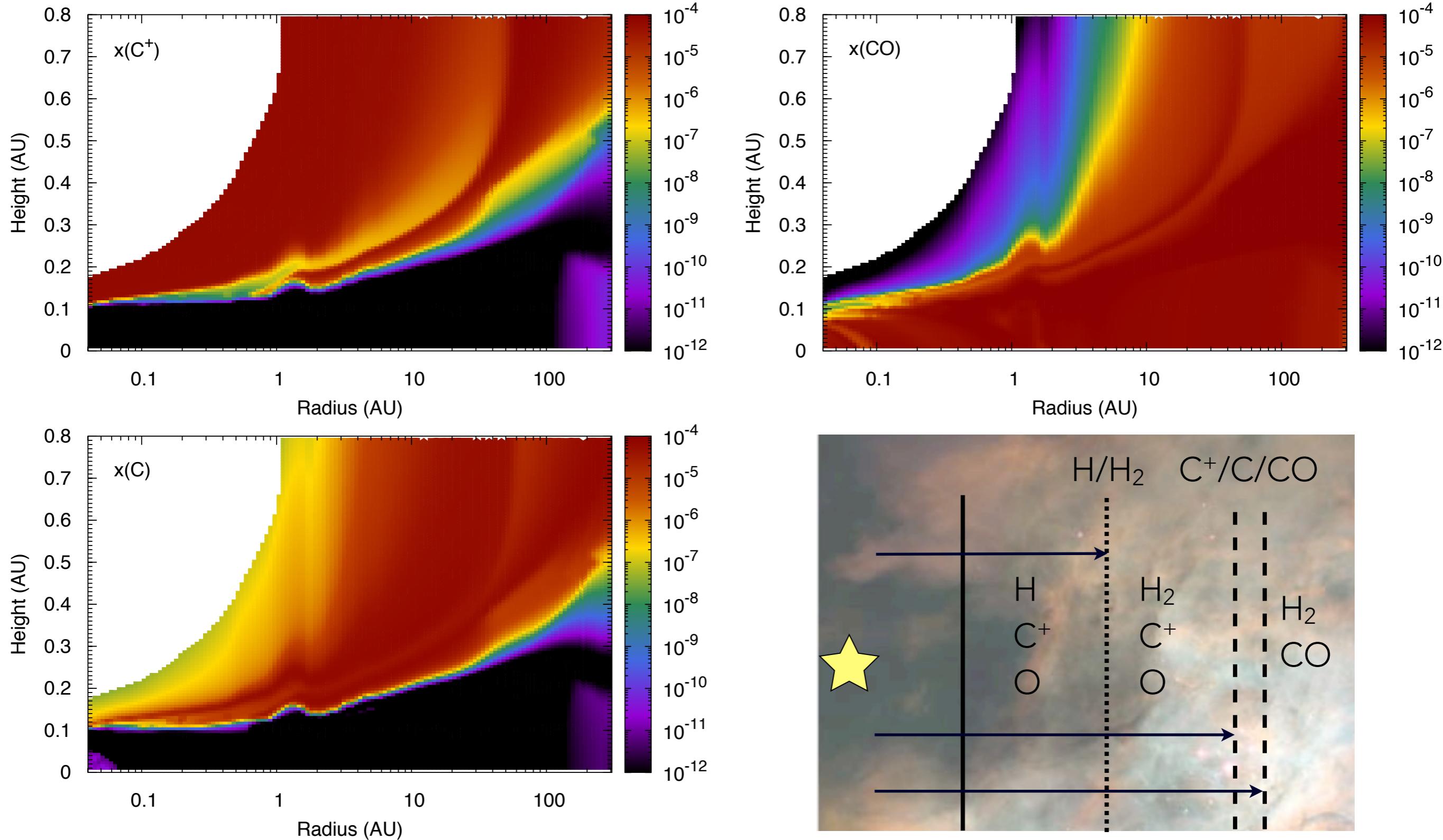
X-rays also influence the  $H^+/H/H_2$  transition regions



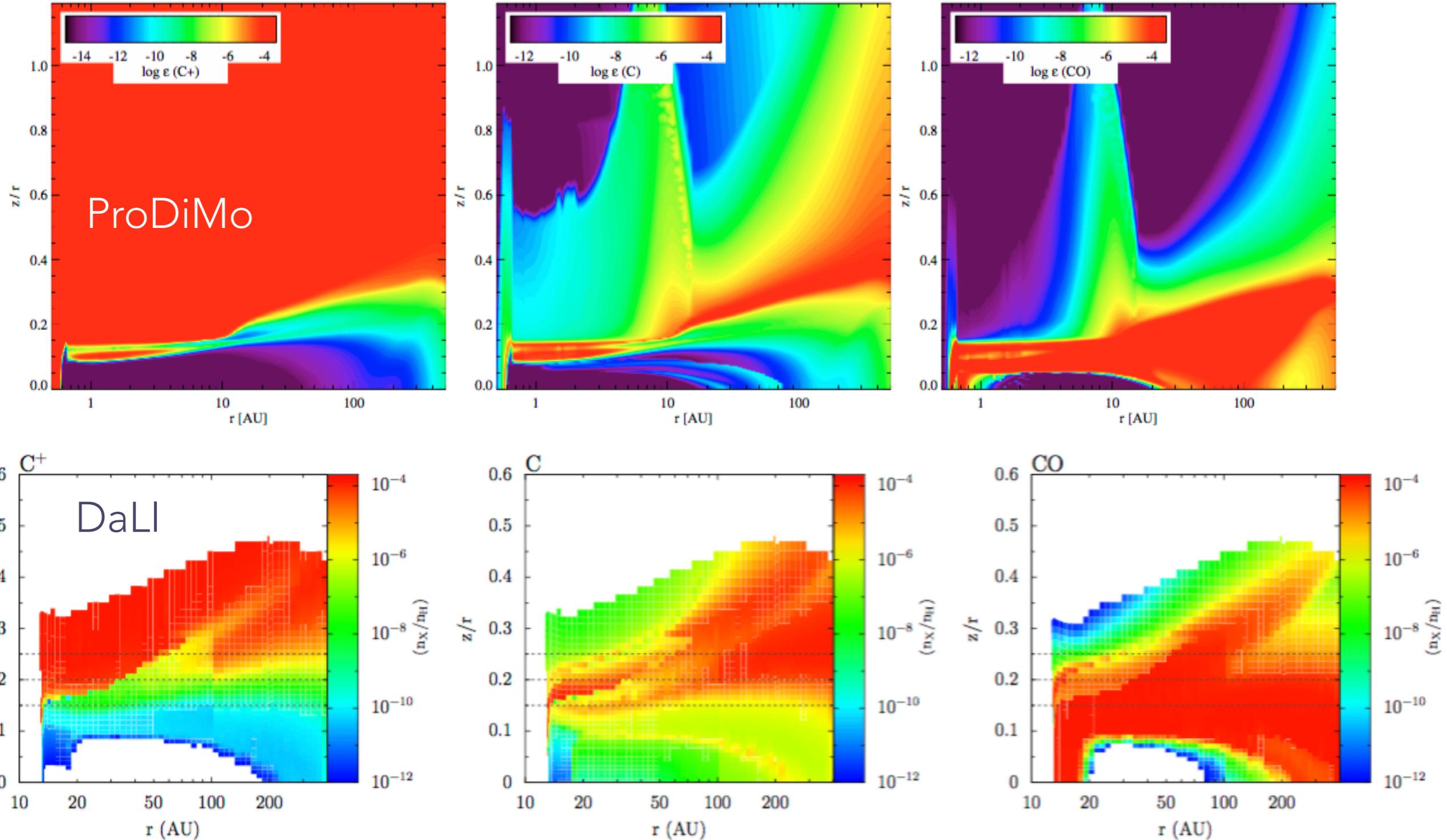
# A more complicated network: CO



# $C^+/C/CO$ in protoplanetary disks

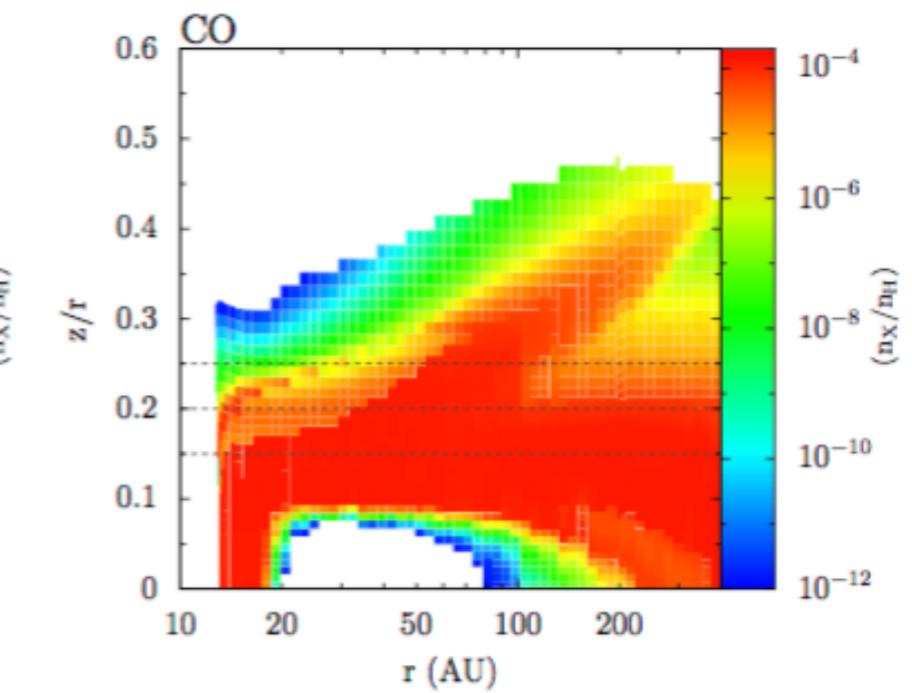
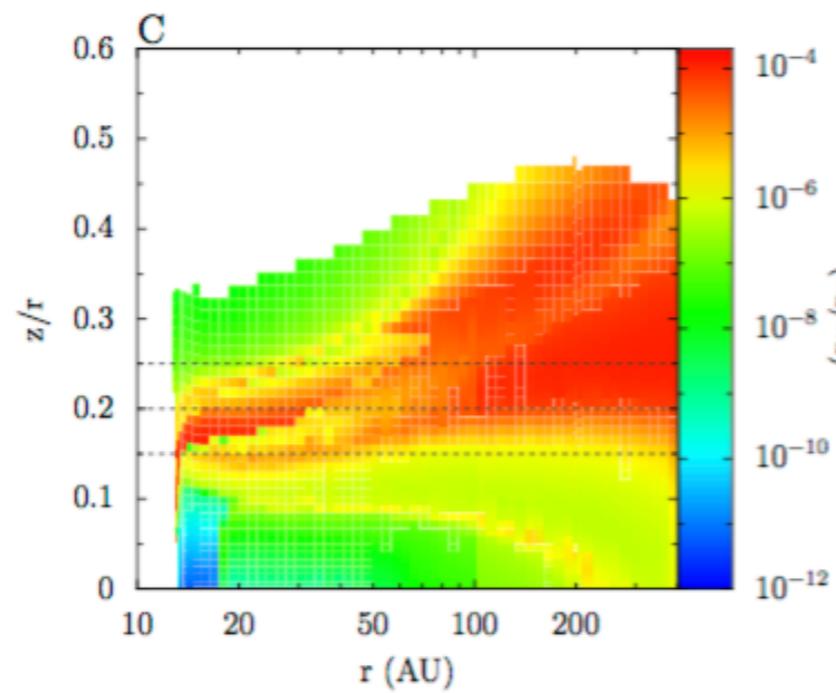
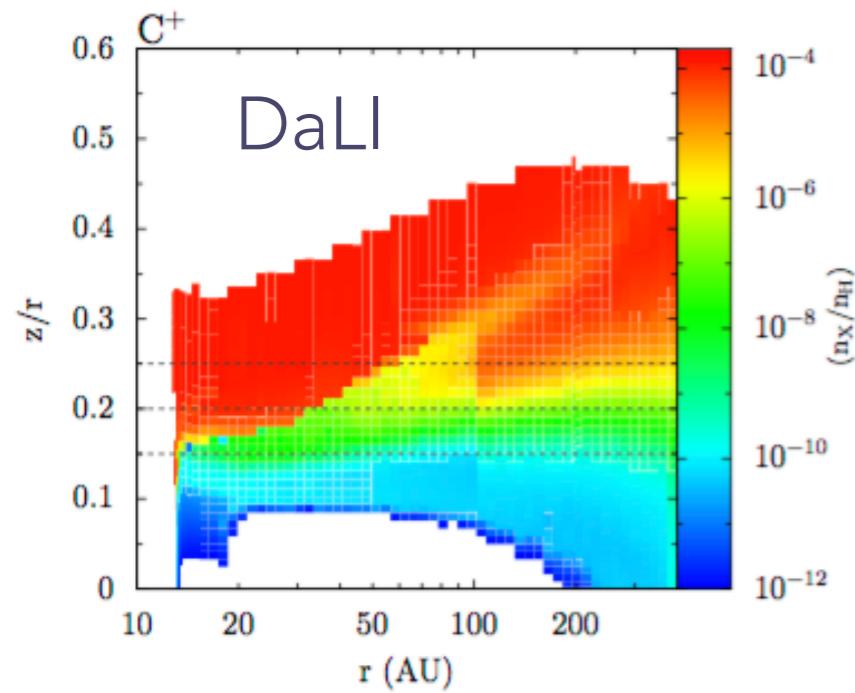
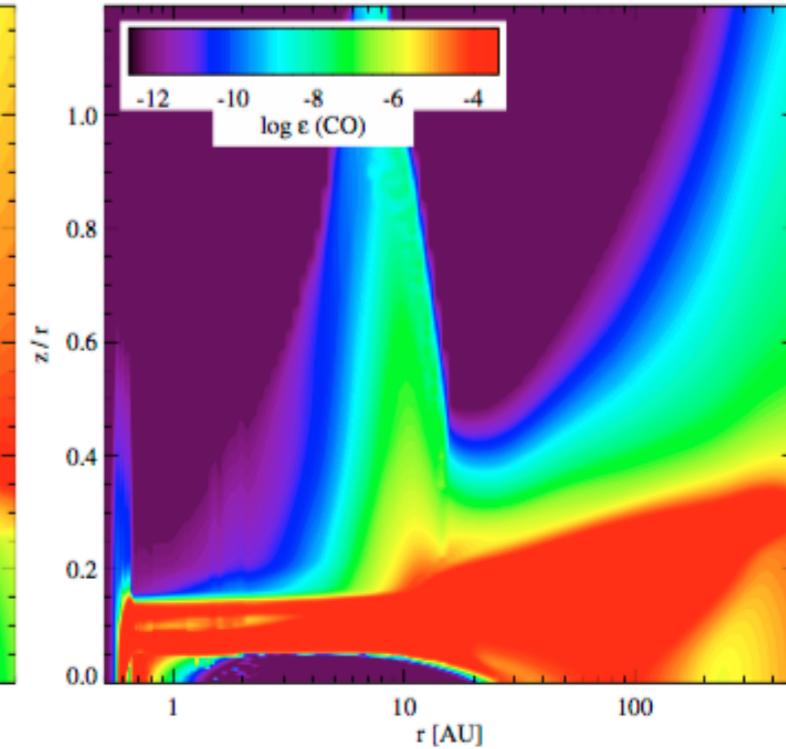
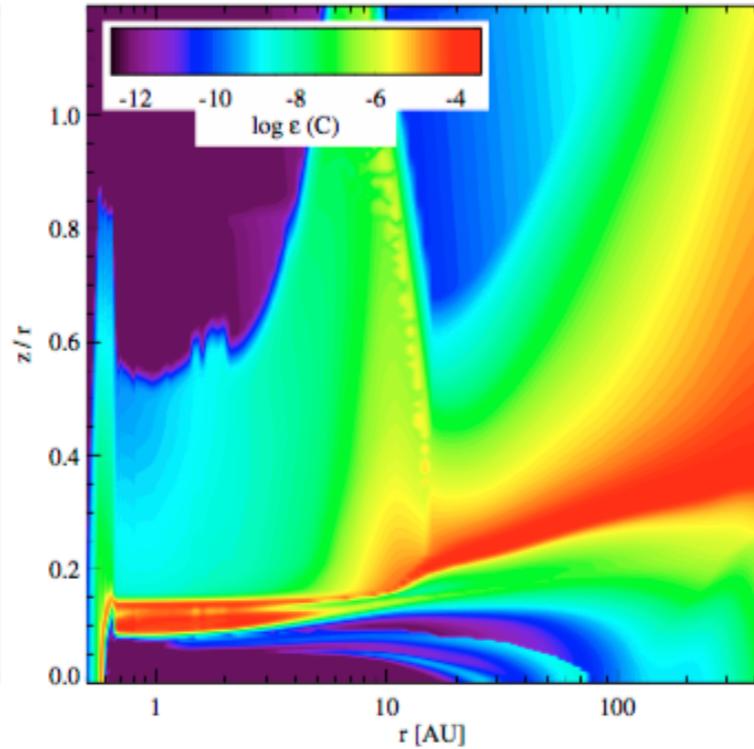
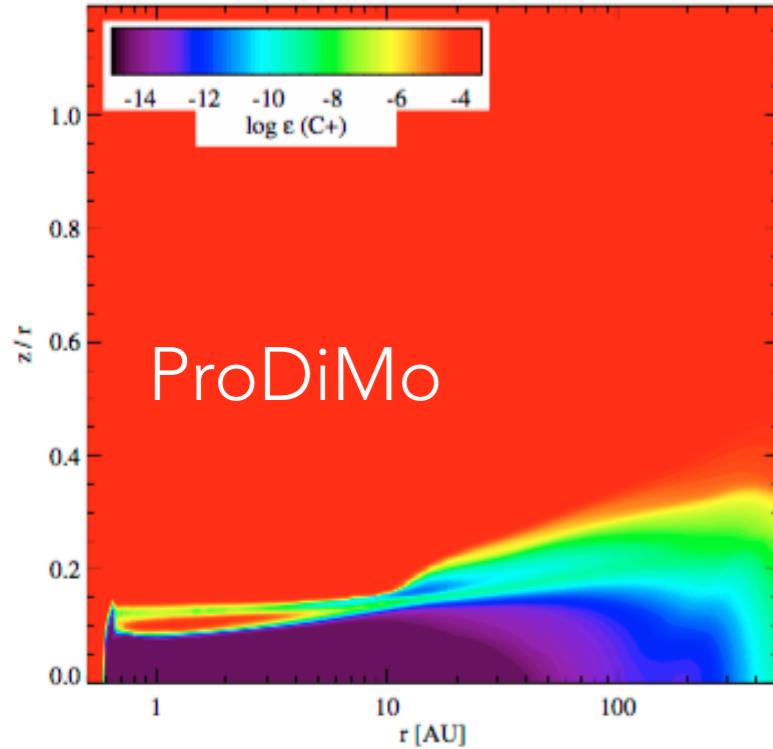


# $C^+$ /C/CO in protoplanetary disks

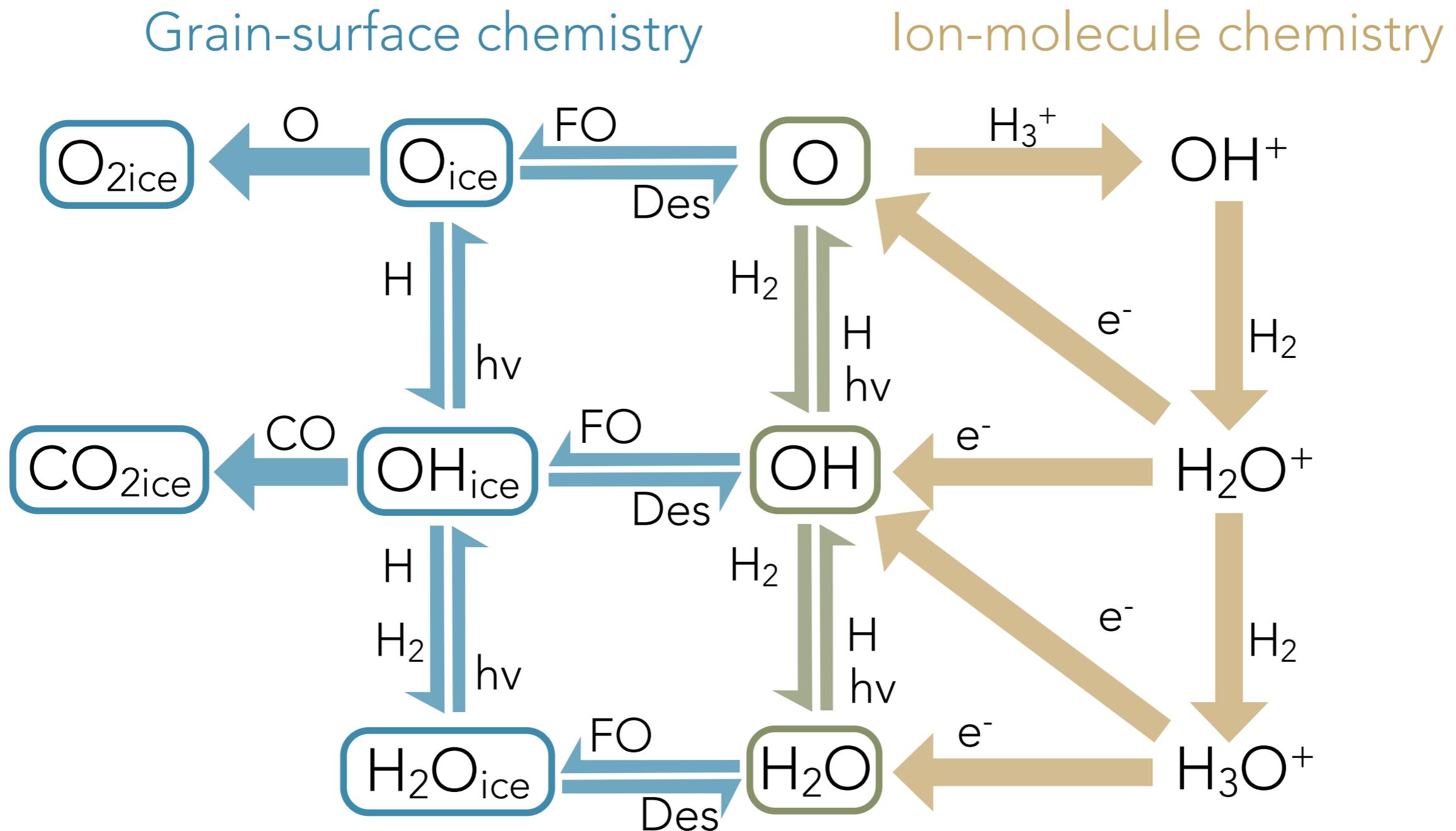


# $C^+$ /C/CO in protoplanetary disks

Similar stratification is seen in numerous physico-chemical models

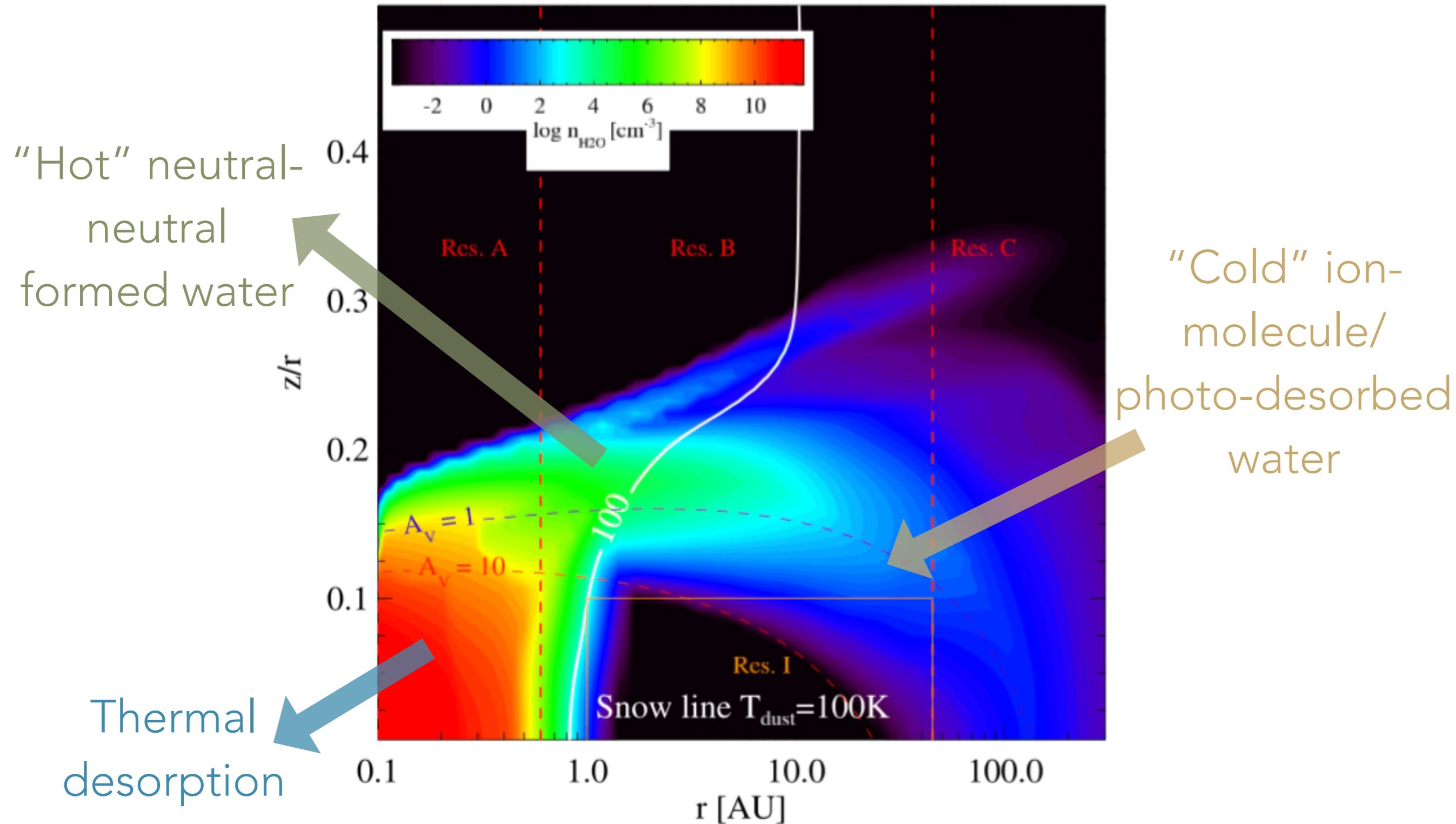


# An even more complicated network: $\text{H}_2\text{O}$

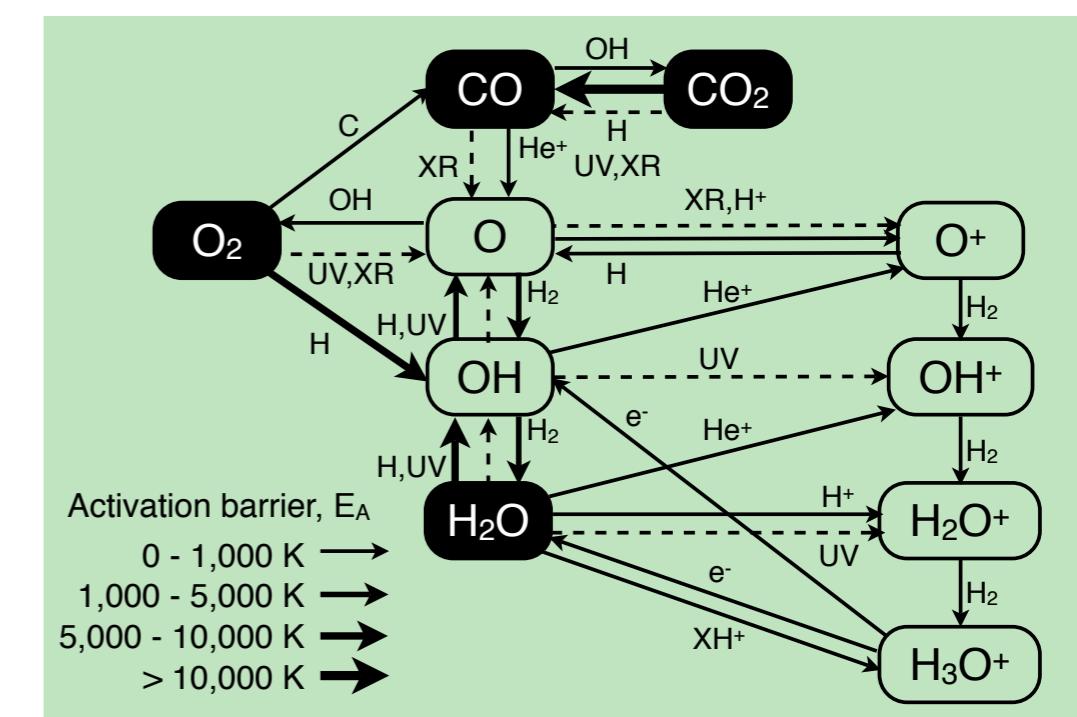
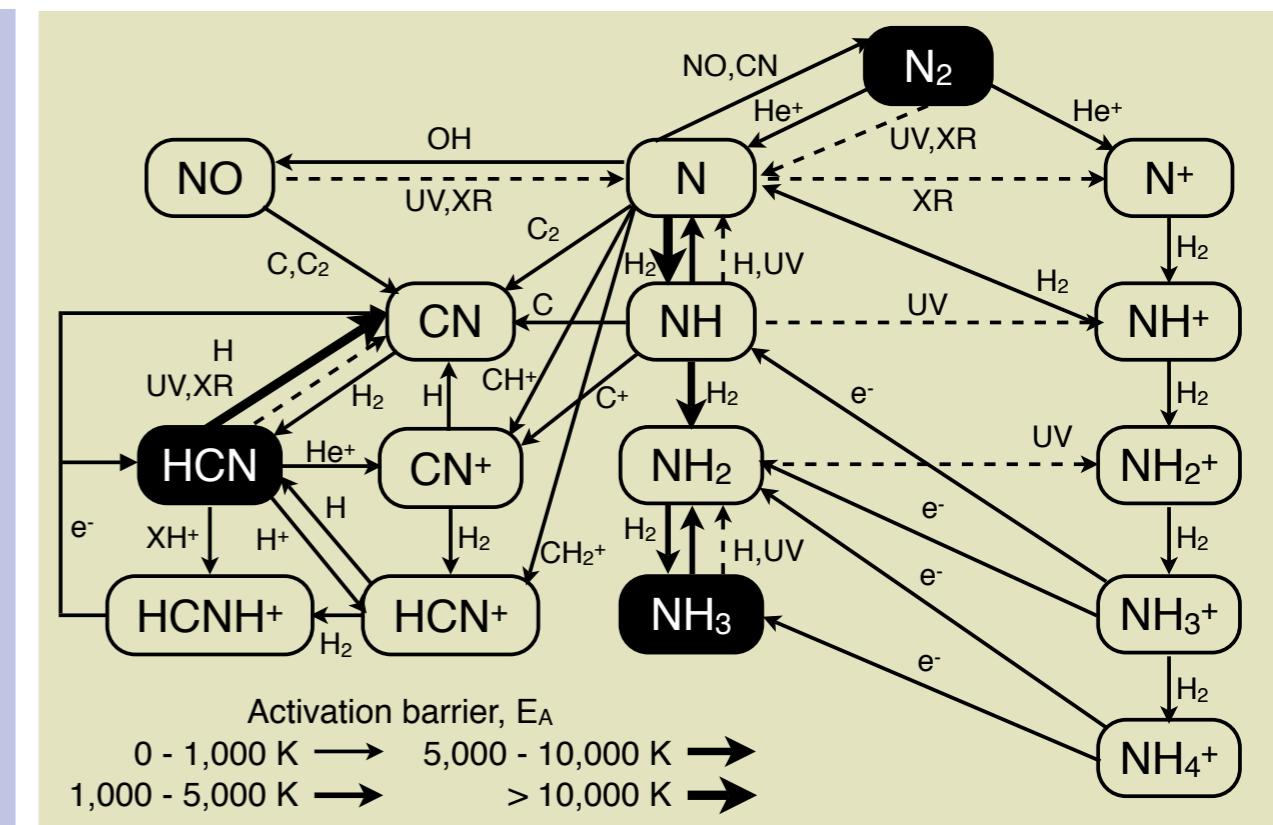
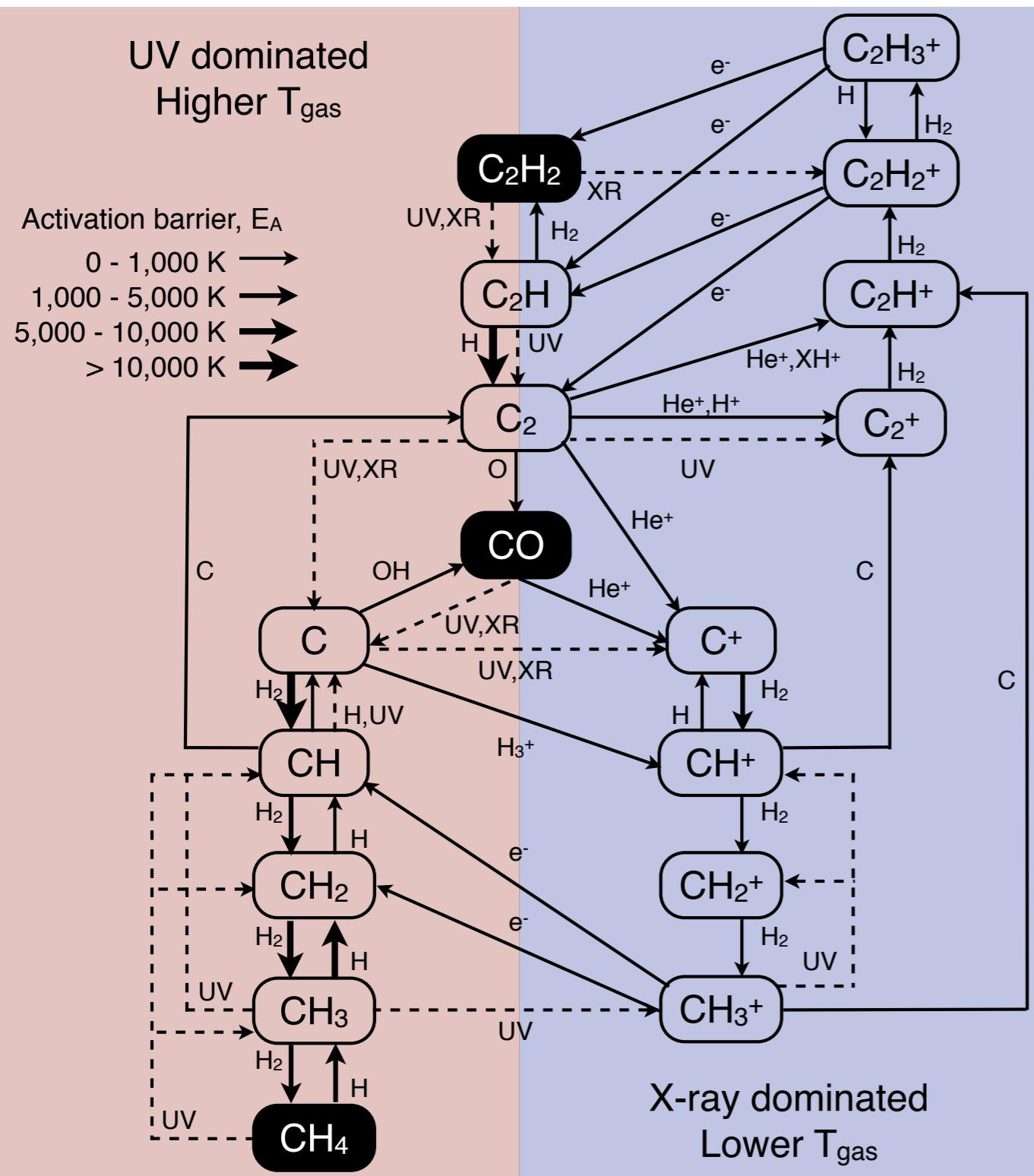


“Hot” neutral-neutral chemistry

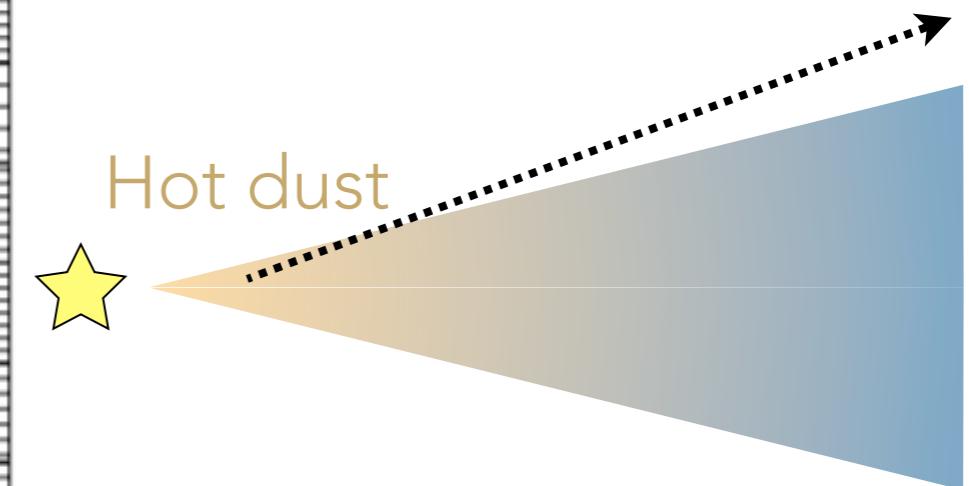
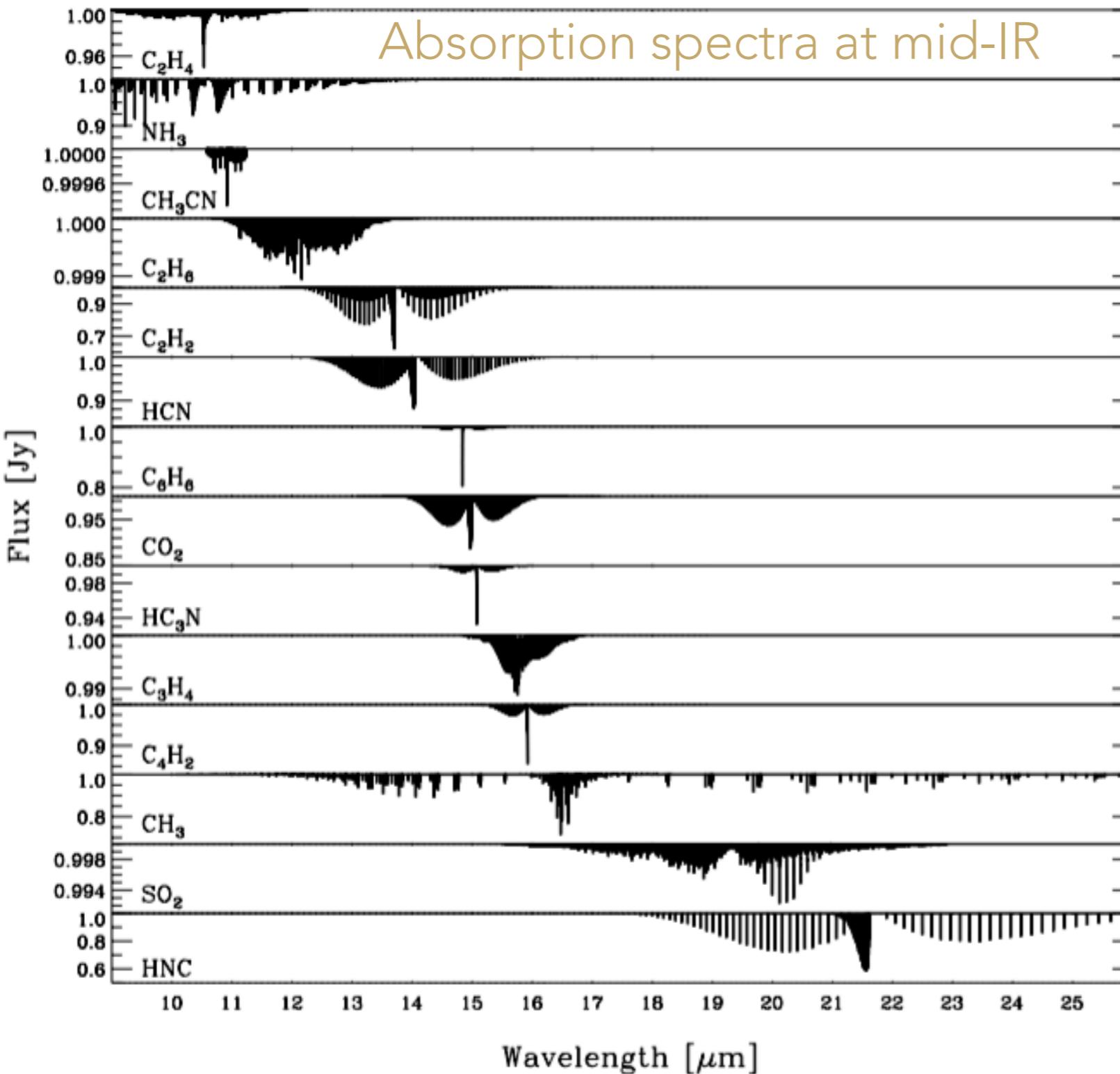
# Water in protoplanetary disks



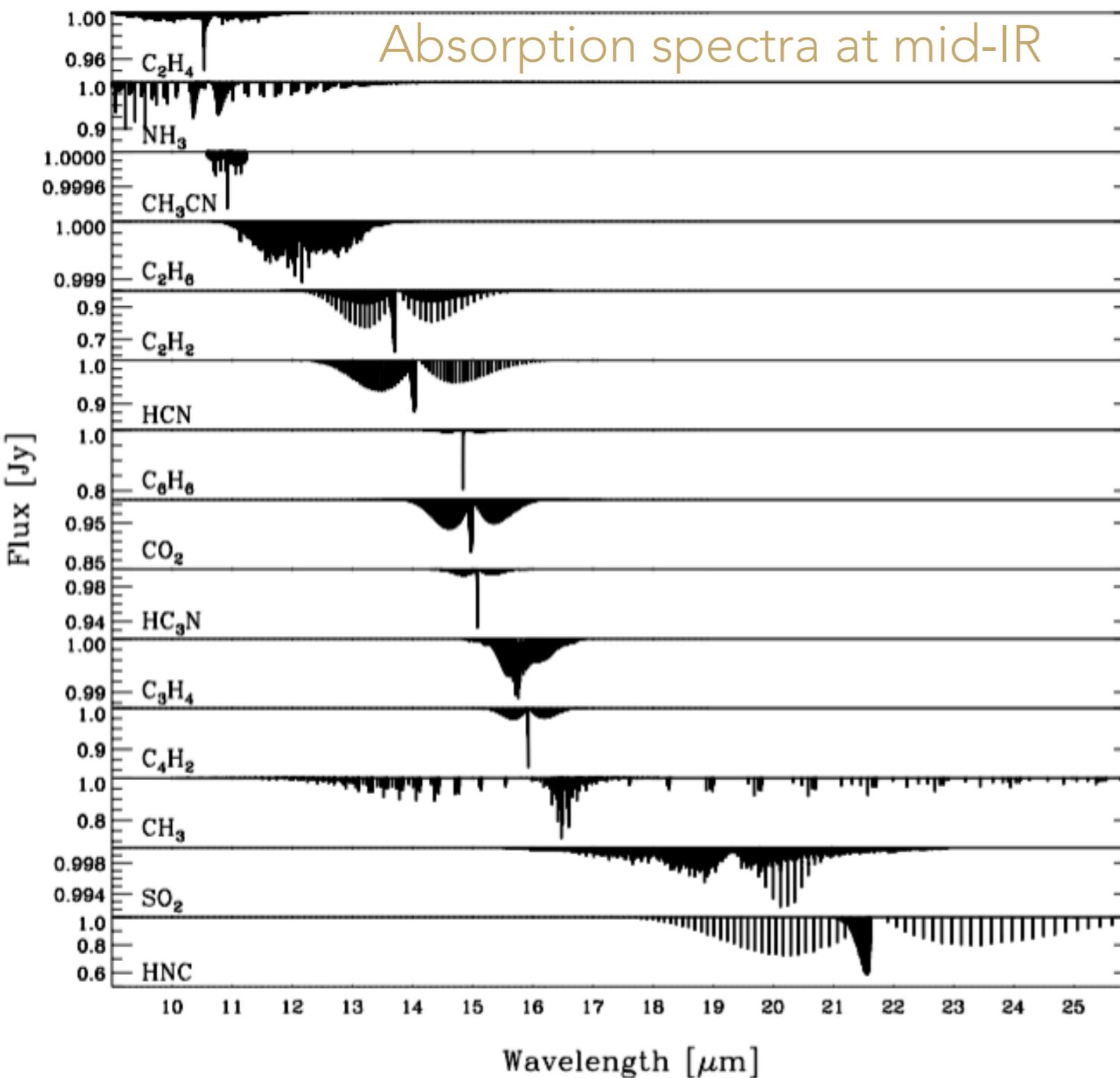
# Networks can quickly become complicated!



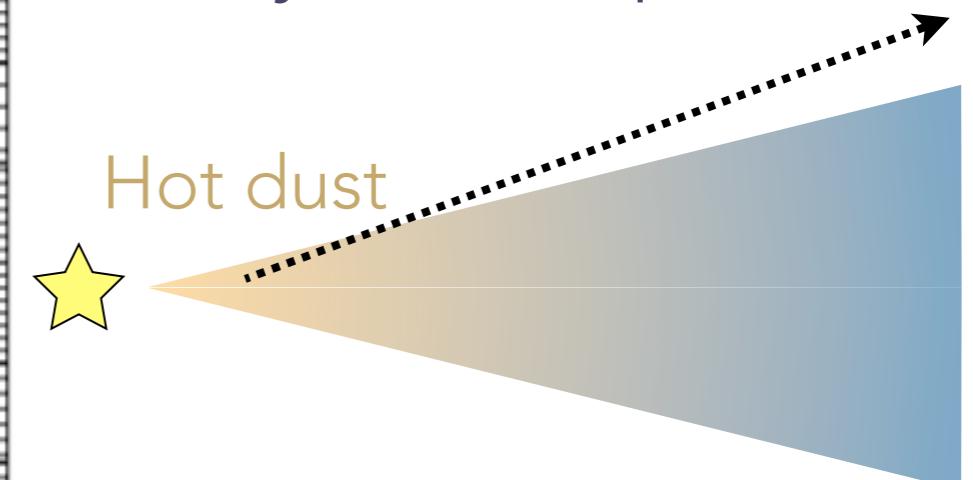
# Creating synthetic observations



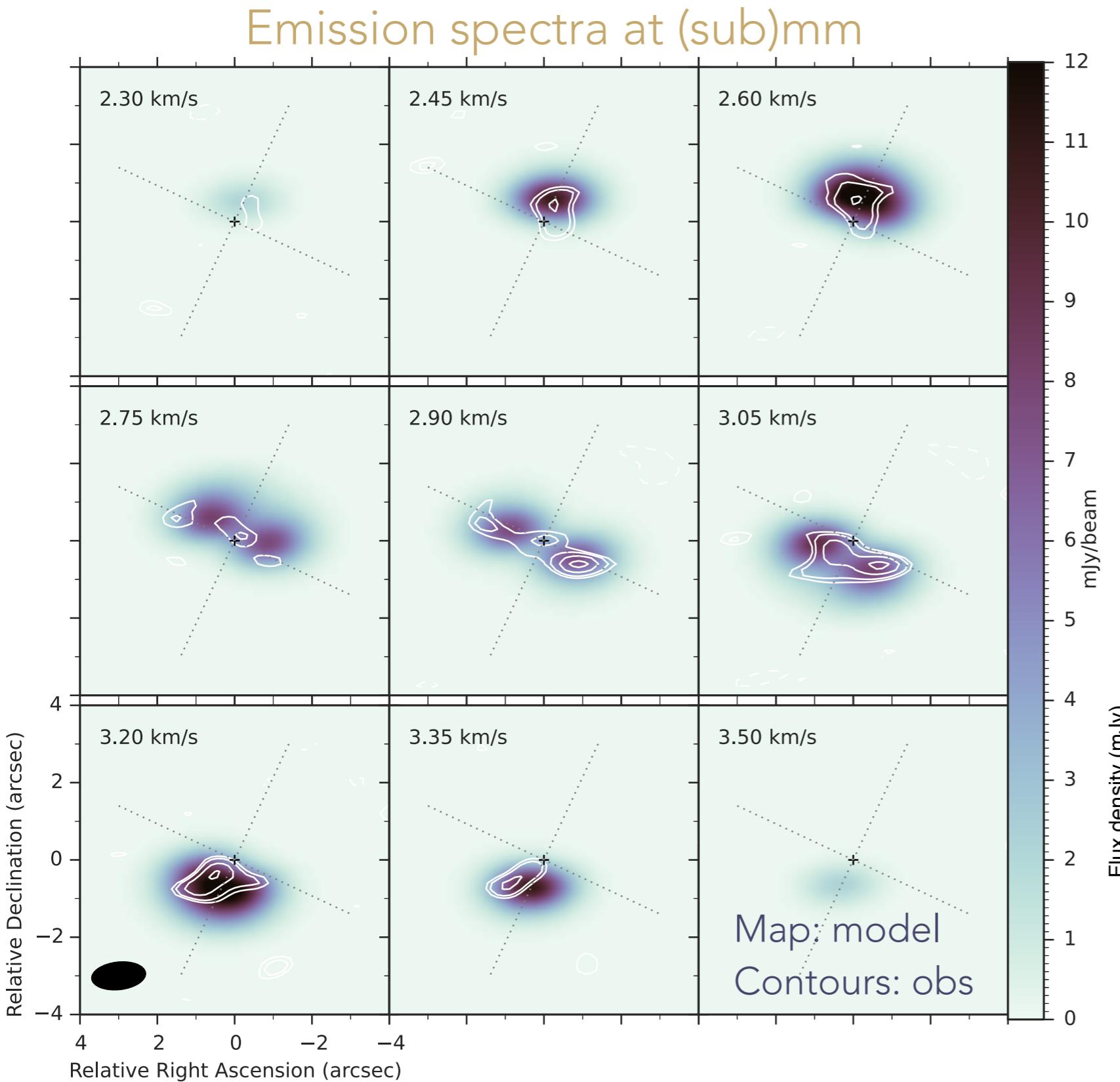
# Creating synthetic observations



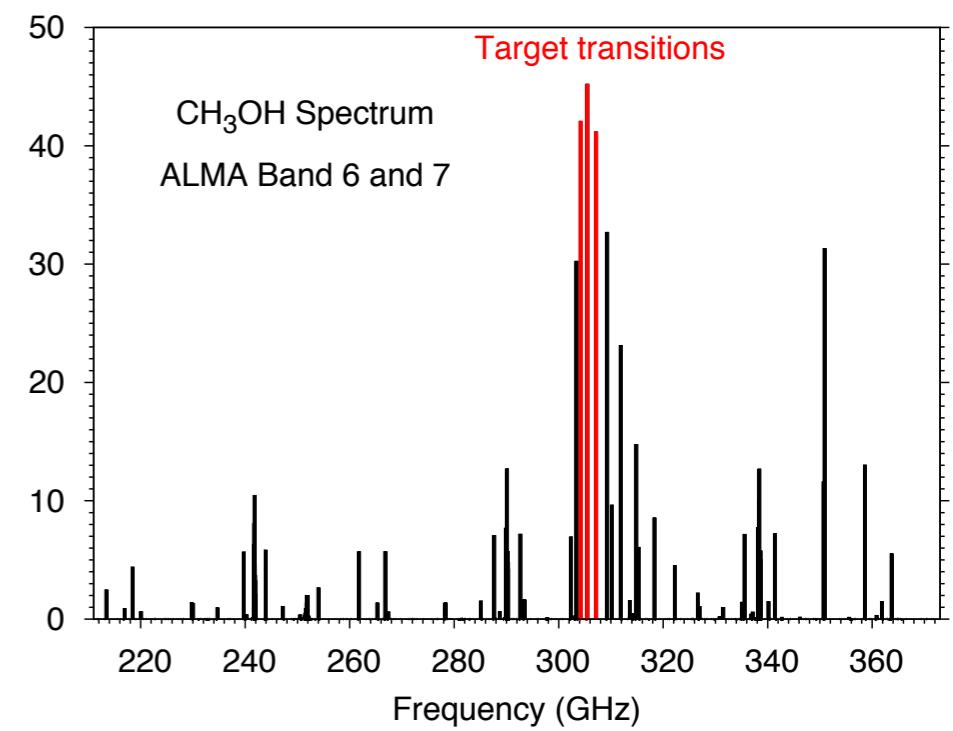
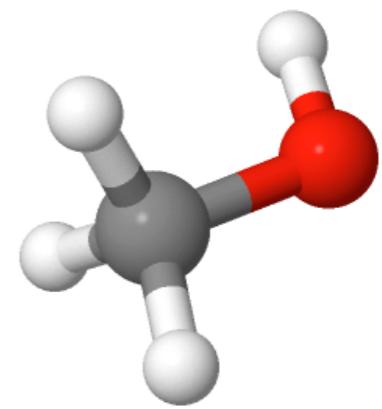
Radiative transfer codes (dust continuum and lines) are used along with molecular data to create synthetic spectra



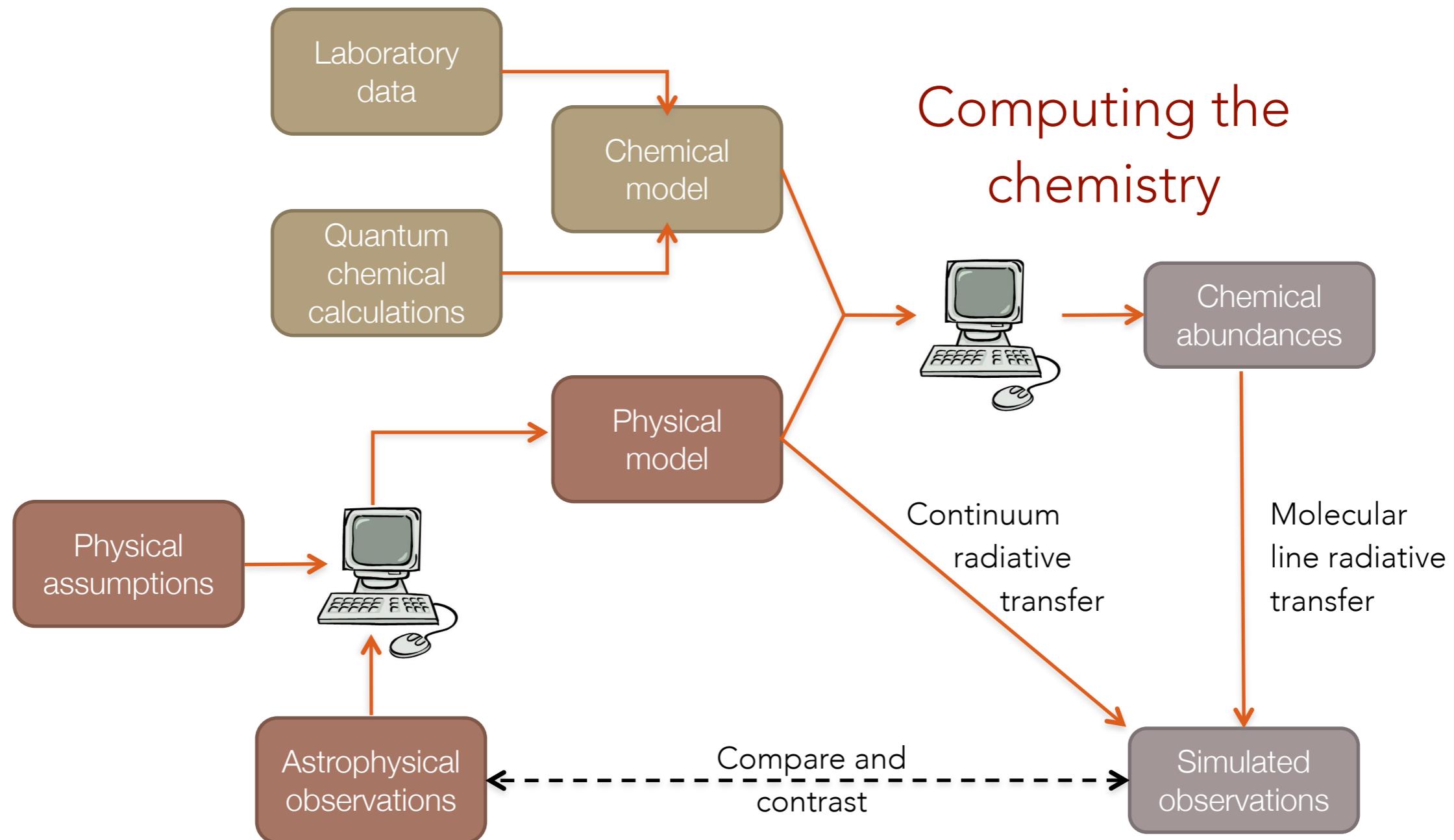
# Creating synthetic observations



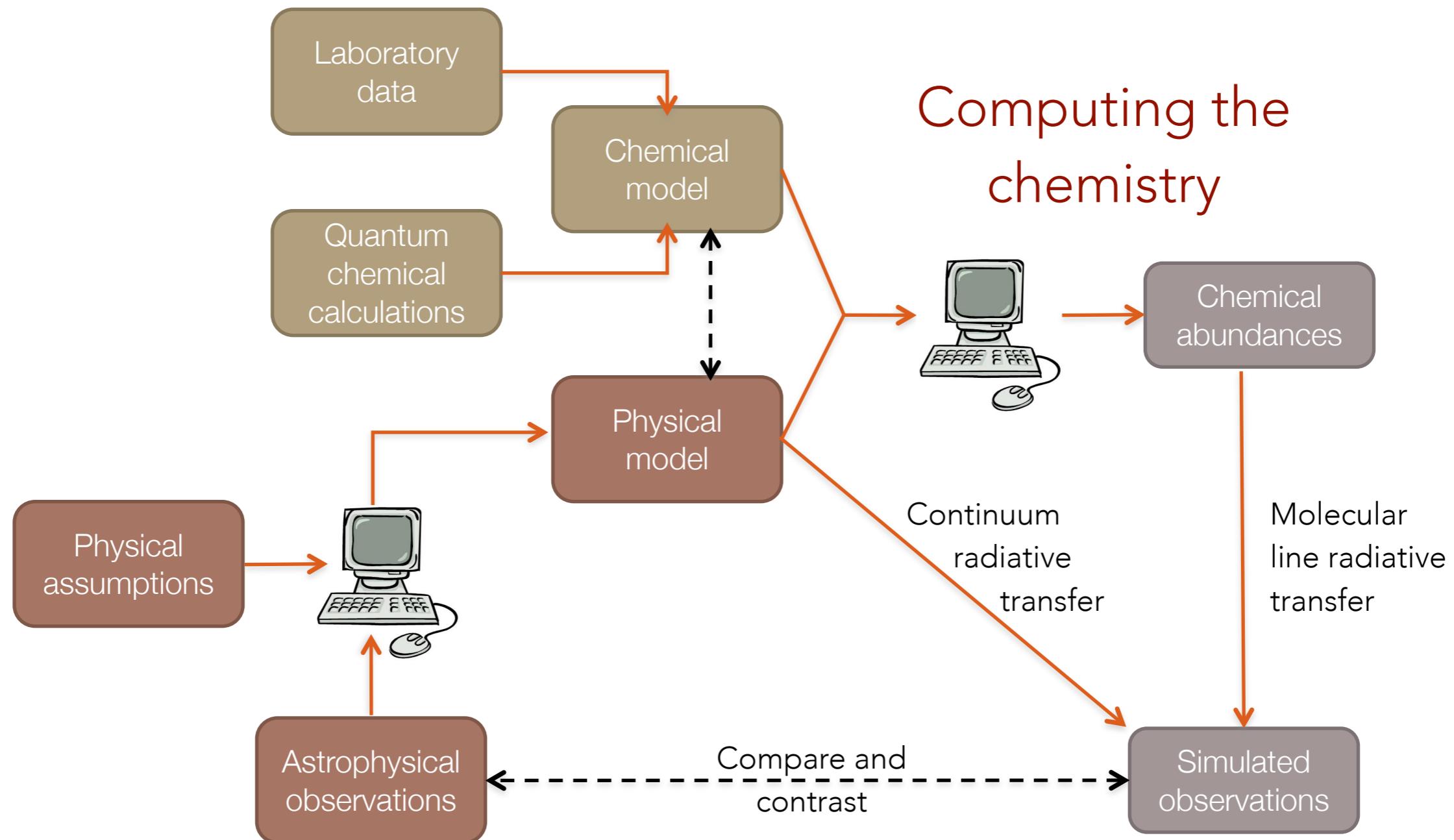
First detection of methanol ( $\text{CH}_3\text{OH}$ ) in the disk around TW Hya



# General outline of a physico-chemical model

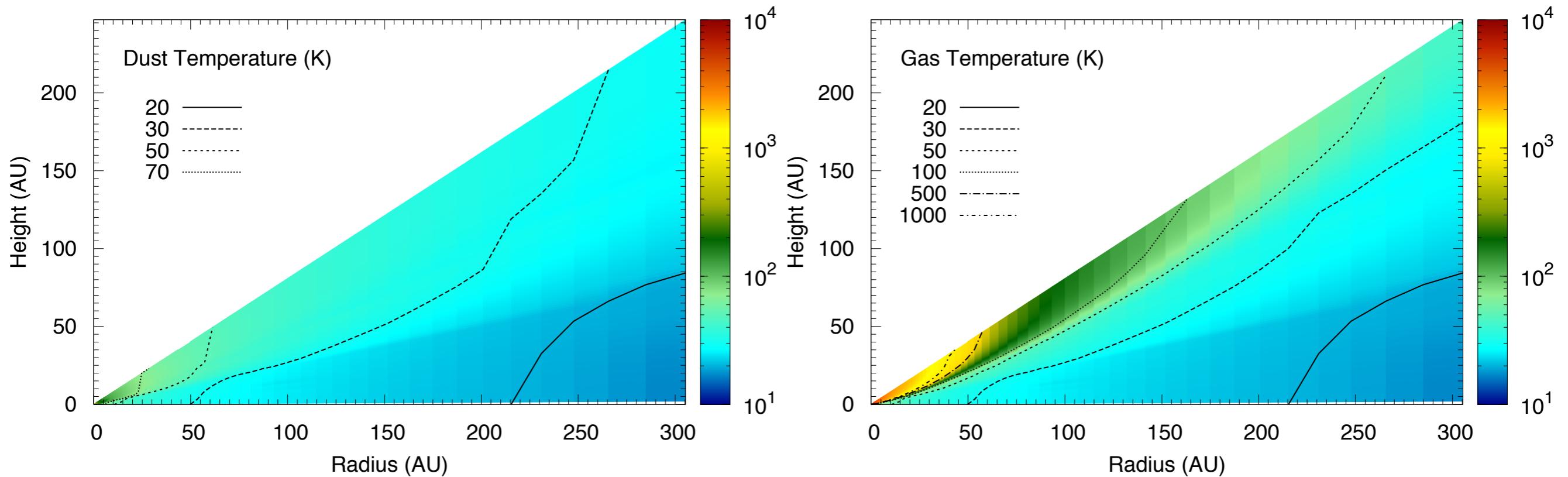


# General outline of a physico-chemical model



# $T_{\text{gas}}$ and $T_{\text{dust}}$ decouple in disk atmosphere

At low densities and high ultraviolet fluxes, gas-grain collisions are inefficient and gas cools radiatively (which is slow)



Gas temperature calculation needs to be coupled with small chemical network to compute self-consistently the abundances of the main coolants: [CI], [OI], CO, H<sub>2</sub>O

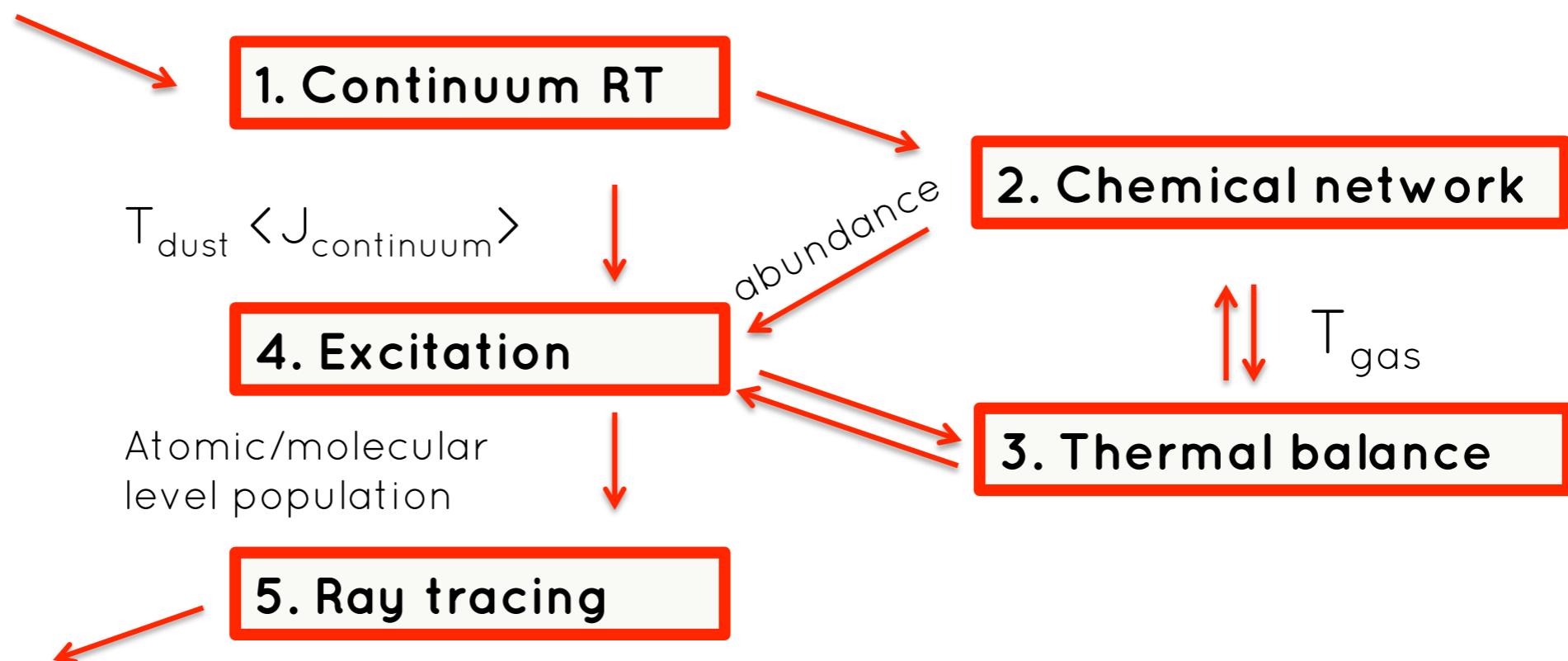
# Coupled physico-chemical models

DALI

Bruderer et al. (2012); Bruderer (2013)

## INPUTS

- Density structure
- Stellar spectrum



## OUTPUTS

- Spectra
- Image cubes

See also, ProDiMo (Woitke et al. 2009, A&A, 501, 383)

# Chemical networks for astrochemistry

Talk to an  
astrochemist!

Gas-phase chemistry

<http://www.udfa.net/>

<http://kida.obs.u-bordeaux1.fr/>

<http://kinetics.nist.gov/kinetics/index.jsp>

Photoionisation/photodissociation

<http://home.strw.leidenuniv.nl/~ewine/photo/>

<http://phidrates.space.swri.edu/>

Freezeout/desorption

Grain-surface chemistry

<http://kida.obs.u-bordeaux1.fr/>

<http://faculty.virginia.edu/ericherb/research.html>

# Molecular data

Talk to an  
astrochemist!

LAMDA: Leiden Atomic and Molecular  
Database

<http://home.strw.leidenuniv.nl/~moldata/>

Cologne Database for Molecular Spectroscopy

<http://www.astro.uni-koeln.de/cdms/>

HITRAN/HITEMP

<http://hitran.org/>

JPL Molecular Spectroscopy

<http://spec.jpl.nasa.gov/>

ExoMol

<http://www.exomol.com/>

# Radiative transfer codes

Talk to an  
astrochemist!

TORUS  
dust and lines

[https://www.astro.ex.ac.uk/people/  
th2/torus\\_html/homepage.html](https://www.astro.ex.ac.uk/people/th2/torus_html/homepage.html)

RATRAN  
lines

[https://personal.sron.nl/  
~vdtak/ratran/frames.html](https://personal.sron.nl/~vdtak/ratran/frames.html)

RADMC2D/3D  
dust and lines

[http://www.ita.uni-heidelberg.de/  
~dullemond/software/radmc-3d/](http://www.ita.uni-heidelberg.de/~dullemond/software/radmc-3d/)

HYPERION  
dust

<http://www.hyperion-rt.org/>

LIME  
dust and lines

<http://www.nbi.dk/~brinch/lime.php>

# Radiative transfer codes

Talk to an  
astrochemist!

TORUS

dust and lines

[https://www.astro.ex.ac.uk/people/th2/torus\\_html/](https://www.astro.ex.ac.uk/people/th2/torus_html/)

Dust optical constants

<http://www.mpi-astronomy.de/HJPDOC/>

RATRAN

lines

<https://personal.sron.nl/~vdtak/ratran/frames.html>

RADMC2D/3D  
dust and lines

<http://www.ita.uni-heidelberg.de/~dullemond/software/radmc-3d/>

HJPDOC

HYPERION  
dust

<http://www.hyperion-rt.org/>

LIME

dust and lines

<http://www.nbi.dk/~brinch/lime.php>

# Future outlook

- ▶ Coupling dust evolution models with thermo-chemical and complex chemistry models: dust models are inherently 1D
- ▶ Correct treatment of viscous effects on disk structure and chemistry
- ▶ Large-scale mixing: connection with the solar system
- ▶ Breaking axisymmetry: creation of vortices, dust traps, and corresponding chemical effects
- ▶ Using molecular lines to distinguish between different models to explain dust morphology as observed with ALMA
- ▶ Chemistry in evolving disks: fingerprints of early conditions